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(54) Title: IMIDAZO[2,1-b][3]BENZAZEPINE DERIVATIVES, COMPOSITIONS AND METHOD OF USE

(57) Abstract

The present invention is concerned with novel imidazo[2,1-b][3]benzazepines of formula (I), the pharmaceutically acceptable addition salts and stereochemically isomeric forms thereof, wherein each of the dotted lines independently represents an optional bond; R^1 represents hydrogen, halo, $C_{1.4}$ alkyl or $C_{1.4}$ alkyloxy; R^2 represents hydrogen, halo, $C_{1.4}$ alkyl or $C_{1.4}$ alkylloxy; R³ represents hydrogen, C₁₋₄alkyl, ethenyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl, C₁₋₄alkyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl, hydroxyC₁₋₄alkyl, formyl or hydroxycarbonyl; R⁴ represents hydrogen, C₁₋₄alkyl, hydroxyC₁₋₄alkyl, phenyl or halo; R⁵ represents hydrogen, C₁₋₄alkyl or halo; L represents hydrogen; C₁₋₆alkyl; C₁₋₆alkyl substituted with one substituent selected from the group consisting of hydroxy, halo, C₁₋₄alkyloxy, hydroxycarbonyl, $C_{1.4}$ alkyloxycarbonyl, $C_{1.4}$ alkyloxycarbonyl- $C_{1.4}$ lamino, C₁₋₄alkylaminocarbonyl, C₁₋₄alkylaminocarbonylamino, C₁₋₄alkylaminothiocarbonylamino, aryl, aryloxy and arylcarbonyl; C₁₋₆alkyl substituted with both hydroxy and aryloxy; C₃₋₆alkenyl; C₃₋₆alkenyl substituted with aryl; or, L represents a radical of formula -Alk-Y-Het¹ (a-1), -Alk-NH-CO-Het² (a-2) or -Alk-Het³ (a-3); provided that 6,11-dihydro-11-(4-piperidinylidene)-5H-imidazo[2,1-b][3]benzazepine is excluded, which are useful antiallergic compounds. Compositions comprising said compounds, methods of using and processes for preparing the same.

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IMIDAZO[2,1-b][3]BENZAZEPINE DERIVATIVES, COMPOSITIONS AND METHOD OF USE

10 Background of the invention

In WO 88/03138 there are described benzo[5,6]cycloheptapyridines which possess antiallergic and anti-inflammatory activity. In EP-A-0,339,978 there are described (benzo- or pyrido)cyclohepta heterocyclics which are useful as PAF antagonists, antihistaminics and/or anti-inflammatory agents.

In WO 92/06981 there are described 6,11-dihydro-11-(4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine and 1-acetyl-4-(5,6-dihydro-11<u>H</u>-imidazol[1,2-b][3]-benzazepine-11-ylidene)piperidine, the latter of which is useful as a PAF antagonist.

In the J. Med. Chem., <u>26</u> (1983), 974-980 there are described some 1-methyl-4-piperidinylidene-9-substituted pyrrolo[2,1-b][3]benzazepine derivatives having neuroleptic properties.

The compounds of the present invention differ structurally from the cited art-known compounds by the fact that the central 7-membered ring invariably contains a nitrogen atom of a fused imidazole ring, and by their favorable antiallergic activity.

Description of the invention

The present invention is concerned with novel imidazo[2,1-b][3]benzazepines of formula

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the pharmaceutically acceptable addition salts and stereochemically isomeric forms thereof, wherein

as each of the dotted lines independently represents an optional bond;

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- R¹ represents hydrogen, halo, C₁-4alkyl or C₁-4alkyloxy;
- R² represents hydrogen, halo, C₁-4alkyl or C₁-4alkyloxy;
- R³ represents hydrogen, C₁₋₄alkyl, ethenyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl, C₁₋₄alkyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl, hydroxyC₁₋₄alkyl, formyl or hydroxycarbonyl;
- R⁴ represents hydrogen, C₁-4alkyl, hydroxyC₁-4alkyl, phenyl or halo;
- R⁵ represents hydrogen, C₁₋₄alkyl or halo;
- L represents hydrogen; C₁₋₆alkyl; C₁₋₆alkyl substituted with one substituent selected from the group consisting of hydroxy, halo, C₁₋₄alkyloxy, hydroxycarbonyl,
- C1-4alkyloxycarbonyl, C1-4alkyloxycarbonylC1-4alkyloxy, hydroxycarbonyl-C1-4alkyloxy, C1-4alkyloxycarbonylamino, C1-4alkylaminocarbonyl, C1-4alkylaminocarbonylamino, C1-4alkylaminothiocarbonylamino, aryl, aryloxy and arylcarbonyl; C1-6alkyl substituted with both hydroxy and aryloxy;
 - C₃-6alkenyl; C₃-6alkenyl substituted with aryl;
- wherein each aryl is phenyl or phenyl substituted with halo, cyano, hydroxy, C₁₋₄alkyl, C₁₋₄alkyloxy, aminocarbonyl or phenyl substituted with C₁₋₄alkyloxycarbonyl or hydroxycarbonyl; or,
 - L represents a radical of formula

-Alk-Y-Het¹

(a-1),

-Alk-NH-CO-Het²

(a-2) or

-Alk-Het³

(a-3); wherein

Alk represents C₁-4alkanediyl;

Y represents O, S or NH;

Het¹, Het² and Het³ each represent furanyl, thienyl, oxazolyl, thiazolyl or imidazolyl each optionally substituted with one or two C₁-4alkyl substituents; pyrrolyl or pyrazolyl optionally substituted with formyl, hydroxyC₁-4alkyl, hydroxycarbonyl, C₁-4alkyloxy-carbonyl or one or two C₁-4alkyl substituents; thiadiazolyl or oxadiazolyl optionally substituted with amino or C₁-4alkyl; pyridinyl, pyrimidinyl, pyrazinyl or pyridazinyl each optionally substituted with C₁-4alkyl, C₁-4alkyloxy, amino, hydroxy or halo; imidazo[4,5-c]pyridin-2-yl; and

Het 3 may also represent 4,5-dihydro-5-oxo- $1\underline{H}$ -tetrazolyl substituted with C_{1-4} alkyl, 2-oxo-3-oxazolidinyl, 2,3-dihydro-2-oxo- $1\underline{H}$ -benzimidazol-1-yl or a radical of formula

$$R^6$$
—NH N CH_3 or Z^N CH_3 wherein $(b-1)$ $(b-2)$

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-3-

R⁶ represents hydrogen or C₁₋₄alkyl; and

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A-Z represents -S-CH=CH-, -S-CH₂-CH₂-, -S-CH₂-CH₂-CH₂-, -CH=CH-CH=CH-, -CH₂-CH₂-CH₂-CH₂-, -N(CH₃)-C(CH₃)=CH- or -CH=C(CH₃)-O-;

provided that 6,11-dihydro-11-(4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine is ecxluded.

As used in the foregoing definitions halo defines fluoro, chloro, bromo and iodo; C1-4alkyl defines straight and branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as, for example, methyl, ethyl, propyl, 1-methylethyl, butyl, 1-methylpropyl, 2-methylpropyl and 1,1-dimethylethyl; C1-6alkyl defines

C1-4alkyl radicals as defined hereinbefore and the higher homologs thereof having from 5 to 6 carbon atoms such as, for example, pentyl and hexyl; C3-6alkenyl defines straight and branched chain hydrocarbon radicals containing one double bond and having from 3 to 6 carbon atoms such as, for example, 2-propenyl, 2-butenyl, 3-butenyl, 2-methyl-2-propenyl, 2-pentenyl, 3-pentenyl, 3,3-dimethyl-2-propenyl, hexenyl and the like;

C1-4alkanediyl defines bivalent straight or branched chain hydrocarbon radicals containing from 1 to 4 carbon atoms such as, for example, methylene, 1,1-ethanediyl, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl and the like.

The term pharmaceutically acceptable addition salt as used hereinbefore defines the nontoxic, therapeutically active addition salt forms which the compounds of formula (I) may 20 form. The compounds of formula (I) having basic properties may be converted into the corresponding therapeutically active, non-toxic acid addition salt forms by treating the free base form with a suitable amount of an appropriate acid following conventional procedures. Examples of appropriate acids are for example, inorganic acids, for example, hydrohalic acid, e.g. hydrochloric, hydrobromic and the like acids, sulfuric 25 acid, nitric acid, phosphoric acid and the like; or organic acids, such as, for example, acetic, propanoic, hydroxyacetic, 2-hydroxypropanoic, 2-oxopropanoic, ethanedioic, propanedioic, butanedioic, (Z)-2-butenedioic, (E)-2-butenedioic, 2-hydroxybutanedioic, 2,3-dihydroxybutanedioic, 2-hydroxy-1,2,3-propanetricarboxylic, methanesulfonic, ethanesulfonic, benzenesulfonic, 4-methylbenzenesulfonic, cyclohexanesulfamic, 30 2-hydroxybenzoic, 4-amino-2-hydroxybenzoic and the like acids. The compounds of formula (I) having acidic properties may be converted in a similar manner into the corresponding therapeutically active, non-toxic base addition salt forms. Examples of such base addition salt forms are, for example, the sodium, potassium, calcium salts, and also the salts with pharmaceutically acceptable amines such as, for 35 example, ammonia, alkylamines, benzathine, N-methyl-D-glucamine, hydrabamine, amino acids, e.g. arginine, lysine. The term pharmaceutically acceptable addition salts

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also comprises the solvates which the compounds of formula (I) may form, e.g. the hydrates, alcoholates and the like.

The term stereochemically isomeric forms as used hereinbefore defines the possible

different isomeric as well as conformational forms which the compounds of formula (I)
may possess. Unless otherwise mentioned or indicated, the chemical designation of
compounds denotes the mixture of all possible stereochemically and conformationally
isomeric forms, said mixtures containing all diastereomers, enantiomers and/or
conformers of the basic molecular structure. All stereochemically isomeric forms of the
compounds of formula (I) both in pure form or in admixture with each other are intended
to be embraced within the scope of the present invention.

Some compounds of the present invention may exist in different tautomeric forms and all such tautomeric forms are intended to be included within the scope of the present invention.

Interesting compounds are those compounds of formula (I) wherein each of the dotted lines independently represents an optional bond;

- R¹ represents hydrogen, halo or C₁₋₄alkyl;
- 20 R² represents hydrogen, halo, C₁₋₄alkyl or C₁₋₄alkyloxy;
 - R³ represents hydrogen, C₁₋₄alkyl, hydroxyC₁₋₄alkyl, formyl or hydroxycarbonyl;
 - R⁴ represents hydrogen, C₁-4alkyl, hydroxyC₁-4alkyl, phenyl or halo;
 - R⁵ represents hydrogen;
 - L represents hydrogen; C₁-6alkyl; C₁-6alkyl substituted with hydroxy, C₁-4alkyloxycarbonylamino, C₁-4alkylaminocarbonyl, C₁-4alkylaminocarbonylamino, C₁-4alkylaminothiocarbonylamino, aryl or aryloxy; C₃-6alkenyl; C₃-6alkenyl substituted with aryl;

wherein each aryl is phenyl or phenyl substituted with halo, C₁₋₄alkyloxy; or,

30 L represents a radical of formula

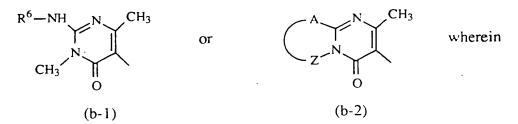
-Alk-Y-Het¹ (a-1), -Alk-NH-CO-Het² (a-2) or -Alk-Het³ (a-3); wherein

Alk represents C1-4alkanediyl;

35 Y represents O, S or NH;

Het¹, Het² and Het³ each represent furanyl, thienyl, pyrrolyl, oxazolyl, thiazolyl or imidazolyl each optionally substituted with one or two C₁-4alkyl substituents;

thiadiazolyl or oxadiazolyl optionally substituted with amino or C₁-4alkyl; pyridinyl, pyrimidinyl, pyrazinyl or pyridazinyl each optionally substituted with C₁-4alkyl, C₁-4alkyloxy, amino, hydroxy or halo; imidazo[4,5-c]pyridin-2-yl; and Het ³ may also represent 4,5-dihydro-5-oxo-1<u>H</u>-tetrazolyl substituted with C₁-4alkyl, 2-oxo-3-oxazolidinyl, 2,3-dihydro-2-oxo-1<u>H</u>-benzimidazol-1-yl or a radical of formula



R⁶ represents hydrogen or C₁₋₄alkyl; and

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A-Z represents -S-CH=CH-, -S-CH₂-CH₂-, -S-CH₂-CH₂-, -CH=CH-CH=CH-or CH₂-CH₂-CH₂-CH₂;

provided that 6,11-dihydro-11-(4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine is ecxluded.

Another group of interesting compounds comprises those compounds of formula (I) wherein L is C₁₋₄alkyl or C₁₋₄alkyl substituted with hydroxycarbonyl or C₁₋₄alkyloxy-carbonyl.

Further interesting compounds are those compounds of formula (I) wherein R¹, R², R³, R⁴ and R⁵ represent hydrogen.

Yet another group of interesting compounds of formula (I) are those of formula

$$\begin{array}{c|c}
R^1 & R^2 \\
R^5 & R^5 \\
R^4 & R^3 \\
\end{array}$$

wherein R¹, R², R³, R⁴, R⁵ and L are as defined under formula (I).

Preferred compounds are those compounds of formula (I) wherein R³ represents hydrogen, C₁-4alkyl, formyl, hydroxyC₁-4alkyl or hydroxycarbonyl; R⁴ represents hydrogen, halo or hydroxyC₁-4alkyl; and

L represents hydrogen, C₁-4alkyl, haloC₁-4alkyl, hydroxycarbonylC₁-4alkyl, C₁-4alkyloxycarbonylC₁-4alkyl, C₁-4alkyloxycarbonylaminoC₁-4alkyl, aryl-C₁-4alkyl, propenyl, or

L is a radical of formula (a-1), (a-2) or (a-3), wherein

- Het¹, Het², and Het³ each represent furanyl, oxazolyl or thiazolyl each optionally substituted with C₁₋₄alkyl; thiadiazolyl optionally substituted with amino, pyridinyl; or pyrimidinyl each optionally substituted with hydroxy; imidazo[4,5-c]pyridin-2-yl; and Het³ may also represent a radical of formula (b-2).
- 10 More preferred compounds are those preferred compounds wherein
 - R¹ represents hydrogen or halo;

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- R² represents hydrogen, halo or C₁-4alkyloxy; and
- L represents hydrogen, C₁₋₄alkyl, haloC₁₋₄alkyl, hydroxycarbonylC₁₋₄alkyl, C₁₋₄alkyloxycarbonylC₁₋₄alkyl, or a radical of formula (a-1), wherein Y represents NH.

Still more preferred are those more preferred compounds wherein R⁴ represents hydrogen or halo; and

L represents hydrogen, C₁₋₄alkyl, hydroxycarbonylC₁₋₄alkyl, C₁₋₄alkyloxycarbonylC₁₋₄alkyl or a radical of formula (a-1), wherein Het¹ is thiazolyl, or imidazo[4,5-c]pyridin-2-yl.

The most preferred compounds are:

- 5,6-dihydro-11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepine;
- 9-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]-benzazepine;
 - 11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepine;
 - 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine-3-methanol;
- 8-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]-benzazepine;
 - 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine-3-carboxaldehyde;
 - $6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5\underline{H}-imidazo[2,1-b][3] benzazepine-3-11-dihydro-11-(1-methyl-4-piperidinylidene)$
- 35 carboxylic acid;
 - 7-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]-benzazepine;

4-(8-fluoro-5,6-dihydro-11<u>H</u>-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidine-propanoic acid dihydrate,

the stereoisomers and the pharmaceutically acceptable acid-addition salts thereof.

In the following paragraphs there are described different ways of preparing the compounds of formula (I). In order to simplify the structural formulae of the compounds of formula (I) and the intermediates intervening in their preparation, the imidazo[2,1-b] [3]benzazepine moiety will be represented by the symbol T hereinafter.

$$R^{1} \qquad R^{2}$$

$$R^{5} \qquad = \qquad \qquad T$$

$$R^{4} \qquad R^{3}$$

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The compounds of formula (I) can be prepared by cyclizing an alcohol of formula (II) or a ketone of formula (III).

Said cyclization reaction is conveniently conducted by treating the intermediate of formula (II) or (III) with an appropriate acid, thus yielding a reactive intermediate which

cyclizes to a compound of formula (I). Appropriate acids are, for example, strong acids, in particular superacid systems, e.g. methanesulfonic acid, trifluoromethanesulfonic acid, trifluoroacetic acid, methanesulfonic acid / boron trifluoride, hydrofluoric acid / boron trifluoride, or Lewis acids, e.g. aluminum chloride and the like. Obviously, only those compounds of formula (I) wherein L is stable under the given reaction conditions can be prepared according to the above reaction procedure. In case of superacids the reaction is preferably conducted in an excess of said acid; in case of solid Lewis acids, e.g. aluminum chloride, the reaction can be conducted by fusing the starting material and the reagent, preferably in the presence of an additional salt such as sodium chloride. The cyclodehydration reaction with trimethylsilyl iodide is conveniently conducted in a reaction-inert solvent such as, for example, a halogenated hydrocarbon, e.g. trichloromethane. Particularly noteworthy is the fact that the latter reaction also can be performed on intermediates of formula (II) or (III) wherein L represents C₁₋₄alkyloxycarbonyl; in this case - besides cyclodehydration - also cleavage of the carbamate is observed and a compound of formula (I) wherein L is hydrogen is obtained.

In the foregoing and following preparations, the reaction mixture is worked up following art-known methods and the reaction product is isolated and, if necessary, further purified.

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The compounds of formula (I) wherein the central ring of the tricyclic moiety does not contain an optional bond may also be prepared by cyclizing an intermediate of formula (IV).

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In formula (IV) and hereinafter W represents an appropriate leaving group such as, for example, halo, e.g. chloro, bromo and the like; or a sulfonyloxy group such as, for example, methansulfonyloxy, 4-methylbenzenesulfonyloxy and the like.

Said cyclization reaction can conveniently be conducted in a reaction-inert solvent such as, for example, an aromatic hydrocarbon, e.g., benzene, methylbenzene, dimethylbenzene and the like; an alkanol, e.g., methanol, ethanol, 1-butanol and the like;

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a ketone, e.g., 2-propanone, 4-methyl-2-pentanone and the like; an ether, e.g., tetrahydrofuran, 1,4-dioxane, 1,1'-oxybisethane and the like; a dipolar aprotic solvent, e.g., N,N-dimethylformamide, N,N-dimethylacetamide, dimethyl sulfoxide, nitrobenzene, 1-methyl-2-pyrrolidinone and the like; a halogenated hydrocarbon, e.g. dichloromethane, 1,2-dichloroethane and the like; or a mixture of such solvents. The addition of an appropriate base such as, for example, an alkali or an earth alkaline metal carbonate, hydrogen carbonate, alkoxide, hydride, amide, hydroxide or oxide, e.g., sodium carbonate, sodium hydrogen carbonate, potassium carbonate, sodium methoxide, sodium ethoxide, potassium tert. butoxide, sodium hydride, sodium amide, sodium hydroxide, calcium carbonate, calcium hydroxide, calcium oxide and the like; or an organic base, such as, for example, an amine, e.g., N,N-diethylethanamine, N-(1-methylethyl)-2-propanamine, 4-ethylmorpholine, pyridine and the like may be utilized to pick up the acid which is liberated during the course of the reaction. In some instances the addition of an iodide salt, preferably an alkali metal iodide, is appropriate. Somewhat elevated temperatures and stirring may enhance the rate of the reaction.

Alternatively, the compounds of formula (I) wherein a double bond exists between the piperidinyl and the imidazo[2,1-b][3]benzazepine moiety, said compounds being represented by formula (I-a), can be prepared by dehydrating an alcohol of formula (V) or (VI).

Said dehydration reaction can conveniently be conducted employing conventional dehydrating reagents following art-known methodologies. Appropriate dehydrating

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reagents are, for example, acids, e.g. sulfuric acid, phosphoric acid, hydrochloric acid, methanesulfonic acid, carboxylic acids, e.g. acetic acid, trifluoroacetic acid and mixtures thereof; anhydrides, e.g. acetic anhydride, phosphorus pentoxide and the like; other suitable reagents, e.g. zinc chloride, thionyl chloride, boron trifluoride etherate, phosphoryl chloride pyridine, potassium bisulfate, potassium hydroxide. In some instances said dehydration reaction may require heating the reaction mixture, more particularly up to the reflux temperature. Again, only those compounds of formula (I-a) wherein L is stable under the given reaction conditions can be prepared according to the above reaction procedure. Particularly noteworthy is the fact that the latter reaction when performed on intermediate (V) wherein the dotted line does not represent an optional bond, in some instances may also yield a compound of formula (I) with a double bond in the tricyclic moiety and a single bond bridging the tricyclic moiety and the piperidine:

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The compounds of formula (I) wherein L is C_{1-6} alkyl, said compounds being represented by the formula (I-b) can be converted into the compounds of formula (I), wherein L is hydrogen, said compounds being represented by the formula (I-c) in a number of manners. A first method involves dealkylating - carbonylating the compounds of formula (I-b) with a C_{1-4} alkylchloroformate and subsequently hydrolyzing the thus obtained compound of formula (VII-a).

The reaction with the C_{1-4} alkylchloroformate is conveniently conducted by stirring and heating the starting material (I-b) with the reagent in an appropriate solvent and in the presence of a suitable base. Appropriate solvents are, for example, aromatic hydrocarbons, e.g. methylbenzene, dimethylbenzene, chlorobenzene; ethers, e.g. 1,2-dimethoxyethane, and the like solvents. Suitable bases are, for example, alkali or earth alkaline metal carbonates, hydrogen carbonates, hydroxides, or organic bases such as, N,N-diethylethanamine, N-(1-methylethyl)-2-propanamine, and the like.

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The compounds of formula (VII-a) are hydrolyzed in acidic or basic media following conventional methods. For example, concentrated acids such as hydrobromic, hydrochloric acid or sulfuric acid can be used, or alternatively bases such as alkali metal or earth alkaline metal hydroxides in water, an alkanol or a mixture of water-alkanol may be used. Suitable alkanols are methanol, ethanol, 2-propanol and the like. In order to enhance the rate of the reaction it is advantageous to heat the reaction mixture, in particular up to the reflux temperature.

The compounds of formula (I-b) may also be converted directly into the compounds of formula (I-c) by stirring and heating them with an α-halo-C₁₋₄alkyl chloroformate in an appropriate solvent such as, for example, a halogenated hydrocarbon, e.g. dichloromethane, trichloromethane; an aromatic hydrocarbon, e.g. methylbenzene, dimethylbenzene; an ether, e.g. 1,2-dimethoxyethane; an alcohol, e.g. methanol, ethanol, 2-propanol, optionally in the presence of a base such as, for example, an alkali or earth alkaline metal carbonate, hydrogen carbonate, hydroxide or an amine, e.g. N,N-diethylethanamine, N-(1-methylethyl)-2-propanamine, and the like.

The compounds of formula (I-c) can also be prepared by debenzylating a compound of formula (I-d) by catalytic hydrogenation in the presence of hydrogen and an appropriate catalyst in a reaction-inert solvent.

$$CH_2-N$$
 $H-N$
 $(I-c)$

A suitable catalyst in the above reaction is, for example, platinum-on-charcoal, palladium-on-charcoal, and the like. An appropriate reaction-inert solvent for said debenzylation reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like, an ester, e.g. ethylacetate and the like, an acid, e.g. acetic acid and the like.

The compounds of formula (I) wherein L is other than hydrogen, said compounds being represented by formula (I-e) and said L by L^1 , can be prepared by N-alkylating the compounds of formula (I-c) with a reagent of formula L^1 -W (VIII).

$$L^1-W$$
(I-c)
$$L^1-W$$
(VIII)
(I-e)

Said N-alkylation reaction can conveniently be conducted in a reaction-inert solvent such as, for example, an aromatic hydrocarbon, e.g., benzene, methylbenzene, dimethylbenzene and the like; an alkanol, e.g., methanol, ethanol, 1-butanol and the like; a ketone, e.g., 2-propanone, 4-methyl-2-pentanone and the like; an ether, e.g., tetrahydrofuran, 1,4-dioxane, 1,1'-oxybisethane and the like; a dipolar aprotic solvent, e.g., N,N-dimethylformamide, N,N-dimethylacetamide, dimethyl sulfoxide, nitrobenzene, 1-methyl-2-pyrrolidinone and the like; a halogenated hydrocarbon, e.g. dichloromethane, 1,2-dichloroethane and the like; or a mixture of such solvents. The addition of an appropriate base such as, for example, an alkali or an earth alkaline metal carbonate, hydrogen carbonate, alkoxide, hydride, amide, hydroxide or oxide, e.g., sodium carbonate, sodium hydrogen carbonate, potassium carbonate, sodium methoxide, sodium ethoxide, potassium tert. butoxide, sodium hydride, sodium amide, sodium hydroxide, calcium carbonate, calcium hydroxide, calcium oxide and the like; or an organic base, such as, for example, an amine, e.g., N,N-diethylethanamine, \underline{N} -(1-methylethyl)-2-propanamine, 4-ethylmorpholine, pyridine and the like may be utilized to pick up the acid which is liberated during the course of the reaction. In some instances the addition of an iodide salt, preferably an alkali metal iodide, is appropriate. Somewhat elevated temperatures and stirring may enhance the rate of the reaction. Alternatively, said N-alkylation may be carried out by applying art-known conditions of phase transfer catalysis reactions.

The compounds of formula (I) wherein L is C_{1-6} alkyl or substituted C_{1-6} alkyl, said L being represented by the radical L²H- and said compounds by formula (I-f), can also be prepared by reductive N-alkylation of the compounds of formula (I-c) with an appropriate ketone or aldehyde of formula L²=O (IX). L²=O represents an intermediate of formula L²H₂ wherein two geminal hydrogen atoms have been replaced by oxygen (=O) and L² is a geminal bivalent C_{1-6} alkylidene radical which optionally may be substituted.

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H-N
$$L^2=0$$
 L^2H-N T (I-f)

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Said reductive N-alkylation reaction may conveniently be carried out by reducing a mixture of the reactants in a suitable reaction-inert solvent following art-known reductive N-alkylation procedures. In particular, the reaction mixture may be stirred and/or heated in order to enhance the reaction rate. Suitable solvents are, for example, water; C₁₋₆alkanols, e.g. methanol, ethanol, 2-propanol and the like; esters, e.g. ethyl acetate, γ-butyrolactone and the like; ethers, e.g. 1,4-dioxane, tetrahydrofuran, 1,1'-oxybisethane, 2-methoxyethanol and the like; halogenated hydrocarbons, e.g. dichloromethane, trichloromethane and the like; dipolar aprotic solvents, e.g. N,N-dimethylformamide, dimethyl sulfoxide and the like; carboxylic acids, e.g. acetic acid, propanoic acid and the like; or a mixture of such solvents. The term "art-known reductive N-alkylation procedures" means that the reaction is carried out either with sodium cyanoborohydride, sodium borohydride, formic acid or a salt thereof, e.g. ammonium formate and the like reducing agents, or alternatively under hydrogen atmosphere, optionally at an increased temperature and/or pressure, in the presence of an appropriate catalyst such as, for example, palladium-on-charcoal, platinum-on-charcoal and the like. In order to prevent the undesired further hydrogenation of certain functional groups in the reactants and the reaction products, it may be advantageous to add an appropriate catalyst-poison to the reaction mixture, e.g., thiophene, quinoline-sulphur and the like. In some instances it may also be advantageous to add an alkali metal salt to the reaction mixture such as, for example, potassium fluoride, potassium acetate and the like salts.

The compounds of formula (I) wherein L represents a radical Het³-C₂₋₄alkyl, said compounds being represented by formula (I-g) can be prepared by the addition reaction of a compound of formula (I-c) to an appropriate alkene of formula (X).

The compounds wherein L is 2-hydroxy-C₂₋₆alkyl, or aryloxy-2-hydroxy-C₂₋₆alkyl said compounds being represented by formula (I-h), can be prepared by reacting a compound of formula (I-c) with an epoxide (XI) wherein R⁷ represents hydrogen, C₁₋₄alkyl or aryloxyC₁₋₄alkyl.

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The reaction of (I-c) with respectively (X) or (XI) can be conducted by stirring and, if desired, heating the reactants in a reaction-inert solvent such as, for example, a ketone, e.g. 2-propanone, 4-methyl-2-pentanone; an ether, e.g. tetrahydrofuran; an alcohol, e.g. methanol, ethanol, 1-butanol; a dipolar aprotic solvent, e.g. N,N-dimethylformamide and the like.

The compounds of formula (VII-b) can be prepared from a compound of formula (I-i) wherein L represents P-NH-C₂₋₄alkyl and P is a protective group such as, for example, C₁₋₄alkyloxycarbonyl, following art-known deprotection methods.

$$P-NH-C_{2-4}alkyl-N \xrightarrow{\qquad \qquad } H_2N-C_{2-4}alkyl-N \xrightarrow{\qquad \qquad } T$$

$$(I-i) \qquad \qquad (VII-b)$$

The compounds of formula (VII-b) can also be prepared by reducing a compound of formula (VII-c).

Said reduction can be conducted by stirring and, if desired, heating the starting material in a hydrogen containing medium in the presence of a catalyst, e.g. palladium-on-charcoal, platinum-on-charcoal, Raney Nickel and the like, in a suitable solvent, e.g. methanol, ethanol and the like, or by reduction with a metal hydride, e.g. lithium aluminum hydride in an ether, e.g. tetrahydrofuran.

The compounds of formula (I) wherein L is a radical of formula -Alk-Y-Het¹, said compounds being represented by formula (I-j), can be prepared by alkylating a compound of formula (I-k) with a reagent of formula (XII).

$$H-Y-Alk-N$$

$$(I-k)$$
 $Het^1-Y-Alk-N$

$$(I-j)$$

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Alternatively, the compounds of formula (I-j) can also be prepared by reacting a compound of formula (VII-d) with a reagent of formula (XIII).

$$W-Alk-N \longrightarrow T \qquad \frac{Het^1-Y-H}{(XIII)} \qquad Het^1-Y-Alk-N \longrightarrow T$$

$$(VII-d) \qquad (I-j)$$

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The above alkylation reactions may conveniently be conducted in a reaction-inert solvent, e.g. methylbenzene, dimethylbenzene, 2-propanone, 4-methyl-2-pentanone, 1,4-dioxane, tetrahydrofuran, N,N-dimethylformamide, N,N-dimethylacetamide, methanol, ethanol, 1-butanol and the like. The addition of an appropriate base, e.g. an alkali metal or earth alkaline metal carbonate or hydrogen carbonate, sodium hydride, N,N-diethylethanamine or N-(1-methylethyl)-2-propanamine may be used to pick up the acid liberated during the course of the reaction. In some instances the addition of an iodide salt, preferably an alkali metal iodide, is appropriate. In order to enhance the rate of the reaction the reaction mixture may be heated.

The compounds of formula (I) wherein L represents a radical of formula -Alk-NH-CO-Het², said compounds being represented by formula (I-l) can be prepared by N-acylating a compound of formula (VII-b) with a carboxylic acid of formula (XIV) or a reactive functional derivative thereof.

$$\begin{array}{c|c} H_2N-C_{2-4}alkyl-N & & & \\ \hline \\ (VII-b) & & & \\ \end{array}$$

The reaction of (XIV) with (VII-b) may generally be conducted following art-known amidation reaction procedures. For example, the carboxylic acid may be converted into a reactive derivative, e.g. an anhydride or a carboxylic acid halide, which subsequently is reacted with (VII-b); or by reacting (XIV) and (VII-b) with a suitable reagent capable of forming amides, e.g., N,N-methanetetraylbis[cyclohexamine], 2-chloro-1-methyl-pyridinium iodide and the like. Said reactions are conveniently conducted in a suitable solvent such as, for example, an ether, e.g., tetrahydrofuran, a halogenated hydrocarbon, e.g., dichloromethane, trichloromethane, a dipolar aprotic solvent and the like. The addition of a base such as, for example, N,N-diethylethanamine and the like may be appropriate.

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The compounds of formula (I) wherein L represents C_{1-4} alkylamino(thio)carbonylamino C_{1-4} alkyl, said compounds being represented by the formula (I-m), can be prepared from the compounds of formula (VII-b) by reaction with a C_{1-4} alkyliso(thio)cyanate in a reaction-inert solvent such as, for example, an ether, e.g. tetrahydrofuran.

The compounds of formula (I) wherein Het ¹ represents an imidazo[4,5-c]pyridin-2-yl radical and Y represents NH, said compounds being represented by formula (I-n) can be prepared from a compound of formula (VII-b) according to the following reaction scheme.

$$H_{2}N-C_{2.4}alkyl-N \longrightarrow T$$

$$(VII-b)$$

$$S=C=N-C_{2.4}alkyl-N \longrightarrow T$$

$$(VII-e)$$

$$N \longrightarrow N+2$$

$$N+C_{2.4}alkyl-N \longrightarrow T$$

$$(I-n)$$

$$(VII-e)$$

$$N \longrightarrow N+2$$

$$N+C_{2.4}alkyl-N \longrightarrow T$$

$$(VII-f)$$

The isocyanate (VII-e) is prepared by reacting (VII-b) with carbon disulfide in the presence of a dehydrating reagent such as N,N-methanetetraylbis[cyclohexanamine] in a reaction-inert solvent such as an ether, e.g. tetrahydrofuran. The isothiocyanate is reacted with 3,4-diaminopyridine in a reaction-inert solvent such as an ether, e.g. tetrahydrofuran, and the resulting thiourea is cyclized by treatment with an appropriate metal oxide such as mercury(II)oxide. In certain instances if may be appropriate to supplement the reaction mixture with a small amount of sulfur.

The compound (VII-e) or the corresponding isocyanate can also be employed to prepare compounds of formula (I-m), by reacting (VII-e) or the corresponding

isocyanate with a C_{1-4} alkylamine in a reaction-inert solvent such as an ether, e.g. tetrahydrofuran.

$$D=C=N-C_{2-4}alkyl-N \qquad \qquad T \qquad + \qquad C_{1-4}alkyl-NH_2 \qquad \qquad D$$

$$C_{1-4}$$
alkyl-NH- C -NH- C_{2-4} alkyl-N

D is S: (I-m-1)
D is O: (I-m-2)

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The compounds of formula (I) wherein Het¹ represents an imidazole and Y represents NH, said compounds being represented by formula (I-o) can be prepared from the compounds (VII-b) according to the following reaction scheme.

$$H_2N-C_{2-4}alkyl-N$$
 CH_3O
 $CH-CH_2-NH-C-S-CH_3$
 CH_3O
 CH_3O

$$\begin{array}{c} \text{CH}_{3}\text{O} \\ \text{CH}-\text{CH}_{2}-\text{NH}-\text{C}-\text{NH}-\text{C}_{2\text{-4}}\text{alkyl}-\text{N} \\ \text{CH}_{3}\text{O} \end{array}$$

$$CH_3$$
 N
 $NH-C_{2-4}alkyl-N$
 $(I-o)$

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The compound (VII-b) is reacted with a reagent of formula (XV) in a reaction-inert solvent such as an alcohol, e.g. 2-propanol and the thus obtained intermediate (VII-g) is cyclized by treatment with an acidic aqueous solution, such as a hydrochloric acid aqueous solution.

The compounds of formula (I) wherein R³ and/or R⁴ represent hydroxymethyl can be prepared by formylating the compounds of formula (I), wherein R³ and/or R⁴ are hydrogen, said compounds being represented by the formula (I-p) with formaldehyde, optionally in the presence of an appropriate carboxylic acid - carboxylate mixture such

as, for example, acetic acid - sodium acetate and the like. In order to enhance the rate of the reaction, the reaction mixture is advantageously heated up to the reflux temperature.

The thus obtained compounds (I-q) and (I-r) can be further oxidized to the corresponding aldehyde or carboxylic acid by reaction with suitable reagents such as, for example, manganese(IV)oxide, respectively, silver nitrate.

The compounds of formula (I) wherein R⁴ is halo, said compounds being represented by formula (I-s), can be prepared by halogenating the compounds of formula (I), wherein R⁴ is hydrogen, said compounds being represented by the formula (I-t).

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Said halogenation reaction can conveniently be conducted by treating the starting material with dihalide in an appropriate solvent such as, for example, a carboxylic acid, e.g. acetic acid, optionally in admixture with a carboxylate salt, e.g. sodium acetate. In order to enhance the rate of the reaction, the reaction mixture may be heated.

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The compounds of formula (I) wherein Het³ represents a pyrrolyl radical, said compounds being represented by the formula (I-u), can be prepared by reacting a compound of formula (VII-b) with a reagent of formula (XVI).

H₂N-C₂₋₄alkyl-N
$$\longrightarrow$$
 T + C₁₋₄alkyl-O \longrightarrow O \longrightarrow O \longrightarrow C₁₋₄alkyl \longrightarrow N-C₂₋₄alkyl-N \longrightarrow T \longrightarrow (I-u)

In a similar way, the compounds of formula (I) wherein Het³ represents a 2-C₁₋₄alkyloxycarbonyl-1-pyrrolyl radical, said compounds being represented by the formula (I-v), can be prepared by reacting a compound of formula (VII-b) with a reagent of formula (XVII).

$$\begin{array}{c} O \\ C \\ C \\ -O \\ -C_{1-4}alkyl \\ O \\ O \\ -O \\ -C_{1-4}alkyl \\ O \\ O \\ -O \\ -C_{1-4}alkyl \\ O \\ C \\ -O \\ -C_{1-4}alkyl \\ O \\ -O \\ -O \\ -C_{1-4}alkyl \\ O \\ -C_{1-4}alkyl \\$$

The above reactions of (VII-b) with (XVI) and (XVII), respectively, preferably are conducted in the presence of an acid, such as, for example, acetic acid.

Further, the compounds of formula (I-u) may be converted in the corresponding aldehyde and alcohol compounds, said compounds being represented by the formulae (I-w) and (I-x), respectively, by the following reaction sequence.

(I-u)
$$N-C_{2-4}alkyl-N$$
 $m-C_{2-4}alkyl-N$ $m-C_{$

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The formylation of (I-u) into (I-w) can conviently be conducted in a reaction-inert solvent such as, for example, a dipolar aprotic solvent, e.g. N,N-dimethylformamide,

<u>N,N</u>-dimethylacetamide and the like, in the presence of a formylating reagent such as, for example, phosphoryl chloride, zinc cyanide and hydrochloric acid, trichloromethane and hydroxide ions, and the like. The compounds of formula (I-w) can be reduced into the compounds of formula (I-x) in a reaction-inert solvent, such as, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like in the presence of an appropriate reductant, such as, for example, metallic hydrides, e.g. lithium aluminium hydride, sodium borohydride, sodium cyanoborohydride, and the like.

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The compounds of formula (I-v) and (I-w), can be converted in the corresponding compounds of formula (I) wherein Het³ is a 2-hydroxycarbonyl-1-pyrrolyl radical by the hydrolysis of (I-v) in the presence of an acid or a base, or oxidation of (I-w) in the presence of a suitable oxidizing reagent.

The compounds of formula (I) wherein R^3 is C_{1-4} alkyloxycarbonylethenyl, said compounds being represented by the formula (I-y), can be prepared by reacting a compound of formula (I) wherein R^3 is formyl, said compounds being represented by the formula (I-z) with a reagent of formula (XVIII) in the presence of a base e.g. piperidine, pyridine, and the like.

$$L-N$$

$$R^{1}$$

$$R^{2}$$

$$HO-C-CH_{2}-C-O-C_{1-4}alkyl$$

$$(I-z)$$

$$R^{1}$$

$$R^{2}$$

$$R^{2}$$

$$R^{3}$$

$$R^{4}$$

$$R^{5}$$

$$R^{5}$$

$$R^{5}$$

$$R^{5}$$

$$R^{5}$$

$$R^{6}$$

$$R^{7}$$

The compounds of formula (I-y) can be further hydrolyzed into a compound of formula (I) wherein R³ is hydroxycarbonylethenyl, in the presence of an acid or a base.

The compounds of formula (I) wherein R³ is methoxycarbonylmethyl, said compounds being represented by the formula (I-aa), can be prepared by reacting a

compound of formula (I-z) with a reagent of formula (XIX) in the presence of benzyltrimethyl ammonium hydroxide in a reaction-inert solvent e.g. tetrahydrofuran.

The compounds of formula (I-aa) can be further hydrolyzed into a compound of formula (I) wherein R³ is hydroxycarbonylmethyl, in the presence of an acid or a base.

The compounds of formula (I) may further be converted into each other following artknown functional group transformation procedures.

For example, the compounds of formula (I) wherein L contains a C₁₋₄alkyloxy-carbonyl moiety can be hydrolyzed into a compound of formula (I) wherein L contains a hydroxycarbonyl moiety in the presence of an acid or a base.

The compounds of formula (I) wherein L is C₁₋₄alkyloxyphenylC₁₋₆alkyl can be converted into a compound of formula (I) wherein L is hydroxyphenylC₁₋₆alkyl upon treatment with an acid, such as, for example, hydrobromic acid, hydroiodic acid or a Lewis acid, e.g. borontrifluoride, aluminiumtrichloride and the like.

The compounds of formula (VII-a to VII-g) intervening in the preparations described hereinbefore are novel and have especially been developed for use as intermediates in said preparations. Consequently, the present invention also relates to novel compounds of formula

$$Q-N \xrightarrow{R^1 \qquad R^2} R^5$$

$$Q-N \xrightarrow{N \qquad N} R^3$$

$$R^4 \qquad (VII)$$

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the addition salt forms thereof and the stereochemically isomeric forms thereof, wherein each of the dotted lines independently represents an optional bond,

R¹, R², R³, R⁴ and R⁵ are as defined under formula (I); and

Q is (C₁₋₆alkyl or phenyl)oxycarbonyl, C₁₋₄alkylcarbonyl or C₁₋₆alkyl substituted with halo, cyano, amino, isothiocyanato, (4-amino-3-pyridinyl)aminothiocarbonylamino, (CH₃O)₂CH-CH₂-NH-C(=NCH₃)-NH- or methylsulfonyloxy; provided that 1-acetyl-4-(5,6-dihydro-11 \underline{H} -imidazol[1,2-b][3]benzazepine-11-ylidene)piperidine is excluded.

Particularly interesting compounds of formula (VII) are those wherein Q represents $(C_{1-6}alkyl \text{ or phenyl})$ oxycarbonyl, $C_{1-4}alkyl$ carbonyl or $C_{1-6}alkyl$ substituted with cyano or amino, the pharmaceutically acceptable acid addition salts thereof and the stereochemically isomeric forms thereof.

In the following paragraphs there are described several methods of preparing the starting materials employed in the foregoing preparations.

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The intermediates of formula (II) can be prepared from the corresponding ketones of formula (III) by reduction.

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Said reduction can conveniently be conducted by reacting the starting ketone (III) with hydrogen in a solvent such as, for example, an alcohol, e.g. methanol, ethanol; an acid, e.g. acetic acid; an ester, e.g. ethyl acetate; in the presence of a hydrogenation catalyst, e.g. palladium-on-charcoal, platinum-on-charcoal, Raney Nickel.

In order to enhance the rate of the reaction, the reaction mixture may be heated and, if desired, the pressure of the hydrogen gas may be raised.

Alternatively, the alcohols of formula (II) can also be prepared by reducing the ketones (III) with a reducing agent such as, for example, lithium aluminum hydride, sodium borohydride, sodium cyanoborohydride and the like in a suitable solvent such as, for example, an ether, e.g. 1,1'-oxybisethane, tetrahydrofuran and the like; an alcohol, e.g. methanol, ethanol and the like.

The ketones of formula (III) can be prepared by the addition of a compound of formula (XX) to a reagent of formula (XXI) under the reaction conditions described hereinbefore for the preparation of the compounds of formula (I-g) from the compounds of formula (I-c).

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The ketones of formula (III) wherein the dotted line is not an optional bond can be prepared by N-alkylating an intermediate of formula (XX) with a reagent of formula (XXII) wherein W represents a reactive leaving group as defined hereinbefore.

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Said N-alkylation reaction can conveniently be conducted following the procedures employed in preparing the compounds of formula (I-e) from the compounds of formula (I-c).

Further, the ketones of formula (III) wherein the dotted line is not an optional bond may also be prepared by reductive N-alkylation of the compounds of formula (XX) under the reaction conditions described for the preparation of the compounds of formula (I-f) from the compounds of formula (I-c).

The intermediates of formula (XX) are conveniently prepared from an ester of formula (XXIII) by reaction with a protected imidazole derivative of formula (XXIV) by reaction with a strong base such as, for example, methyl lithium, butyl lithium, sodium amide, a dialkyl lithium amide, e.g. diisopropyl lithium amide, or a mixture thereof, in a reaction-inert solvent, e.g. tetrahydrofuran, hexane, methylbenzene and the like, or a mixture thereof.

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In (XXIV) P represents a protective group such as, for example, $di(C_{1-4}alkoxy)$ -methyl, $C_{1-4}alkoxymethyl$, benzenesulfonyl, trimethylsilylethoxymethyl, N,N-dialkylaminomethyl which can be removed by acid hydrolysis. The reaction of (XXIII) and (XXIV) is conveniently conducted at low temperatures. For example, the reagent (XXIV) may be added at a temperature between about -80°C to about -40°C to the strong base. Subsequently, the ester (XXIII) is added and the reaction mixture is allowed to warm up gently to room temperature. The thus obtained product is converted into intermediate (XX) by very mild acid hydrolysis and isolated in a conventional manner.

The ketones of formula (III) wherein L represents methyl, can be prepared from the ketones wherein L represents hydrogen by reductive N-alkylation with formaldehyde following the methods described hereinbefore for the preparation of the compounds of formula (I-f) from the compounds of formula (I-c).

The ketones of formula (III) wherein L represents hydrogen are prepared by hydrolysis of a carbamate of formula (III-a) in acidic or basic media following conventional methods as described hereinbefore for the preparation of compounds of formula (I-c) from the compounds of formula (I-b).

-25-

$$R^1$$
 R^2
 R^5
 R^5
 R^5
 R^3

The intermediates of formula (III-a) can be prepared by reacting an acid halide of formula (XXV) with an imidazole derivative of formula (XXVI).

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Said reaction is conveniently conducted by stirring and heating the reactants in the presence of a base such as, for example, an amine, e.g. N.N-diethylethanamine, N-methylmorpholine and the like, in a suitable solvent such as, for example, pyridine, acetonitrile or a mixture thereof.

The intermediates of formula (III-c) can also be prepared from an ester of formula (XXVII) by reaction with an imidazole of formula (XXVI) in the presence of a strong base such as, for example, methyl lithium, butyl lithium, sodium amide, a dialkyl lithium amide, e.g. diisopropyl lithium amide, or a mixture thereof, in a suitable reaction-inert solvent, e.g. tetrahydrofuran, hexane, methylbenzene and the like, or a mixture thereof.

Said reaction is conveniently conducted at low temperatures. For example the reagent (XVI) may be added at a temperature between about -80°C to about -40°C to the strong base. Subsequently the ester is added and the reaction mixture is allowed to warm up gently to room temperature.

$$CH_3 - N - C - O - C_{1-4}alkyl$$
 (III-c)
$$(XXVII)$$

The intermediates of formula (V) can be prepared by addition of a Grignard reagent (XXVIII) to a ketone of formula (XXIX) in a reaction-inert solvent, e.g. tetrahydrofuran.

$$L-N \longrightarrow Mg-halo + O \longrightarrow R^1 \longrightarrow R^2$$

$$L-N \longrightarrow Mg-halo + O \longrightarrow R^3$$

$$R^3 \longrightarrow R^3$$

$$(XXIX)$$

$$(V) \longrightarrow R^4$$

The tricyclic ketones of formula (XXIX) in turn are prepared from intermediates of formula (XXX) by oxidation with suitable oxidizing reagent in a reaction-inert solvent.

$$R^1$$
 R^2
 R^5
 R^5
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3

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Suitable oxidizing reagents are, for example, manganese dioxide, selenium dioxide, ceric ammonium nitrate and the like. Reaction-inert solvents are, for example, halogenated hydrocarbons, e.g. dichloromethane, trichloromethane and the like.

The compounds of formula (XXX) wherein the dotted lines do not represent an optional bond, can be prepared from the corresponding compounds of formula (XXX) wherein said dotted lines do represent an optional bond, following art-known hydrogenation procedures, e.g. by reaction with hydrogen in the presence of a hydrogenation catalyst.

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$$R^1$$
 R^2
 R^5

hydrogenation

 R^1
 R^2
 R^5
 R^5
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^5

The intermediates of formula (XXX-a) can be prepared from a benzazepine of formula (XXXI) by reaction with a reagent of formula (XXXII) and cyclization of the thus obtained intermediate (XXXIII) in an acidic medium. In (XXXII) R represents C_{1-4} alkyl or both radicals R taken together represent C_{2-6} alkanediyl, e.g. 1,2-ethanediyl, 1,3-propanediyl, 2,2-dimethyl-1,3-propanediyl.

The preparation of (XXXIII) is conveniently conducted by stirring and heating the reactants in a reaction-inert solvent such as, for example, an alcohol, e.g. methanol,

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ethanol and the like.

The cyclization reaction to the intermediates of formula (XXX-a) is conducted by stirring and heating the starting material (XXXIII) in a carboxylic acid such as, for example, acetic acid, propanoic acid, optionally in admixture with a mineral acid such as, for example, hydrochloric acid.

The intermediates of formula (XXX) can also be prepared from cyclization of an intermediate of formula (XXXIV).

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$$R^1$$
 R^2
 R^5
 R^5
 R^4
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3
 R^4
 R^3

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Said cyclization reaction is conveniently conducted in the presence of a Lewis acid, e.g. aluminium chloride, and the like. In some instances it may be appropriate to supplement the reaction mixture with a suitable amount of sodium chloride.

The intermediates of formula (V) can also be prepared from the cyclization of an intermediate of formula (III) in the presence of an acid in a reaction inert solvent.

An appropriate acid in the above reaction is, for example, a Lewis acid, e.g. tin(IV)chloride and the like. A suitable reaction-inert solvent is, for example, a halogenated hydrocarbon, e.g. dichloromethane, 1,2-dichloroethane, and the like.

The intermediates of formula (VI) can be prepared by reaction of a ketone of formula (XXXV) with an intermediate of formula (XXX) in the precence of e.g. lithium diisopropylamide in a reaction-inert solvent, e.g. tetrahydrofuran.

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The intermediates of formula (VII-c) can be prepared by N-alkylating a compound of formula (I-c) with a reagent of formula (XXXVI) following the procedures described hereinbefore for the preparation of the compounds of formula (I-e).

(I-c)
$$\frac{NC-C_{1.3}alkyl-W}{(XXXVI)} NC-C_{1.3}alkyl-N$$
(VII-c)

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The intermediates of formula (VII-d) can be prepared from the compounds of formula (I-k) wherein Y is oxygen by reaction with a halogenating reagent such as, for example, thionyl chloride, phosphorous trichloride, phosphoryl chloride and the like, or by reaction with a sulfonating reagent such as, for example, methanesulfonyl chloride, 4-methylbenzenesulfonyl chloride and the like.

$$HO-Alk-N$$
 $W-Alk-N$
 $W-A$

The intermediates of formula (XV) can be prepared by the following reaction sequence.

$$\begin{array}{c} \text{CH}_{3O} \\ \text{CH}_{3O} \\ \text{CH}_{2} - \text{NH}_{2} \\ \text{CH}_{3O} \\ \text{(XXXVI)} \end{array} \begin{array}{c} \text{CH}_{3} - \text{N} = \text{C} = \text{S} \\ \text{CH}_{3O} \\ \text{CH}_{2} - \text{NH} - \text{C} - \text{NH} - \text{CH}_{3} \\ \text{CH}_{3O} \\ \text{CH}_{3O} \\ \text{CH}_{3O} \\ \text{CH}_{2} - \text{NH} - \text{C} = \text{N} - \text{CH}_{3} \\ \text{CH}_{3O} \\$$

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The reaction of (XXXVI) with the isothiocyanate reagent can conveniently be conducted in a reaction-inert solvent such as, for example, an ether, e.g. tetrahydrofuran and the like. The resulting intermediate of formula (XXXVII) is methylated in a reaction-inert solvent such as, for example, a ketone, e.g. 2-propanone and the like.

The compounds of formula (XXX) intervening in the preparations described hereinbefore are novel, except for 2-methylimidazo[2,1-b][3]benzazepine, 2-phenylimidazo[2,1-b][3]benzazepine and 8,9-dimethoxyimidazo[2,1-b][3]benzazepine and have especially been developed for use as intermediates in said preparations. Consequently, the present invention also relates to novel compounds of formula

 R^1 R^2 R^5 R^5 R^4 R^3

the addition salt forms thereof and the stereochemically isomeric forms thereof, wherein R¹, R², R³, R⁴, and R⁵ are as defined under formula (I), 2-methylimidazo[2,1-b][3]benzazepine, 2-phenylimidazo[2,1-b][3]benzazepine and 8,9-dimethoxyimidazo[2,1-b][3]benzazepine being excluded.

The compounds of formula (I) and some of the compounds of formula (VII), in particular those wherein Q is (C₁₋₆alkyl or phenyl)oxycarbonyl, C₁₋₄alkylcarbonyl or C₁₋₆alkyl substituted with cyano or amino, the pharmaceutically acceptable acid addition salts and stereochemically isomeric forms thereof possess useful pharmacological properties. In particular they are active antiallergic agents, which activity can clearly be demonstrated by he test results obtained in a number of indicative tests.

- 25 Antihistaminic activity can be demonstrated in 'Protection of Rats from Compound 48/80 induced Lethality' test (Arch. Int. Pharmacodyn. Ther., 234, 164-176, 1978); 'Histamine induced Lethality in Guinea Pigs' test (Arch. Int. Pharmacodyn. Ther., 251, 39-51, 1981);
- and the broad antiallergic activity can be demonstrated in 'Passive cutaneous anaphylaxis in Rats' test (Drug Dev. Res., 5, 137-145, 1985) (For some compounds this test has been modified by replacing compound 48/80 by Ascaris

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allergens) and the

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'Ascaris Allergy in Dogs' test (Arch. Int. Pharmacodyn. Ther., <u>251</u>, 39-51, 1981 and Drug Dev. Res., <u>8</u>, 95-102, 1986).

The compounds of the present invention show a broad spectrum antiallergic profile as is evidenced by the results obtained in the diversity of test procedures cited hereinbefore.

A second advantageous feature of the compounds of the present invention resides in their excellent oral activity; the present compounds when administered orally have been found to be practically equipotent with the same being administered subcutaneously.

A particularly important asset of most of the present compounds is their lack of sedating properties at therapeutic dose levels, a troublesome side effect associated with many antihistaminic and antiallergic compounds. The non-sedating properties of the present compounds can be demonstrated, for example, by the results obtained in studying the sleep - wakefulness cycle of the rat (Psychopharmacology, <u>97</u>, 436-442, (1989)).

Another interesting feature of the present compounds relates to their fast onset of action and the favorable duration of their action.

In view of their antiallergic properties, the compounds of formula (I) and (VII), wherein Q is (C₁₋₆alkyl or phenyl)oxycarbonyl, C₁₋₄alkylcarbonyl or C₁₋₆alkyl substituted with cyano or amino, and their acid addition salts are very useful in the treatment of broad range of allergic diseases such as, for example, allergic rhinitis, allergic conjunctivitis, chronic urticaria, allergic asthma and the like.

In view of their useful antiallergic properties the subject compounds may be formulated into various pharmaceutical forms for administration purposes. To prepare the antiallergic compositions of this invention, an effective amount of the particular compound, in base or acid addition salt form, as the active ingredient is combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirably in unitary dosage form suitable, preferably, for administration orally, rectally, percutaneously, or by parenteral injection. For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups,

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elixirs and solutions: or solid carriers such as starches, sugars, kaolin, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules and tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit form, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example to aid solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined with suitable additives of any nature in minor proportions, which additives do not introduce a significant deleterious effect on the skin. Said additives may facilitate the administration to the skin and/or may be helpful for preparing the desired compositions. These compositions may be administered in various ways, e.g., as a transdermal patch, as a spot-on or as an ointment. Acid addition salts of the subject compounds due to their increased water solubility over the corresponding base form, are obviously more suitable in the preparation of aqueous compositions.

It is especially advantageous to formulate the aforementioned pharmaceutical compositions in dosage unit form for ease of administration and uniformity of dosage. Dosage unit form as used in the specification and claims herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such dosage unit forms are tablets (including scored or coated tablets), capsules, pills, powder packets, wafers, injectable solutions or suspensions, teaspoonfuls, tablespoonfuls and the like, and segregated multiples thereof.

30 The present invention also relates to a method of treating warm-blooded animals suffering from said allergic diseases by administering to said warm-blooded animals an effective antiallergic amount of a compound of formula (I) and (VII), wherein Q is (C₁₋₆alkyl or phenyl)oxycarbonyl, C₁₋₄alkylcarbonyl or C₁₋₆alkyl substituted with cyano or amino or a pharmaceutically acceptable acid addition salt form thereof.

In general it is contemplated that an effective antiallergic amount would be from about 0.001 mg/kg to about 20 mg/kg body weight, and more preferably from about 0.01 mg/kg to about 5 mg/kg body weight.

The following examples are intended to illustrate and not to limit the scope of the present invention in all its aspects.

5 Experimental part

A. Preparation of the intermediates

Example 1

- a) To a cooled mixture of 54.2 g of 1-(2-phenylethyl)-1H-imidazole, 34.7 g of N.N-diethylethanamine and 50 ml of pyridine there were added dropwise 69.2 g of ethyl
- 4-chlorocarbonyl-1-piperidinecarboxylate(temp. ≤.20 °C) and then 30 ml of acetonitrile. The whole was stirred for 2 hours at room temperature and for 4 hours at reflux temperature. After cooling, there were added 30 ml NaOH 50% and refluxing was continued for 1/2 hour. The cooled reaction mixture was evaporated and the residue was taken up in water. The product was extracted with dichloromethane and the extract was
- dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 97:3). The eluent of the desired fraction was evaporated and the residue was dried, yielding 38 g (33.9 %) of ethyl 4-[[1-(2-phenylethyl)-1<u>H</u>-imidazol-2-yl]carbonyl]-1-piperidinecarboxylate (interm. 1).

In a similar manner there was also prepared:

- ethyl 4-[[1-[2-(2-chlorophenyl)ethyl]-1<u>H</u>-imidazol-2-yl]carbonyl]-1-piperidine-carboxylate (interm. 37).
 - b) A mixture of 9 g of intermediate (1) and 50 ml of hydrobromic acid 48% was stirred for 5 hours at 80°C. The reaction mixture was evaporated and the residue was boiled in 2-propanol. After cooling, the precipitate was filtered off and dried, yielding 10.85 g
- 25 (97.5%) of [1-(2-phenylethyl)-1<u>H</u>-imidazol-2-yl] (4-piperidinyl)methanone dihydrobromide; mp. 275.3 °C (interm. 2).

In a similar manner there was also prepared:

In a similar manner there were also prepared:

- [1-[2-(2-methylphenyl)ethyl]-1<u>H</u>-imidazol-2-yl](4-piperidinyl)methanone dihydrobromide hemihydrate; mp. 231.7°C (interm. 38).
- c) A mixture of 55 g of intermediate (2), 70 ml of formaldehyde and 70 ml of formic acid was stirred for 5 hours at reflux temperature. After cooling, the reaction mixture was diluted with water and basified with NaOH(aq.). The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 90:10). The eluent of the desired fraction was evaporated and the residue was dried, yielding 30 g (82.0%) of (1-methyl-4-piperidinyl) [1-(2-phenylethyl)-1H-imidazol-2-yl]methanone (interm. 3).

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[1-[2-(4-fluorophenyl)ethyl]-1<u>H</u>-imidazol-2-yl] (1-methyl-4-piperidinyl)methanone (interm. 4); and

[1-[2-(2-chlorophenyl)ethyl]-1<u>H</u>-imidazol-2-yl] (1-methyl-4-piperidinyl)methanone (interm. 39).

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Example 2

A mixture of 70.6 g of intermediate (2) and 700 ml of methanol was hydrogenated at normal pressure and at room temperature in the presence of 2 g of platinum-on-charcoal catalyst 5%. After the calculated amount of hydrogen was taken up, the catalyst was filtered off and the filtrate was evaporated. The residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 54 g (75.7 %) of α-[1-(2-phenylethyl)-1H-imidazol-2-yl]-4-piperidinemethanol; mp. 144.6 °C (interm. 5).

Example 3

- a) A mixture of 28.9 g of 2-(4-methylphenyl)ethanol methanesulfonate, 18.6 g of 1H-imidazole, 22.7 g of potassium carbonate and 600 ml of tetrahydrofuran was stirred for 18 hours at reflux temperature. After cooling, the reaction mixture was evaporated and the residue was taken up in water and extracted with 4-methyl-2-pentanone. The extract was dried, filtered and evaporated and the residue was destilled (13.3 Pa; 120)
- °C), yielding 20.1 g (83.0%) of 1-[2-(4-methylphenyl)ethyl]-1H-imidazole (interm. 6). In a similar manner there were also prepared:
 - 1-[2-(3-methylphenyl)ethyl]-1H-imidazole; bp. 120°C at 13.3 Pa (interm. 7),
 - 1-[2-(4-bromophenyl)ethyl]-1H-imidazole (interm. 8), and
 - 1-[2-(3-chlorophenyl)ethyl]-1H-imidazole; bp. 134°C at 13.3 Pa (interm. 9).
- b) A mixture of 67 g of 1-(2-chloroethyl)-3-methoxybenzene, 53.1 g of 1<u>H</u>-imidazole, 99 g of sodium carbonate, 500 ml of 4-methyl-2-pentanone and a few crystals of potassium iodide was stirred for 48 hours at reflux temperature. After cooling, the reaction mixture was diluted with water and extracted with 4-methyl-2-pentanone. The extract was dried, filtered and evaporated and the residue was destilled (13.3 Pa;
- 30 160 °C), yielding 49.5 g (62.8%) of 1-[2-(3-methoxyphenyl)ethyl]-1<u>H</u>-imidazole (interm. 10).

Example 4

a) To a stirred amount of 250 ml of N,N-dimethylformamide under nitrogen, there were
 added portionwise 6 g of a dispersion of sodium hydride in mineral oil and 82.1 g of
 4-methylimidazole and then dropwise 132 g of phenylxirane. The whole was stirred for
 50 hours and then diluted with 1000 ml of water. The precipitate was filtered off,

washed with water and 2,2'-oxybispropane and recrystallized from a mixture of acetonitrile and ethanol. The product was filtered off and dried, yielding 58.1 parts (28.7%) of 5-methyl- α -phenyl- $1\underline{H}$ -imidazole-1-ethanol; mp. 192.7 °C (interm. 11).

b) A mixture of 57.1 g of intermediate (11), 130 ml of 2-propanol saturated with HCl and 500 ml of methanol was hydrogenated at normal pressure and at room temperature in the presence of 5 g of palladium-on-charcoal catalyst 10%. After the calculated amount of hydrogen was taken up, the catalyst was filtered off and the filtrate was evaporated. The residue was diluted with water and the whole was basified with NaOH(aq.). The product was extracted with dichloromethane and the extracted was dried, filtered and evaporated. The residue was co-evaporated with methylbenzene (3x), yielding 52.9 g (100%) of 5-methyl-1-(2-phenylethyl)-1H-imidazole (interm. 12).

In a similar manner there was also prepared:

1-[2-(2-methylphenyl)ethyl]-1H-imidazole (interm. 49).

. 15 <u>Example 5</u>

- a) To a cooled mixture (ice-bath) of 10.1 g of intermediate (10), 12 g of N,N-diethylethanamine and 150 ml of acetonitrile there were added dropwise 21.95 g of ethyl 4-chlorocarbonyl-1-piperidinecarboxylate, keeping the temperature below 20 °C. After stirring for 2 hours at room temperature and 4 hours at reflux temperature, there were
- added dropwise 10 ml NaOH. The whole was refluxed for 1/2 hour, cooled and evaporated. The residue was taken up in water and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated, yielding 22 g (100%) of ethyl 4-[[1-[2-(3-methoxyphenyl)ethyl]-1H-imidazol-2-yl]-carbonyl]-1-piperidine-carboxylate (interm. 13).
- In a similar manner there were also prepared:

 ethyl 4-[[1-[2-(3-chlorophenyl)ethyl]-1<u>H</u>-imidazol-2-yl]carbonyl]-1-piperidinecarboxylate (interm. 14),

 1-acetyl-4-[[1-[2-(4-methylphenyl)ethyl]-1<u>H</u>-imidazol-2-yl]carbonyl]piperidine
 (interm. 15),
- 30 ethyl 4-[[5-methyl-1-(2-phenylethyl)-1H-imidazol-2-yl]carbonyl]-1-piperidine-carboxylate (interm. 16),
 ethyl 4-[[1-[2-(3-methylphenyl)ethyl]-1H-imidazol-2-yl]carbonyl]-1-piperidine-carboxylate (interm. 17),
- ethyl 4-[[1-[2-(4-bromophenyl)ethyl]-1<u>H</u>-imidazol-2-yl]carbonyl]-1-piperidine-35 carboxylate (interm. 18), and ethyl 4-[[1-[2-(2-methylphenyl)ethyl]-1H-imidazol-2-yl]carbonyl]-1-piperidinecarboxylate (interm. 40).

- b) A mixture of 4.4 g of intermediate (13) and 120 ml of hydrochloric acid 12N was stirred for 72 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water, basified with NaOH and extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 90:10). The eluent of the desired
- 5 chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 90:10). The eluent of the desired fraction was evaporated, yielding 2.63 g (83.9 %) of [1-[2-(3-methoxyphenyl)ethyl]-1H-imidazol-2-yl] (4-piperidinyl)methanone (interm. 19).

In a similar manner there were also prepared:

- $[1-[2-(4-methylphenyl)ethyl]-1\underline{H}-imidazol-2-yl]$ (4-piperidinyl)methanone
- 10 dihydrochloride (interm. 20),
 - [1-[2-(3-chlorophenyl)ethyl]- $1\underline{H}$ -imidazol-2-yl] (4-piperidinyl)methanone (interm. 21), and
 - [1-[2-(2-methylphenyl)ethyl]-1H-imidazol-2-yl] (4-piperidinyl)methanone dihydrobromide; mp. 268.1°C (interm. 41).
- c) A mixture of 130 g of intermediate (16) and 1000 ml of hydrobromic acid 48% was stirred for 24 hours at 80 °C. The reaction mixture was evaporated and the residue was recrystallized from 2-propanol. After cooling, the precipitate was filtered off and dried, yielding 124.2 g (95.6%) of [5-methyl-1-(2-phenylethyl)-1H-imidazol-2-yl] (4-piperidinyl)methanone dihydrobromide (interm. 22).
- In a similar manner there were also prepared:

 [1-[2-(3-methylphenyl)ethyl]-1H-imidazol-2-yl] (4-piperidinyl)methanone dihydrobromide (interm. 23), and

 [1-[2-(4-bromophenyl)ethyl]-1H-imidazol-2-yl] (4-piperidinyl)methanone dihydrobromide hemihydrate (interm. 24).

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Example 6

A mixture of 5.24 g of intermediate (24), 2 g of polyoxymethylene, 3 g of potassium acetate, 2 ml of a solution of thiophene in methanol 4% and 150 ml of methanol was hydrogenated at normal pressure and at room temperature in the presence of 2 g of platinum-on-charcoal catalyst 5%. After the calculated amount of hydrogen was taken up, the catalyst was filtered off and the filtrate was evaporated. The residue was taken up in water and the whole was basified with K₂CO₃. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated, yielding 3.2 g (85.0%) of [1-[2-(4-bromophenyl)ethyl]-1H-imidazol-2-yl] (1-methyl-4-piperidinyl)-methanone (interm. 25).

In a similar manner there were also prepared:

[1-[2-(3-chlorophenyl)ethyl]-1<u>H</u>-imidazol-2-yl] (1-methyl-4-piperidinyl)methanone (interm. 26), and

[1-[2-(3-methoxyphenyl)ethyl]-1<u>H</u>-imidazol-2-yl] (1-methyl-4-piperidinyl)methanone (interm. 27).

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Example 7

- a) A mixture of 3.16 g of $1\underline{H}$ -3-benzazepin-2-amine, 4.17 g of 2,2-dimethoxy-ethanamine and 50 ml of methanol was stirred for 16 hours at reflux temperature. The reaction mixture was evaporated and the residue was purified by column chromatography (silica gel; CH_2Cl_2/CH_3OH 95:5). The eluent of the desired fraction was evaporated and the residue was stirred in hexane. The precipitate was filtered off, yielding 4.9 g (100%) of \underline{N} -(2,2-dimethoxyethyl)- $1\underline{H}$ -3-benzazepin-2-amine (interm. 28).
- b) A mixture of 4.9 g of intermediate (28), 70 ml of acetic acid and 9 ml of hydrochloric acid 36% was stirred for 18 hours at 70°C. The reaction mixture was evaporated and the residue was taken up in water. The whole was basified with NaOH(aq.) and extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was treated with active charcoal in 1,1'-oxybisethane. The whole was filtered and the filtrate was evaporated. The residue was triturated in hexane. The product was filtered off and dried, yielding 1.04 g (28.5%) of 11H-imidazo[2,1-b][3]benzazepine; mp. 85.5 °C (interm. 29).
 - c) A mixture of 5 g of intermediate (29), 20 g of manganese dioxide and 150 ml of trichloromethane was stirred for 50 hours at reflux temperature. The whole was filtered over diatomaceous earth, 20 g of manganese dioxide were added and refluxing was continued for 48 hours (2x). The reaction mixture was filtered and the filtrate was evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was triturated in 1,1'-oxybisethane and then boiled in acetonitrile. After cooling, the product was filtered off and dried, yielding 2.61 g (53.2%) of 11H-imidazo-[2,1-b][3]benza-zepin-11-one; mp. 218.9 °C (interm. 30).
 - d) A mixture of 10 ml of tetrahydrofuran and 1.24 g of magnesium was stirred under a nitrogen atmosphere. 1 Crystal of iodine and then dropwise 1.2 g of bromoethane were added and at reflux tempereature there was added a solution of 6.7 g of 4-chloro-1-methylpiperidine in 25 ml of tetrahydrofuran. After refluxing for 1 hour, the reaction mixture was cooled (0 °C). Then there were added 25 ml of tetrahydrofuran and portionwise 9.8 parts of intermediate (30), keeping the temperature below 10 °C. The whole

was stirred for 1 hour at room temperature and decomposed with NH₄Cl (aq.). The product was extracted with trichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH (NH₃) 95:5). The eluent of the second fraction was evaporated and the residue was crystallized from acetonitrile in 2 fractions, yielding 4.76 parts (32.2%) of 11-(1-methyl-4-piperidinyl)-11<u>H</u>-imidazo[2,1-b][3]benzazepin-11-ol; mp. 155.2 °C (interm. 31).

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Following the procedure of example 10 (c) and (d) 2-phenyl-11<u>H</u>-imidazo[2,1-b][3] benzazepine-11-one was converted into 11-(1-methyl-4-piperidinyl)-2-phenyl-11<u>H</u>-imidazo[2,1-b][3]benzazepin-11-ol; mp. 239.8 °C (interm. 32).

A mixture of 6 g of intermediate (32) and 300 ml of methanol was hydrogenated at normal pressure and at room temperature in the presence of 3 g of palladium-on-charcoal catalyst 10%. After the calculated amount of hydrogen was taken up, the catalyst was filtered off and the filtrate was evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5 → CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 3.2 g (53.5%) of 6,11-dihydro-11-(1-methyl-4-piperidinyl)-2-phenyl-5H-imidazo[2,1-b][3]benzazepin-11-ol; mp. 225.3 °C (interm. 33).

Example 9

- a) To a cooled (0°C) mixture of 46.2 g of 3-fluorobenzenethanol, 40 ml of N,N-diethylethanamine and 500 ml of dichloromethane, there were added dropwise 41.2 g of methanesulfonyl chloride, keeping the temperature below 5°C. After stirring for 18 hours at room temperature, the reaction mixture was diluted with water and extracted with dichloromethane. The extract was dried, filtered and evaporated, yielding 81 g (100%) of 2-(3-fluorophenyl)ethanol methanesulfonate (ester) (interm. 34).
 - b) A mixture of 72 g of intermediate (34), 45 g of 1H-imidazole, 55.5 g of potassium carbonate and 1000 ml of tetrahydrofuran was stirred over weekend at reflux temperature. The reaction mixture was filtered over diatomaceous earth and the filtrate was evaporated. The residue was taken up in water and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of

the desired fraction was evaporated and the residue was distilled (53.2 Pa; 130 °C), yielding 37.8 parts (60.2%) of 1-[2-(3-fluorophenyl)ethyl]-1 \underline{H} -imidazole (interm. 35). c) To a cooled (-70 °C) mixture of 5.5 g of 2-methyl- \underline{N} -(1-methylethyl)ethanamine and 100 ml of tetrahydrofuran under a nitrogen atmosphere there were added dropwise 22 ml of butyllithium and after stirring for 15 min. at -40 °C, 9.5 g of intermediate (35) at -70°C. Stirring at -70 °C was continued for 1 hour and then there were added 9.4 g of ethyl 1-methyl-4-piperidinecarboxylate. The whole was stirred for 18 hours at room temperature, decomposed with water and evaporated. The residue was taken up in water and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5 \rightarrow CH₂Cl₂ / CH₃OH 80:20). The eluent of the desired fraction was evaporated, yielding 8 g (50.7%) of [1-[2-(3-fluorophenyl)ethyl]-1 \underline{H} -imidazol-2-yl] (1-methyl-4-piperidinyl)methanone (interm. 36).

15 Example 10

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a) To a stirred and cooled (-70°C) mixture of 18.8 g of N-(1-methylethyl)-2-propanamine in 200 ml of tetrahydrofuran (under nitrogen atmosphere) were added portionwise 42 ml of butyllithium 2.5M in hexane. The mixture was brought to -40°C and stirred at this temperature for 15 minutes. The mixture was cooled again to -70°C and a solution of 17 g of 1-(diethoxymethyl)-1H-imidazole in tetrahydrofuran was added dropwise at this temperature. Stirring was continued for 1 hour and a solution of 18.8 g of ethyl 1-methyl-4-piperidinecarboxylate in 200 ml of tetrahydrofuran was added. After stirring for 1 hour at -70°C and for another hour at room temperature, the mixture was decomposed with water, acidified with HCl and evaporated. The residue was taken up in water, alkalized with potassium carbonate and extracted with a mixture of dichloromethane and methanol. The extract was dried, filtered and evaporated. The residue was purified on silica (eluent: CH₂Cl₂/(CH₃OH/NH₃) 95/5). The pure fractions were collected and the eluent was evaporated. The residue was crystallized from acetonitrile. The product was filtered off and dried, yielding: 2.75 g of (1H-imidazol-2-yl)(1-methyl-4-piperidinyl)methanone (12.9%); mp. 143.6°C (interm. 42).

b) To 200 ml of N,N-dimethylformamide were added portionwise 13.2 g of a sodium hydride dispersion 50% in mineral oil and then 48.3 g of intermediate (42) under nitrogen atmosphere while stirring. After stirring for 1.5 hours at room temperature, a solution of 65 g of 2-fluorobenzeneethanol methanesulfonate (ester) in N,N-dimethylormamide was added dropwise to the reaction mixture. The reaction mixture was stirred for 18 hours at 60°C, cooled and decomposed with water. The product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was

taken up in water, acidified with hydrochloric acid, washed twice with 2,2'-oxybis-propane, treated with potassium carbonate and extracted again with dichloromethane. The extract was dried, filtered and evaporated. The residue was converted into the (E)-2-butenedioate salt in ethanol. The salt was filtered off and dried, yielding 61.9 g (50.6%) of [1-[2-(2-fluorophenyl)ethyl]-1H-imidazol-2-yl](1-methyl-4-piperidinyl)methanone (E)-2-butenedioate (2:3); mp. 131.7°C (interm. 43).

Example 11

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22.3 g of methyl 4'-methyl-(1,1'biphenyl)-2-carboxylate were dissolved in 900 ml of tetrachloromethane under a nitrogen flow. Then there were added 17.8 g of 1-bromo-2,5-pyrrolidinedione and a catalytic amount of dibenzoyl peroxide. After stirring for 2.5 hours at reflux temperature under a nitrogen atmosphere, the reaction mixture was cooled and filtered. The filtrate was evaporated, yielding > 30 g (100%) of methyl 4'-(bromomethyl)[1,1'-biphenyl]-2-carboxylate as a crude residue (interm. 44).

Example 12

- a) To a freshly prepared sodium methoxide solution, prepared in the usual manner starting from 23 g of sodium and 350 ml of methanol was added a solution of 68 g of 1H-imidazole in 100 ml of methanol. The solvent was evaporated and the residue was taken up in 320 ml of N,N-dimethylformamide. The solvent was removed again till the temperature rose to 125°C. After cooling to 30°C, 185 g of (2-bromoethyl)benzene were added to the residue and the whole was stirred overnight. The reaction mixture was diluted with 1500 ml of water and 230 ml of benzene. The separated aqueous layer was extracted twice with benzene. The combined organic layers were treated with 750 ml of a hydrochloric acid solution 4 N and than basified. The product was extracted with benzene. The extract was dried, filtered and evaporated. The oily residue was distilled in vacuo, yielding 55 g of 1-(2-phenylethyl)-1H-imidazole; bp. 140-145°C at 23.3 Pa (interm. 45).
- b) A mixture of 34.5 g of intermediate (45) and 200 ml of formaldehyde 37% in water was stirred and refluxed for 48 hours. After evaporation, the residue was taken up in water and treated with a diluted ammonium hydroxide solution while cooling. The whole was stirred for 30 minutes and extracted with methylbenzene. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (eluent: CH₂Cl₂/CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was stirred in 2,2'-oxybispropane. The precipitated product was filtered off and dried in vacuo, yielding 29.9 g (73.8%) of 1-(2-phenylethyl)-1H-imidazole-2-methanol; mp. 75.4°C (interm. 46).

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- c) To 50 ml of thionyl chloride were added portionwise 4 g of intermediate (46). The reaction mixture was stirred and refluxed for 30 minutes. The reaction mixture was evaporated and the residue was stirred in 2,2'-oxybispropane. The precipitated product was filtered off and dried, yielding 4.61 g (89.6%) of 2-(chloromethyl)-1-(2-phenyl-ethyl)-1<u>H</u>-imidazole monohydrochloride; mp. 240.2°C (interm. 47).
- d) A mixture of 19.6 g of intermediate 47, 59 g of aluminium chloride and 25.5 g of sodium chloride was stirred for 30 minutes at 100°C. After cooling, the reaction mixture was poured into ice water and treated with sodium hydroxide. The product was extracted with methylbenzene. The extract was dried, filtered and evaporated, yielding 13.1 g (93.5%) of 6,11-dihydro-5H-imidazo[2,1-b][3]benzazepine (interm. 48).

B. Preparation of the final compounds

Example 13

A mixture of 2.5 g of intermediate (26) and 10 ml of trifluoromethanesulfonic acid was stirred for 72 hours at 110°C under nitrogen. After cooling, the reaction mixture was poured into ice-water and the whole was basified with NaOH (aq.). The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH (NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 0.95 g (40.4%) of 8-chloro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5H-imidazo[2,1-b][3]-benzaze-pine; mp. 186.6°C (comp. 3.10).

Example 14

A mixture of 2 g of intermediate (27) and 10 ml of methanesulfonic acid was stirred for 1 hour at 100°C. The reaction mixture was poured into ice-water and the whole was basified with NaOH (aq.). The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate (1:2) salt in 2-propanone. The salt was filtered off and dried, yielding 1 g (30.8%) of 6,11-dihydro-8-methoxy-11-(1-methyl-4-piperidinylidene)-5H-imidazo[2,1-b][3]-benzazepine (Z)-2-butenedioate(1:2); mp. 190.3°C (comp. 3.01).

35 <u>Example 15</u>

A mixture of 8 g of intermediate (36), 24 g of aluminum chloride and 10.3 g of sodium chloride was stirred at 140°C until the whole was melted. Stirring was continued for

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1 hour at 120°C. The reaction mixture was poured into ice-water and the whole was basified with NaOH (aq.). The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5 → CH₂Cl₂ / CH₃OH(NH₃) 90:10).

The eluent of the desired fraction was evaporated and the residue was triturated in 2,2'-oxybispropane and recrystallized from 4-methyl-2-pentanone. The product was filtered off and dried, yielding 0.58 g (10.8%) of 8-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5H-imidazo[2,1-b][3]benzazepine; mp. 152.4°C (comp. 3.15).

10 Example 16

A mixture of 3.5 g of intermediate (5) and 10 ml of trifluoromethanesulfonic acid was stirred for 18 hours at 110°C. The reaction mixture was poured into ice-water and the whole was basified with NaOH (aq.). The product was extracted with dichloromethane and the extract was washed with water, dried, filtered and evaporated. The residue was converted into the (E)-2-butenedioate (1:2) salt in ethanol. The salt was recrystallized from ethanol, yielding 0.8 g (13.3%) of 6,11-dihydro-11-(4-piperidinyl)-5H-imidazo-[2,1-b][3]benzazepine (E)-2-butenedioate (1:2); mp. 220.2°C (comp. 5.01).

Example 17

- A mixture of 2.2 g of intermediate (33), 10 ml of sulfuric acid and 10 ml of methane-sulfonic acid was stirred for 2 hours at 70°C. The reaction mixture was poured into icewater and the whole was basified with NaOH (aq.). The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5 → CH₂Cl₂ /
- 25 CH₃OH(NH₃) 90:10). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 0.73 g (34.2%) of 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-2-phenyl-5<u>H</u>-imidazo-[2,1-b][3]benzazepine; mp. 171.5°C (comp. 4.01).

30 Example 18

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A mixture of 14.7 g of intermediate (31) and 150 ml of acetic anhydride was stirred for 2 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. The whole was basified with NaOH (aq.) and then extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH_2Cl_2 / CH_3OH 95:5 \rightarrow CH_2Cl_2 / CH_3OH (NH₃) 95:5). The eluent of the first fraction was evaporated and the residue was taken up in 1,1'-oxybisethane. The whole was filtered and the filtrate was treated with

activated charcoal. After filtration, the solution was evaporated and the residue was triturated in 2,2'-oxybispropane. The product was filtered off and dried, yielding 1.6 g (11.5%) of product. The second fraction was also evaporated and the residue taken up in 1,1'-oxybisethane. The whole was filtered and the filtrate was combined with the 2,2'-oxybispropane-filtrate of the first fraction, and evaporated, yielding an additional 8.2 g (59.1%) of product. Total yield: 9.8 g (70.6%) of 11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepine; mp. 135.8°C (comp. 6.01).

Example 19

To a stirred and refluxing mixture of 7.2 g of compound (3.10), 4.6 g of N,N-diethyl-ethanamine and 200 ml of methylbenzene there were added dropwise 12.5 g of ethyl chloroformate. After refluxing for 1 hour and subsequent cooling, the reaction mixture was diluted with water. The whole was basified with K₂CO₃ and then extracted with methylbenzene. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was stirred in 1,1'-oxybisethane. The product was filtered off and dried, yielding 6.62 g (77.4%) of ethyl 4-(8-chloro-5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidine-carboxylate; mp. 140.3°C (comp. 3.11).

Example 20

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- a) A mixture of 2.5 g of compound (1.03) and 50 ml of formaldehyde 40% was stirred for 1 week at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. The whole was basified with NH₄OH, stirred for 1/2 hour and extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 0.45 g (16.3%) of ethyl 4-[5,6-dihydro-3-(hydroxymethyl)-11H-imidazo[2,1-b][3]benzazepin-11-ylidene]-1-piperidinecarboxylate; mp. 191.9°C (comp. 4.11).
- b) A mixture of 20 g of compound (1.03) and 400 ml of formaldehyde 40% was stirred for 2 weeks at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. After basifying with NH₄OH, the product was extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH_2Cl_2 / CH_3OH 95:5 $\rightarrow CH_2Cl_2 / CH_3OH$ (NH₃) 95:5). The eluent of the third fraction was evaporated, yielding 4.1 g (17.2%) of ethyl 4-[5,6-dihydro-2,3-bis(hydroxymethyl)-11H-imidazo-[2,1-b][3]-

benzazepin-11-ylidene]-1-piperidinecarboxylate (comp. 4.18).

Example 21

A mixture of 13 g of compound (1.03), 13 g of potassium hydroxide and 100 ml of 2-propanol was stirred for 6 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 90:10). The eluent of the desired fraction was evaporated and the residue was converted into the (E)-2-butenedioate (1:2) salt in ethanol. The salt was filtered off and dried, yielding 3.52 g (18.3%) of 6,11-dihydro-11-(4-piperidinylidene)-5H-imidazo[2,1-b][3]benzazepine (E)-2-butenedioate (1:2) hemihydrate; mp. 192.5°C (comp. 1.04).

Example 22

A mixture of 60 g of compound (6.02) and 500 ml of hydrobromic acid 48% was stirred for 5 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. After basifying with NaOH (aq.), the product was extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 95:5 → CH₂Cl₂ / CH₃OH(NH₃) 90:10). The eluent of the first fraction was evaporated and the residue was converted into the dihydrobromide salt in ethanol. The salt was filtered off and dried, yielding 27.3 g (37.7%) of 11-(4-piperidinylidene)-11H-imidazo[2,1-b]-[3]benzazepine dihydrobromide hemihydrate; mp. 246.9°C (comp. 6.03).

25 Example 23

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A mixture of 6.1 g of compound (3.11) and 100 ml of hydrochloric acid 12N was stirred for 18 hours at reflux temperature. The reaction mixture was evaporated and the residue was boiled in 2-propanol. After cooling, the precipitate was filtered off and taken up in water. The whole was basified with NaOH (aq.) and then extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was boiled in acetonitrile. After cooling, the product was filtered off and dried, yielding 2.9 g (59.0%) of 8-chloro-6,11-dihydro-11-(4-piperidinylidene)-5H-imidazo[2,1-b]-[3]benzazepine; mp. 197.1°C (comp. 3.12).

35 <u>Example 24</u>

To a stirred and cooled (ice-bath) mixture of 5.6 g of compound (2.12), 50 ml of dichloromethane and 2.5 g of N,N-diethylethanamine there was added dropwise a

solution of 2.38 g of ethyl chloroformate in 20 ml of dichloromethane. Stirring was continued for 1 hour at room temperature. The reaction mixture was diluted with water and extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5).

The eluent of the desired fraction was evaporated and the residue was stirred in 1,1'-oxybisethane. The product was filtered off and dried, yielding 2.85 g (40.5%) of ethyl 4-(5,6-dihydro-9-methyl-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinecarboxylate; mp. 156.5°C (comp. 2.13).

10 <u>Example 25</u>

A mixture of 1.79 g of 3-(2-chloroethyl)-2-oxazolidinone, 2.65 g of compound (1.04), 1.3 g of sodium carbonate, 150 ml of 4-methyl-2-pentanone and 1 g of potassium iodide was stirred for 18 hours at reflux temperature. After cooling, the reaction mixture was diluted with water. The aqueous layer was separated and extracted with dichloromethane. The combined organic layers were dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the (E)-2-butenedioate (2:3) salt in ethanol. The salt was filtered off and dried, yielding 3.4 g (61.5%) of 3-[2-[4-[5,6-dihydro-11H-imidazo[2,1-b] [3]benzazepin-11-ylidene]-1-piperidinyl]ethyl]-2-oxazolidinone (E)-2-butenedioate (2:3); mp. 188.8°C (comp. 1.20).

Example 26

A mixture of 2.3 g of 6-(2-chloroethyl)-7-methylthiazolo[3,2-a]pyrimidin-5-one, 2.65 g of compound (1.04), 1.3 g of sodium carbonate and 100 ml of 4-methyl-2-pentanone was stirred for 24 hours at reflux temperature. After cooling, the reaction mixture was diluted with water. The product was extracted with 4-methyl-2-pentanone and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from 2-propanone. The product was filtered off and dried, yielding 1.89 g (41.3%) of 6-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b]-[3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]-7-methyl-5H-thiazolo[3,2-a]pyrimidin-5-one; mp. 181.8°C (comp. 1.13).

Example 27

A mixture of 0.83 g of chloroacetonitrile, 2.65 g of compound (1.04), 1.1 g of N,N-diethylethanamine and 80 ml of N,N-dimethylacetamide was stirred for 18 hours at room temperature. The reaction mixture was poured into water and the product was

extracted with dichloromethane. The extract was dried, filtered and evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 2.0 g (65.7%) of 4-(5,6-dihydro-11H-imidazo[2,1-b][3]-benzazepin-11-ylidene)-1-piperidineacetonitrile; mp. 220.4°C (comp. 1.26).

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Example 28

A mixture of 1.0 g of 3-chloro-2-methyl-1-propene, 2.6 g of compound (1.04), 1.6 g of sodium carbonate and 50 ml of N,N-dimethylacetamide was stirred for 20 hours at 50°C. After cooling, there were added 100 ml of ethyl acetate. The whole was washed with water (3x), dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the (E)-2-butene-dioate (2:3) salt in 2-propanol. The salt was filtered off and dried, yielding 2.8 g (56.7%) of 6,11-dihydro-11-[1-(2-methyl-2-propenyl)-4-piperidinylidene]-5H-imidazo[2,1-b] [3]benzazepine (E)-2-butenedioate (2:3); mp. 179.5°C (comp.1.08).

Example 29

A mixture of 1.57 g of 4-chloro-2-methyl-2-butene (dissolved in N,N-dimethyl-formamide), 2.65 g of compound (1.04), 1.1 g of sodium carbonate, 0.01 g of potassium iodide and 100 ml of N,N-dimethylacetamide was stirred for 18 hours at room temperature. The reaction mixture was poured into water and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified twice by column chromatography (silica gel; CH₂Cl₂ / CH₃OH / CH₃OH(NH₃) 90:10:1; HPLC; Lichroprep RP18; CH₃COONH₄ in H₂O 0.5% / CH₃OH / CH₃CN 40:55:5). The eluent of the desired fraction was evaporated and the residue was crystallized from 2,2'-oxybispropane. The product was filtered off and dried, yielding 0.25 g (7.5%) of 6,11-dihydro-11-[1-(3-methyl-2-butenyl)-4-piperidinylidene]-5H-imidazo[2,1-b][3]benzazepine; mp. 127.2°C (comp. 1.09).

30 <u>Example 30</u>

A mixture of 19 g of compound (2.03), 6 g of chloroacetonitrile, 8 g of N,N-diethyl-ethanamine and 100 ml of N,N-dimethylformamide was stirred for 18 hours at room temperature. The reaction mixture was evaporated and the residue was taken up in water. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 4.15 g

(19.2%) of 4-(9-fluoro-5,6-dihydro-11<u>H</u>-imidazo[2,1-b][3]-benzazepin-11-ylidene)-1-piperidineacetonitrile; mp. 198.3°C (comp. 2.08).

Example 31

To a stirred mixture of 2.83 g of compound (2.03), 2.12 g of sodium carbonate, 50 ml of N,N-dimethylformamide and 1 g of potassium iodide there were added dropwise 25.4 g of 4-chloro-2-methyl-2-butene (dissolved in N,N-dimethyl-formamide). Stirring at room temperature was continued for 50 hours. The reaction mixture was diluted with water and extracted with 4-methyl-2-pentanone. The extract was dried, filtered and evaporated and the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate (1:2) salt in 2-propanone. The salt was filtered off and dried, yielding 2.65 g (45.4%) of 9-fluoro-6,11-dihydro-11-[1-(3-methyl-2-butenyl)-4-piperidinylidene]-5H-imidazo-[2,1-b][3]benzazepine (Z)-2-butenedioate (1:2); mp. 203.4°C (comp. 2.04).

Example 32

A mixture of 1.5 g of 3-bromo-1-propene, 2.65 g of compound (1.04), 1.0 g of sodium hydrogen carbonate and 100 ml of ethanol was stirred for 5 hours at reflux temperature.

The reaction mixture was evaporated and the residue was taken up in water. The product was extracted with 4-methyl-2-pentanone and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH / CH₃OH(NH₃) 90:10:0 → 90:10:1). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butene-dioate (1:2) salt in

2-propanone. The salt was filtered off and dried for 2 hours in vacuo at 100°C, yielding 1.1 g (20.5%) of 6,11-dihydro-11-[1-(2-propenyl)-4-piperidinylidene]-5H-imidazo-[2,1-b][3]benzazepine (Z)-2-butenedioate (1:2); mp. 160.8°C (comp. 1.07).

Example 33

- A mixture of 2.7 g of compound (3.04), 1 g of polyoxymethylene, 2 ml of a solution of thiophene in methanol 4% and 150 ml of methanol was hydrogenated at normal pressure and 50°C in the presence of 1 g of palladium-on-charcoal catalyst 10%. After the calculated amount of hydrogen was taken up, the catalyst was filtered off and the filtrate was evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and
- 35 CH₂Cl₂ / CH₃OH(NH₃) 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the (E)-2-butenedioate (1:2) salt in 2-propanol. The salt was filtered off and dried, yielding 3.1 g (59.0%) of 6,11-dihydro-8-methyl-11-

 $(1-\text{methyl-}4-\text{piperidinylidene})-5\underline{H}-\text{imidazo}[2,1-b][3]$ benzazepine (E)-2-butenedioate (1:2); mp. 211.0°C (comp. 3.05).

Example 34

A mixture of 2.7 g of compound (5.01), 2 g of polyoxymethylene, 2 ml of a solution of thiophene in methanol 4% and 150 ml of methanol was hydrogenated at normal pressure and room temperature in the presence of 2 g of platinum-on-charcoal catalyst 5%. After the calculated amount of hydrogen was taken up, the catalyst was filtered off and the filtrate was evaporated. The residue was partitioned between dichloromethane and

NH₄OH. The aqueous layer was separated and re-extracted with dichloromethane. The combined organic layers were dried, filtered and evaporated. The residue was crystallized from a mixture of 2,2'-oxybispropane and acetonitrile (2x). The product was filtered off and dried, yielding 0.76 g (26.2%) of 6,11-dihydro-11-(1-methyl-4-piperidinyl)-5H-imidazo[2,1-b][3]benzazepine hemihydrate; mp. 117.8°C (comp. 5.02).

Example 35

A mixture of 2.65 g of compound (1.04), 20 ml of acetic acid and 15 ml of 2-propanone was stirred for 2 hours at room temperature under nitrogen. There were added portionwise 1.5 g of sodium tetrahydroborate and stirring was continued for 18 hours. The reaction mixture was diluted with water and basified with NaOH 15%. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH / CH₃OH(NH₃) 90:10:1). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate (1:2) salt in 2-propanone. The salt was filtered off and dried, yielding 2.5 g (46.3%) of 6,11-dihydro-11-[1-(1-methylethyl)-4-piperidinylidene]-5H-imidazo[2,1-b][3]benzazepine (Z)-2-butenedioate (1:2); mp. 183.6°C (comp. 1.06).

Example 36

A mixture of 4 g of compound (4.03), 2 ml of acetic acid, 3 g of sodium acetate and 20 ml of formaldehyde 37% was stirred for 50 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. The whole was basified with NaOH (aq.) and extracted with a mixture of dichloromethane and methanol. The extract was dried, filtered and evaporated and the residue was purified by column
 chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5 → CH₂Cl₂ / CH₃OH(NH₃) 90:10). The eluent of the desired fraction was evaporated and the residue was crystallized from a mixture of acetonitrile and 2,2'-oxybispropane. The product was filtered off and dried,

yielding 0.4 g (9.2%) of 6,11-dihydro-3-methyl-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine-2-methanol; mp. 166.8°C (comp. 4.21).

Example 37

A mixture of 1.6 g of (2-pyridinyl)ethene, 2.7 g of compound (5.01) and 100 ml of 1-butanol was stirred for 18 hours at reflux temperature. The reaction mixture was evaporated and the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH / CH₃OH:NH₃ 90:10:1). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 1.7 g (45.6%) of 6,11-dihydro-11-[1-[2-(2-pyridinyl)ethyl]-4-piperidinyl]-5H-imidazo-[2,1-b][3]benzazepine; mp. 170.3°C (comp. 5.04).

Example 38

Through a stirred mixture of 32 g of compound (1.04) and 300 ml of methanol was bubbled gaseous oxirane for 1 hour at room temperature. After stirring for 3 hours at room temperature, the mixture was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH / CH₃OH:NH₃ 90:10:0 → 90:10:5). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate salt in acetonitrile. The salt was filtered off and dried, yielding 15 g (23.1%) of 4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidineethanol (Z)-2-butenedioate(1:2); mp. 145.7°C (comp. 1.30).

Example 39

A solution of 9.6 g of compound (4.08) in 300 ml of methanol/NH₃ was hydrogenated in the presence of 3 g of Raney Nickel catalyst. After complete reaction, the catalyst was filtered off and the filtrate was evaporated, yielding 12.5 g (100%) of 4-(5,6-dihydro-3-methyl-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidineethanamine (comp. 4.09).

30 <u>Example 40</u>

0.57 g of lithium aluminum hydride were added portionwise to 100 ml of tetrahydrofuran under nitrogen. A solution of 2.3 g of compound (1.26) in tetrahydrofuran was added dropwise and the reaction mixture was stirred for 3 hours at reflux temperature. The mixture was decomposed with 2 ml of water, 2 ml of a sodium hydroxide solution 15%. After filtration over diatomaceous earth, the filtrate was evaporated, yielding 2.3 g (97.5%) of 4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-yl)-1-piperidineethanamine (comp. 5.07).

Example 41

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A solution of 3.1 g of compound (1.30) in N,N-dimethylacetamide was added dropwise to a mixture of 0.7 g of a sodium hydride dispersion 50% and 200 ml of N,N-dimethylacetamide under nitrogen and at room temperature. After stirring for 1 hour, 1.1 g of 2-chloropyrimidine were added and the whole was stirred for 16 hours at room temperature. The reaction mixture was decomposed with water and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 90:10). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate salt in 2-propanone. The salt was filtered off and dried, yielding 1.4 g (22.6%) of 6,11-dihydro-11-[1-[2-(2-pyrimidinyloxy)ethyl]-4-piperidinylidene]-5H-imidazo[2,1-b][3]benzazepine (Z)-2-butenedioate(1:2); mp. 172.6°C (comp. 1.31).

15 <u>Example 42</u>

A mixture of 3.3 g of 2-chloropyrimidine, 3.2 g of compound (4.09), 1.26 g of sodium hydrogen carbonate and 200 ml of ethanol was stirred for 18 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. The product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH_2Cl_2 / CH_3OH 95:5 \rightarrow 90:10). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile. The product was filtered off and dried, yielding 2.56 g (63.9%) of N-[2-[4-(5,6-dihydro-3-methyl-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]-2-pyrimidinamine; mp. 171.3°C (comp. 4.10).

Example 43

A mixture of 2.0 g of 5-bromo-1,3,4-thiadiazole-2-amine, 3.42 g of compound (1.27), 1.2 g of sodium carbonate, 0.01 g of potassium iodide and 200 ml of N,N-dimethylacetamide was stirred for 4 hours at 120°C. The reaction mixture was evaporated and the residue was stirred in dichloromethane. The organic layer was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH_2Cl_2 / CH_3OH / CH_3OH :NH₃ 90:10:1 \rightarrow 90:7:3). The eluent of the desired fraction was evaporated and the residue was crystallized from a mixture of acetonitrile and ethanol. The product was filtered off and dried, yielding 1.62 g (36.2%) of N^2 -[2-[4-(5,6-dihydro-11 N^2 -midazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]-1,3,4-thiadiazole-2,5-diamine; mp. 251.4°C (comp. 1.33).

Example 44

To a stirred mixture of 1.1 g of 3-furancarboxylic acid, 1.9 g of N,N-diethylethanamine and 200 ml of dichloromethane were added portionwise 2.4 g of 2-chloro-1-methyl-pyridinium iodide. After stirring for 1 hour at room temperature, a solution of 2.9 parts of compound (1.27) in dichloromethane was added dropwise. Upon completion, the reaction mixture was stirred for 18 hours at room temperature. The reaction mixture was basified with K₂CO₃(aq.) and extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 94:6 → 90:10). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate salt in 2-propanone. The salt was filtered off and dried, yielding 1.88 g (31.5%) of N-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]-ethyl]-3-furancarboxamide (Z)-2-butenedioate(1:2); mp. 202.9°C (comp. 1.35).

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A mixture of 0.6 g of isocyanatomethane, 3.1 g of compound (1.27) and 100 ml of tetrahydrofuran was stirred for 18 hours at room temperature. The reaction mixture was evaporated and the residue was crystallized from acetonitrile. The precipitated product was filtered off and dried, yielding 2.0 g (54.7%) of N-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]-ethyl]-N'-methylurea; mp. 178.1°C (comp. 1.36).

Example 46

Example 45

- a) To a stirred and cooled (-10°C) mixture of 18 g of carbon disulfide, 7.22 g of N,N'-methanetetraylbis[cyclohexanamine] and 150 ml of tetrahydrofuran was added dropwise a solution of 10.8 g of compound (1.27) in tetrahydrofuran. After stirring for 1 hour at room temperature, the reaction mixture was evaporated, yielding 12 g (97.5%) of 6,11-dihydro-11-[1-(2-isothiocyanatoethyl)-4-piperidinylidene]-5H-imidazo[2,1-b] [3]benzazepine (comp. 1.37).
 - b) A mixture of 2.7 g of 3,4-pyridinediamine, 8.8 g of compound (1.37) and 150 ml of tetrahydrofuran was stirred for 18 hours at reflux temperature, yielding 11.5 g (100%) of \underline{N} -(4-amino-3-pyridinyl)- \underline{N} '-[2-[4-(5,6-dihydro-11 \underline{H} -imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]thiourea (comp. 1.38).
- 35 c) A mixture of 11.5 g of compound (1.38), 7.6 g of mercury(II)oxide, 0.01 g of sulfur and 150 ml of tetrahydrofuran was refluxed for 5 hours. The reaction mixture was filtered while hot over diatomaceous earth and the filtrate was evaporated. The residue

was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH /CH₃OH:NH₃ 90:5:5). The eluent of the desired fraction was evaporated and the residue was converted into the (E)-2-butenedioate salt in methanol. The salt was filtered off and dried, yielding 1.65 g (14.4%) of N-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b] [3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]-1H-imidazo[4,5-c]pyridin-2-amine (E)-2-butenedioate(1:3) hemihydrate; mp. 203.0°C (comp. 1.39).

Example 47

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1 g of gaseous methanamine was bubbled through 100 ml of tetrahydrofuran. A
10 solution of 3.5 g of compound (1.37) in tetrahydrofuran was added and the reaction mixture was stirred for 2 hours at room temperature. After evaporation, the residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH / CH₃OH:NH₃ 90:10:0 → 90:10:1). The eluent of the desired fraction was evaporated and the residue was crystallized from 2,2'-oxybispropane. The crystallized product was filtered off and dried, yielding 0.9 g (23.0%) of N-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b][3]benza-zepin-11-ylidene)-1-piperidinyl]ethyl]-N'-methylthiourea hemihydrate; mp. 155.2°C (comp. 1.40).

Example 48

- a) A mixture of 7.6 g of compound (1.30) and 100 ml of thionyl chloride was stirred for 2 hours at reflux temperature. The reaction mixture was evaporated and the residue was stirred in methylbenzene (2x). The obtained residue was dissolved in water and treated with sodium carbonate. The product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the (Z)-2-butenedioate salt in 2-propanone. The salt was filtered off and dried, yielding 0.7 g (5%) of 11-[1-(2-chloroethyl)-4-piperidinylidene]-6,11-dihydro-5H-imidazo[2,1-b][3]-benzazepine (Z)-2-butenedioate(1:2); mp. 169.9°C (comp. 1.41).
- b) A mixture of 2.8 g of 1-methyl-1H-imidazol-2-thiol, 6.5 g of compound (1.41),
 8.3 g of potassium carbonate and 200 ml of 2-propanone was stirred for 18 hours at reflux temperature. The reaction mixture was evaporated, the residue was taken up in dichloromethane, washed with water, dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 90:10). The eluent of the desired fraction was evaporated and the residue was taken up in methylbenzene and treated with activated charcoal. The whole was filtered while hot, the filtrate was allowed to cool and was then evaporated. The residue was converted into the cyclo-

hexanesulfamate salt in 2-propanone and ethanol. The salt was filtered off and dried, yielding 1.6 g (10.5%) of 6,11-dihydro-11-[1-[2-[(1-methyl-1<u>H</u>-imidazol-2-yl)thio]-ethyl]-4-piperidinylidene]-5<u>H</u>-imidazo[2,1-b][3]benzazepine cyclohexanesulfamate (1:2); mp. 265.4°C (decomp.) (comp. 1.42).

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Example 49

- a) A mixture of 9.6 g of methyl \underline{N} -(2,2'-dimethoxyethyl)- \underline{N} '-methylcarbamimidothioate hydroiodide, 9.3 g of compound (1.27) and 200 ml of 2-propanol was stirred for 18 hours at reflux temperature. The reaction mixture was evaporated, yielding 17.4 g (100%) of \underline{N} -[2-[4-(5,6-dihydro-11 \underline{H} -imidazo[2,1-b]-[3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]- \underline{N} '-(2,2-dimethoxyethyl)- \underline{N} ''-methylguanidine monohydroiodide (comp. 1.43).
- b) A mixture of 9.3 g of compound (1.43) and 200 ml of a hydrochloric acid solution was stirred for 18 hours at room temperature. The whole was treated with potassium
 15 carbonate and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography HPLC (silica gel; CHCl₃ / CH₃OH 98:2). The eluent of the desired fraction was evaporated and the residue was converted into the cyclohexanesulfamate salt in 2-propanone and ethanol. The salt was filtered off and dried, yielding 0.71 g (3%) of 4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-N-(1-methyl-1H-imidazol-2-yl)-1-piperidine-ethanamine cyclohexanesulfamate (1:3) dihydrate; mp. 153.9°C (comp. 1.44).

Example 50

A mixture of 1.42 g of 2-mercapto-4-pyrimidinone, 3.1 g of compound (1.27) and 1 ml of N,N-dimethylacetamide was stirred for 4 hours at 140°C. After cooling, the mixture was purified by column chromatography (silica gel; CHCl₃ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was converted into the hydrochloride salt in 2-propanone. The salt was filtered off and dried in vacuo, yielding 1.8 g (32.9%) of 2-[[2-[4-(5,6-dihydro-11H-imidazo-[2,1-b][3]-benzazepin-11-ylidene)-1-piperidinyl]ethyl]amino]-4(1H)-pyrimidinone trihydrochloride dihydrate; mp. 234.8°C (comp. 1.45).

Example 51

A mixture of 1 g of compound (4.11), 5 g of manganese(IV)oxide and 100 ml of trichloromethane was stirred for 2 hours at reflux temperature. The reaction mixture was filtered while hot over diatomaceous earth. After evaporation, the residue was purified

by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from 1,1'-oxybisethane. The product was filtered off and dried, yielding 0.48 g (48.6%) of ethyl 4-(3-formyl-5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidine-carboxylate; mp. 138.2°C (comp. 4.15).

Example 52

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To a stirred solution of 9.7 g of compound (4.15) in 100 ml of water was added dropwise a solution of 13.7 g of AgNO₃ in 50 ml of water and then a solution of 13.3 g of potassium hydroxide in 50 ml of water. After stirring for 18 hours, the reaction mixture was filtered and the filtrate acidified with hydrochloric acid. After evaporation, the residue was stirred in methanol, the precipitate was filtered off and the filtrate was evaporated. The residue was purified by column chromatography (silica gel; NH₄OAc / H₂O / CH₃OH 55:0.5:45). The eluent of the desired fraction was evaporated and the residue was stirred in 2-propanone and activated charcoal. The precipitate was filtered off and the filtrate was evaporated. The residue was crystallized first from 2,2'-oxybis-propane and then from acetonitrile. The product was filtered off and dried, yielding 0.3 g (3%) of 11-[1-(ethoxycarbonyl)-4-piperidinylidene]-6,11-dihydro-5H-imidazo[2,1-b]-[3]benzazepine-3-carboxylic acid; mp. 182.2°C (comp. 4.17).

Example 53

To a stirred mixture of 2.93 g of compound (4.03), 1.3 g of sodium acetate and 30 ml of acetic acid was added dropwise a solution of 1.6 g of bromine in 20 ml of acetic acid. After stirring for 1 hour at 30°C, the mixture was evaporated and the residue was taken up in water. The aqueous solution was treated with sodium hydroxide and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5 → CH₂Cl₂ / CH₃OH / CH₃OH:NH₃ 90:8:2). The eluent of the desired fraction was evaporated and the residue was boiled in acetonitrile. After cooling, the precipitated product was filtered off and dried, yielding 0.96 g (25.8%) of 2-bromo-6,11-dihydro-3-methyl-11-(1-methyl-4-piperidinylidene)-5H-imidazo-[2,1-b][3]benzazepine; mp. 116.0°C (comp. 4.22).

Example 54

a) A mixture of 6.1 g of 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5H-imidazo-[2,1-b][3]benzazepine-3-carboxaldehyde and 5.3 g of monoethyl ester propanedioic acid in 1 ml of piperidine and 50 ml of pyridine was stirred and refluxed for 4 hours. The

reaction mixture was evaporated and the residue was taken up in water. The product was extracted with dichloromethane, dried, filtered and evaporated, yielding 13 g (100%) of ethyl 3-[5,6-dihydro-11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepin-3-yl]-2-propenoate (comp. 4.27).

b) A solution of 1.12 g of potassium hydroxide in 40 ml of water was added dropwise to a stirred mixture of 13 g of compound (4.27) in 20 ml of tetrahydrofuran. The mixture was stirred overnight, acidified with HCl and evaporated. The residue was purified by HPLC Lichroprep 18 25μm (eluent : NH₄OAc/H₂O/CH₃CN 0.5/89.5/10 H₂O/CH₃CN 90/10). The eluent of the desired fraction was evaporated and the residue was stirred in 500 ml of 2-propanone, decolourized with activated charcoal and filtered over diatomaceous earth. The filtrate was evaporated and the residue was stirred in 1,1'-oxybisethane. The product was filtered off and dried, yielding 0.9 g (11.9%) of ethyl (E)-3-[5,6-dihydro-11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b]-[3]benzazepin-3-yl]-2-propenoic acid sesquihydrate; mp. 207.3°C (comp. 4.28).

Example 55

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- a) A mixture of 2.64 g of 2,5-dimethoxytetrahydrofuran, 3.1 g of compound (1.27), 30 ml of water and 10 ml of acetic acid was stirred for 1.5 hours at 50°C. The mixture was basified with NaOH(aq.) and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 95:5). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile, yielding 1.17 g (33%) of 6,11-dihydro-11-[1-[2-(1H-pyrrol-1-yl)ethyl]-4-piperidinylidene]-5H-imidazo-[2,1-b][3]benzazepine; mp. 165.5°C (comp. 1.55).
- b) To 60 ml of N,N-dimethylformamide were added dropwise 5.9 g of phosphoryl chloride. After stirring for 30 minutes at room temperature, there was added a solution of 13.7 g of compound (1.55) in N,N-dimethylformamide and stirring at room temperature was continued for 1 hour. The reaction mixture was poured into a mixture of ice, water and potassium carbonate and the product was extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 96:4). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile, yielding 6.4 g (43%) of 1-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]ethyl]-1H-pyrrole-2-carboxaldehyde; mp. 158.5°C (comp. 1.56).
- c) To a cooled mixture (ice-bath) of 4.4 g of compound (1.56) and 100 ml of methanol was added portionwise over 15 minutes 1.1 g of sodium tetrahydroborate. After stirring for 1 hour at room temperature, the reaction mixture was evaporated and the residue was

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taken up in water. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 97:3 to 93:7). The eluent of the desired fraction was evaporated and the residue was crystallized from acetonitrile, yielding 2.74 g (62%) of 1-[2-[4-(5,6-dihydro-11<u>H</u>-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]-ethyl]-1<u>H</u>-pyrrole-2-methanol; mp. 147.4°C (comp. 1.57).

Example 56

a) A mixture of 4.3 g of compound (1.27), 5.2 g of ethyl 2,5-diethoxy-tetrahydofuran-2-carboxylate and 100 ml of acetic acid was stirred for 2 hours at 80°C. The mixture 10 was evaporated and the residue was taken up in water. The whole was basified with potassium carbonate and the product extracted with dichloromethane. The extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; $CH_2Cl_2 / CH_3OH 96:4 \rightarrow 90:10$). The eluent of the desired fraction was 15 evaporated and the residue was crystallized from acetontrile, yielding 4.3 g (70%) of ethyl 1-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1piperidinyl]ethyl]-1H-pyrrole-2-carboxylate; mp. 158.5°C (comp. 1.58). b) A mixture of 3.2 g of compound (1.58), 40 ml of sodium hydroxide (1N), 50 ml of tetrahydrofuran and 100 ml of water was stirred for 48 hours at reflux temperature. The 20 reaction mixture was evaporated and the residue was washed with dichloromethane. The whole was neutralized with HCl (1N) and the product was extracted with dichloromethane. The organic layer was dried, filtered and evaporated. The product was crystallized successively from 2-propanone and acetonitrile, yielding 1.06 g (36%) of 1-[2-[4-(5,6-dihydro-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]-25 ethyl]-1H-pyrrole-2-carboxylic acid hemihydrate; mp. 166.2°C (comp. 1.59).

Example 57

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To a mixture of 3 g of compound (3.23) and 10 ml of tetrahydrofuran was added dropwise a solution of 0.45 g of potassium hydroxide in 20 ml of water. After stirring overnight at room temperature, the reaction mixture was evaporated and the aqueous layer was washed three times with dichloromethane. The aqueous layer was discoloured with activated charcoal, filtered over diatomaceous earth and concentrated. The aqueous layer was neutralized with HCl till pH=7. The precipitate was filtered off, washed with water and dried, yielding 1.26 g (40%) of 4-(8-fluoro-5,6-dihydro-11H-imidazo[2,1-b]-[3]benzazepin-11-ylidene)-1-piperidinepropanoic acid dihydrate; mp. 136.1°C (comp. 3.31).

Example 58

A mixture of 1.9 g of compound (3.28) and 50 ml of hydrobromic acid 48% (aq.) was stirred for 2 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water and basified with potassium carbonate. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was purified by column chromatography (silica gel; CH₂Cl₂ / CH₃OH 94:6 → 90:10). The eluent of the desired fraction was evaporated and the residue was converted into the (E)-2-butenedioate salt (2:3) in 2-propanol; yielding 1.15 g (42%) of 4-[2-[4-(5,6-dihydro-8-methyl-11H-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidinyl]-ethyl]phenol hemiethanolate hemihydrate (E)-2-butenedioate (2:3); mp. 176.0°C (comp. 3.30).

Example 59

- a) A mixture of 4.3 g of compound (4.16), 9 g of methyl (methylthio)methanesulfoxide 97%, 50 ml of tetrahydrofuran and 20 ml of a solution of benzyltrimethylammonium hydroxide in methanol 40% was stirred for 18 hours at reflux temperature. The reaction mixture was evaporated and the residue was taken up in water. The product was extracted with dichloromethane and the extract was dried, filtered and evaporated. The residue was co-evaporated with methylbenzene (2x) and then taken up in 50 ml of methanol. This solution was cooled on ice and gasueous hydrochloride was bubbled through for 1/2 hour. After stirring overnight, the whole was evaporated. The residue was taken up in water and basified with potassium carbonate. The product was extracted with dichloromethane and further purified by column chromatography (silica gel; CH₂Cl₂/C₂H₅OH(NH₃) 97:3). The desired fraction was evaporated, yielding 3.15 g
- 25 CH₂Cl₂ / C₂H₅OH(NH₃) 97:3). The desired fraction was evaporated, yielding 3.15 g (29.9%) of methyl [5,6-dihydro-11-(1-methyl-4-piperidinylidene)-11<u>H</u>-imidazo-[2,1-b][3]benzazepin-3-yl]acetate (comp. 4.30).
- b) To a stirred mixture of 3.15 g of compound (4.30) and 10 ml of tetrahydrofuran there was added dropwise a solution of 0.56 g of potassium hydroxide in 20 ml of water. Stirring was continued overnight. The organic solvent was evaporated and the remaining aqueous layer was successively washed with dichloromethane (3x) and stirred with activated charcoal. After filtration, the whole was concentrated and then neutralized to pH 7. The product was filtered off and purified by column chromatography (RP 18; CH₃COONH₄ (0.5% in H₂O) / CH₃CN 90:10). The eluent of the desired fraction was evaporated and the residue was recrystallized from acetonitrile, yielding 1.39 g (45.9%)
- evaporated and the residue was recrystallized from acetonitrile, yielding 1.39 g (45.9%) of [5,6-dihydro-11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepin-3-yl]acetic acid (comp. 4.31).

All compounds listed in Tables 1-7 were prepared following methods of preparation described in examples 13-59, as is indicated in the column Ex. No.

Table 1

Co. No.	Ex. No.	L-	Physical data
1.01	13	CH ₃ -	mp. 209.3°C / CF ₃ SO ₃ H
1.02	13	CH ₃ -	mp. 154.5°C
1.03	19	H ₅ C ₂ OOC-	mp. 170.6°C
1.04	21	H-	mp. 192.5°C / 1/2 H ₂ O . 2(E)*
1.05	34	C ₂ H ₅ -	mp. 184.2°C / 2(Z)*
1.06	35	CH ₃ CH(CH ₃)-	mp. 183.6°C / 2(Z)*
1.07	32	CH ₂ =CH-CH ₂ -	mp. 160.8°C / 2(Z)*
1.08	28	$CH_2=C(CH_3)-CH_2-$	mp. 179.5°C / 3/2(E)*
1.09	29	CH_3 - $C(CH_3)$ = CH - CH_2 -	mp. 127.2°C
1.10	25	C ₆ H ₅ -CH=CH-CH ₂ -	mp. 172.2°C / (E)
1.11	33	C ₆ H ₅ -CH ₂ -	mp. 207.2°C
1.12	26	CH ₃ O-(CH ₂) ₂	mp. 180.5°C / 2(COOH) ₂
1.13	26	$S \longrightarrow N \longrightarrow CH_3$ $CH_2)_2$	mp. 181.8°C
1.14	25	CH ₃ (CH ₂) ₂ —	mp. 197.8°C/H ₂ O . 3(E)*
1.15	37	$\langle CH_2 \rangle_2$	mp. 163.8°C
1.16	28	H N $-CH_2 N$ $-CH_3$	mp. 199.0°C

Co. No.	Ex. No.	L-	Physical data
1.17	25	CH ₃ N CH ₃ (CH ₂) ₂ —	mp. 257.4°C
1.18	34	$\langle V_{O} \rangle$ CH ₂ —	mp. 160.3°C
1.19	26	$CH_3 \longrightarrow CH_2 -$	mp. 162.1°C / H ₂ O . 2(E)*
1.20	25	$O = (CH_2)_2 - O$	mp. 188.8°C / 3/2(E)*
1.21	25	H_5C_2-N $N-(CH_2)_2-N$	mp. 170.7°C / 2(Z)*
1.22	25	$\begin{array}{c} O \\ HN \\ \longrightarrow \\ $	mp. 194.7°C
1.23	25	C ₂ H ₅ -O-(CH ₂) ₂ -	mp. 176.5°C / 2(Z)*
1.24 1.25 1.26	25 25 27	CH_3 O CH_3 — HC — NH – C — $(CH_2)_2$ — H_5C_2OOC - NH - $(CH_2)_2$ - NC - CH_2 -	mp. 165.5°C mp. 167.2°C / 2(E)* mp. 220.4°C
1.27	21	$H_2N-(CH_2)_2-$	-
1.28	39	H ₂ N-(CH ₂) ₂ -	mp. 186.6°C / 1/2 H ₂ O . 3(E)*
1.29	38	HO-(CH ₂) ₂ -	mp. 225.1°C / CF ₃ SO ₃ H
1.30	38	HO-(CH ₂) ₂ -	mp. 145.7°C / 2(Z)*
1.31	41	$\langle \text{CH}_2 \rangle_2 - \langle \text{CH}_2 \rangle_$	mp. 172.6°C / 2(Z)*
1.32	42	$\langle N \rangle$ NH-(CH ₂) ₂ —	mp. 165.1°C
1.33	43	$N-N$ H_2N S $NH-(CH_2)_2$	mp. 251.4°C
1.34	43	N NH-(CH ₂) ₂ -	mp. 205.5°C / 1/2H ₂ O / 4**
1.35	44	$ \begin{array}{c} O \\ O \\ C \\ -NH-(CH_2)_2- \end{array} $	mp. 202.9°C / 2(Z)*

Co. No.	Ex. No.	L	Physical data
1.36	45	O CH ₃ -NH-C-NH-(CH ₂) ₂ -	mp. 178.1°C
1.37	46a	SCN-(CH ₂) ₂ -	mp. 176.1 C
1.38	46b	$NH_{2} S S S S S S S S S S S S S S S S S S S$	-
1.39	46c	$ \begin{array}{c c} H \\ N \\ N$	mp. 203.0°C / 1/2H ₂ O . 3(E)*
1.40	47	$ \begin{array}{c} S\\CH_3-NH-C-NH-(CH_2)_2-\end{array} $	mp. 155.2°C / 1/2H ₂ O
1.41	48	Cl-(CH ₂) ₂ -	mp. 169.9°C / 2(Z)*
1.42	48	CH_3 $N \rightarrow S - (CH_2)_2 - N$	mp. 265.4°C (dec.) / 2**
1.43	49a	OCH_3 $N-CH_3$ $CH_3O-CH-CH_2-NH-C-NH-(CH_2)_2-$	ні
1.44	49b	CH_3 N	mp. 153.9°C / 2H ₂ O . 3**
1.45	50	$ \begin{array}{c} H \\ N \\ N \\ N \\ N \end{array} $ $ \begin{array}{c} N \\ N \\ N \\ N \end{array} $	mp. 234.8°C / 2H ₂ O . 3HCl
1.46	26	F—C—(CH ₂) ₃ —	mp. 161.0°C
1.47	38	OH 	2-(E)* / mp. 156.4°C
1.48	28	H ₅ C ₂ -O-CO-(CH ₂) ₂ -	-
1.49	27	CH2-	mp. 131.5°C
		COCII ₃	
1.50	27	H ₃ CO—(CH ₂) ₂ —	(E)* . 1/2 H ₂ O . 1/2 ethanolate / mp. 127.4°C
1.51	25	F—(CH ₂) ₃ -	mp. 130.3°C

Co. No.	Ex. No.	L-	Physical data
1.52	25	CH ₃ CH ₃ CH ₃ (CH ₂) ₂ -	mp. 195.9°C
1.53	25	CH ₃ — (CH ₂) ₂ —	mp. 202.9°C
1.54	24	CH ₃ -CO	mp. 182.1°C
1.55	55a	N-(CH ₂) ₂ -	mp. 165.5°C
1.56	55b	N-(CH ₂) ₂ -	mp. 158.5°C
1.57	55c	$N-(CH_2)_2 CH_2-OH$	mp. 147.4°C
1.58	56a	$ \begin{array}{c} N - (CII_2)_2 - \\ C - O - C_2II_5 \\ O \end{array} $	mp. 158.5°C
1.59	56b	N-(CH ₂) ₂ -	1/2 H ₂ O / mp. 166.2°C
1.60	57	HOOC-(CH ₂) ₂ -	2 H ₂ O / mp. 154.9°C
1.61	57	C - OH	ethanolate(1:1) / mp. 208.6°C

*: 2-butenedioate

** : cyclohexanesulfamate

Co. No.	Ex. No.	L-	R ¹	Physical data
2.01	13	CH ₃ -	F	mp. 195.7°C / 2(E)*
2.02	19	H ₅ C ₂ OOC-	F	mp. 175.2°C
2.03	21	Н-	F	mp. 180.1°C
2.04	31	CH_3 - $C(CH_3)$ = CH - CH_2 -	F	mp. 203.4°C / 2(Z)*
2.05	25	$ \begin{array}{c c} S & N & CH_3 \\ \hline N & (CH_2)_2 - \\ O & O \end{array} $	F	mp. 168.9°C / 3/2 H ₂ O . 5/2(E)*
2.06	25	$S \longrightarrow N \longrightarrow CH_3$ $CH_2)_2$	F	mp. 162.2°C / 3/2 H ₂ O . 5/2(E)*
2.07	25	O (CH ₂) ₂ —	F	mp. 201.9°C / 3(E)*
2.08	30	NC-CH ₂ -	F	mp. 198.3°C
2.09	39	H ₂ N-(CH ₂) ₂ -	F	-
2.10	42	N-NH-(CH ₂) ₂	F	mp. 165.1°C / 3(Z)*
2.11	43	H_2N $N-N$ $NH-(CH_2)_2-$	F	mp. 238.6°C
2.12	13	Н-	CH ₃ -	mp. 203.1°C
2.13	24	H ₅ C ₂ OOC-	CH ₃ -	mp. 156.5°C
2.14	33	CH ₃ -	CH ₃ -	mp. 214.3°C
2.15	26	$ \begin{array}{c c} S & N & CH_3 \\ \hline \downarrow & N & CH_2)_2 - \end{array} $	CH ₃ -	mp. 202.2°C
2.16	30	NC-CH ₂ -	CH ₃ -	-
2.17	39	H ₂ N-(CH ₂) ₂ -	CH ₃ -	mp. 219.3°C / 3(E)*
2.18	42	N-NH-(CH ₂) ₂ -	CH ₃ -	mp. 131.1°C
2.19	26	CH_3O \longrightarrow $(CH_2)_2$	CH ₃ -	mp. 192.6°C / 5/2(COOH) ₂
2.20	44	$ \begin{array}{c c} O & O \\ \hline C-NH-(CH_2)_2- \end{array} $	CH ₃ -	mp. 214.2°C / 2(Z)*

Co. No.	Ex. No.	L-	R ¹	Physical data
2.21	13	CH ₃ -	Br	mp. 213.4°C
2.22	25	CH_3 N CH_3 CH_3 CH_3 CH_3 CH_2	F	mp. 187.2°C / H ₂ O

*: 2-butenedioate

Co. No.	Ex. No.	L-	R ²	Physical data
3.01	14	CH ₃ -	CH ₃ O-	mp. 190.3°C / 2(Z)*
3.02	19	H ₅ C ₂ OOC-	CH ₃ O-	mp. 104.4°C
3.03	21	H-	CH ₃ O-	mp. 184.4°C
3.04	13	H-	CH ₃ -	mp. 221.9°C / 2(E)*
3.05	33	CH ₃ -	CH ₃ -	mp. 211.0°C / 2(E)*
3.06	25	S N CH ₃ (CH ₂) ₂ —	СН ₃ -	mp. 199.8°C
3.07	25	CH_3 CH_3 $CH_2)_2$	CH ₃ -	mp. 214.2°C
3.08	25	N CH ₃ (CH ₂) ₂ —	СН ₃ -	mp. 162.3°C / H ₂ O . 3(E)*
3.09	25	S N CH ₃ (CH ₂) ₂ —	СН ₃ -	mp. 235.1°C / 2H ₂ O . 3HCl
3.10	13	CH ₃ -	Cl	mp. 186.6°C
3.11	19	H ₅ C ₂ OOC-	Cl	mp. 140.3°C

Co. No.	Ex. No.	L-	Ŗ ²	Physical data
3.12	23	H- .S. N. CH ₃	Cl	mp. 197.1°C
3.13	26	(CH ₂) ₂ —	Cl	mp. 217.6°C
3.14	30	NC-CH ₂ -	Cl	
3.15	15	CH ₃ -	F	mp. 152.4°C
3.16	19	H ₅ C ₂ OOC-	F	mp. 149.4°C
3.17	21	H-	F	-
3.18	26	$ \begin{array}{c c} S & N & CH_3 \\ \hline \downarrow & N & (CH_2)_2 - \\ O & & \end{array} $	F	mp. 192.2°C / H ₂ O . 3/2(E)*
3.19	29	H ₃ CO—(CH ₂) ₂ —	ОСН3	3/2(E)* . ethanolate / mp. 150.3°C
3.20	32	NII-(CH ₂) ₂ -	Cl	ethanedioate(1:2) / mp. 206.7°C
3.21	37	(CII ₂) ₂ —	Cl	mp. 171.3°C
3.22	39	H ₂ N-(CH ₂) ₂ -	Cl	-
3.23	28	H ₅ C ₂ -O-CO-(CH ₂) ₂ -	F	mp. 114.6°C
3.24	27	NC-CH ₂ -	F	mp. 204.7°C
3.25	27	CH ₃ (CH ₂) ₂ —	F	mp. 211.6°C
3.26	27	II ₃ CO — (CII ₂) ₂ —	F	mp. 149.1°C
3.27	25	O CH ₃	Cl	ethanedioate(2:5), 1/2 ethanolate / mp. 170.7°C

Co. No.	Ex. No.	L-	R ²	Physical data
3.28	25	H ₃ CO — (CH ₂) ₂ —	СН3	cyclohexylsulfamate(1:2), H ₂ O / mp. 149.8°C
3.29	25	HN N-(CH ₂) ₃ -	CH ₃	(E)-2-butenedioate(1:2), 1/2 H ₂ O / mp. 200.3°C
3.30	58	HO—(CH ₂) ₂ —	CH ₃	(E)-2-butenedioate(2:3) . 1/2 ethanolate . 1/2 H ₂ O / mp. 176.0°C
3.31	57	HOOC-(CH ₂) ₂ -	F	2 H ₂ O / mp. 136.1°C
3.32	42	$NH - (CH_2)_2 - $	F	mp. 191.2°C
3.33	44	O	F	mp. 173.5°C
3.34	37	(CH ₂) ₂ —	F	mp. 177.2°C
3.35	58	HO—(CH ₂) ₂ —	F	-
3.36	39	H ₂ N-(CH ₂) ₂ -	F	mp. 141.5°C

Co. No.	Ex. No.	L-	R ²	R ³	R ⁴	Physical data
4.01	17	CH ₃ -	Н	Н	C ₆ H ₅	mp. 171.5 °C
4.02	13	H	Н	-CH ₃	Н	mp. 167.0 °C
4.03	33	CH ₃ -	Н	-CH ₃	Н	mp. 172.2 °C
4.04	25	$S \longrightarrow N \longrightarrow CH_3$ $CH_2)_2$	н	-CH ₃	Н	mp. 212.4 °C
4.05	25	N CH ₃ (CH ₂) ₂ —	Н	-CH ₃	Н	mp. 186.3 °C
4.06	25	CH ₃ O — (CH ₂) ₂ —	Н	-СН ₃	Н	/ 3 (E)* . H ₂ O mp.150.6 °C / 5/2(COOH) ₂ , H ₂ O
4.07	37	$(CH_2)_2$	Н	-CH ₃	н	mp.180.2 °C / 7/2(COOH) ₂
4.08	30	NC-CH ₂ -	Н	-CH ₃	Н	mp. 226.5 °C
4.09	39	$H_2N-(CH_2)_2-$	Н	-CH ₃	Н	-
4.10	42	N-NH-(CH ₂) ₂	Н	-CH ₃	н	mp. 171.3 °C
4.11	20	C ₂ H ₅ OOC-	Н	-CH ₂ OH	Н	mp. 191.9 °C
4.12	21	Н	Н	-CH ₂ OH	Н	mp.>200 °C
						dec./ 5/2 (E)*
4.13	33	CH ₃ - S N CH ₃	н	-CH ₂ OH	Н	mp. 228.3 °C
4.14	26	(CH ₂) ₂ —	Н	-CH ₂ OH	Н	-
4.15	51	O C ₂ H ₅ OOC-	н	-СНО	7.7	120 2 00
4.16	51	CH ₃ -	Н	-CHO -CHO	Н	mp. 138.2 °C
4.17	52	C ₂ H ₅ OOC-	н	-COOH	H H	mp. 171.6 °C
4.18	20	C ₂ H ₅ OOC-	н	-CH ₂ OH	-CH ₂ OH	mp. 182.2 °C
4.19	21	H-	Н	-CH ₂ OH	-CH ₂ OH	_
4.20	33	CH ₃ -	н	-CH ₂ OH	-CH ₂ OH	mp. 206.3 °C
4.21	36	CH ₃ -	н	-CH ₃	-CH ₂ OH	mp. 166.8 °C
4.22	53	CH ₃ -	н	-CH ₃	-Br	mp.116.0°C
				3		

Co. No.	Ex. No.	L-	R ²	R ³ .	R ⁴	Physical data
4.23	52	CH ₃ -	Н	-COOH	Н	mp. 241.3°C
4.24	51	CH ₃ -	F	-СНО	Н	mp. 176.5°C
4.25	36	CH ₃ -	F	-CH ₂ OH	Н	mp. 181.5°C
4.26	36	CH ₃ -	F	-CH ₂ OH	-CH ₂ OH	mp. 220.0°C
4.27	54a.	СН ₃ -	Н	-CH=CH-COOC ₂ H ₅	Н	-
4.28	54b	CH ₃ -	н	-СН=СН-СООН	н	(E) / 3/2H ₂ O mp. 207.3°C
4.29	52	СН ₃ -	F	-СООН	н	1/2 H ₂ O mp. 261.6°C
4.30	59a	CH ₃ -	Н	-CH ₂ -COOCH ₃	Н	-
4.31	59b	CH ₃ -	Н	-СН ₂ -СООН	Н	-

^{* = 2-}butenedioate

Table 5

Co. No.	Ex. No.	L	Physical data
5.01	16	H-	mp. 220.2 °C / 2 (E)*
5.02	34	CH ₃ -	mp. 117.8 °C / 1/2 H ₂ O
5.03	25	S N CH ₃ (CH ₂) ₂ —	mp. 221.6 °C / 2 (COOH) ₂ / 1/2 H ₂ O
5.04	37	(CH ₂) ₂ -	mp. 170.3 °C
5.05	25	N CH ₃ (CH ₂) ₂ —	mp. 193.3 °C
5.06	27	NC-CH ₂ -	mp. 194.7 °C / 1/2 (E)*

Co. No.	Ex. No.	L-	Physical data
5.07	40	H ₂ N-CH ₂ -CH ₂ -	-
5.08	42	$\langle N \rangle$ NH-(CH ₂) ₂ —	mp. 175.1 °C / 7/2 (E)*
5.09	44	$ \begin{array}{c c} \hline O \\ \hline O \\ C-NH-(CH_2)_2- \end{array} $	mp. 203.5 °C
5.10	25	H ₃ CO —(CH ₂) ₂ —	cyclohexylsulfamate(1:2) 1/2 H ₂ O / mp. 125.4°C
5.11	24	CH ₃ -CO-	mp. 153.8°C

^{* = 2-}butenedioate

Table 6

Co.	Ex.	L-	Physical data
No.	No.		
6.01	18	CH ₃ -	mp. 135.8 °C
6.02	19	C ₂ H ₅ OOC-	-
6.03	22	Н-	mp. 246.9 °C / 2HBr 1/2 H ₂ O
6.04	27	$ \begin{array}{c c} S & N & CH_3 \\ \hline N & (CH_2)_2 - \\ O \end{array} $	mp.206.4 °C / 2(COOH) ₂ 1/2 H ₂ O
6.05	26	$ \begin{array}{c c} S & N & CH_3 \\ \hline N & (CH_2)_2 - \\ O & CH_2 \end{array} $	mp.158.9 °C / 5/2(COOH) ₂ .1/2 H ₂ O

$$L-N$$
 R^5

Co. No.	Ex. No.	R ⁵	L-	Physical data
7.01	15	-Cl	CH ₃ -	mp. 181.9 °C
7.02	33	-CH ₃	CH ₃ -	mp. 184.2°C
7.03	42	-СН3	$N = (CH_2)_2 - (CH_2$	ethanedioate (2:7) 1/2 H ₂ O /mp. 171.2°C
7.04	37	-CH ₃	(c.n _D)	(E)-2-butenedioate(2:3) 1/2 H ₂ O /162.2°C
7.05	39	-CH ₃	H ₂ N-(CH ₂) ₂ -	(Z)-2-butenedioate(1:3) / mp. 192.0°C
7.06	13	-CH ₃	Н	-
7.07	13	-CH ₃	Н	2 HCl
7.08	13	-F	CH ₃ -	mp. 164.6°C
7.09	27	-CH ₃	NC-CH ₂ -	mp. 194.1°C
7.10	25	-СН3	$ \begin{array}{c c} S & CH_3 \\ \hline N & (CH_2)_2 - \\ O \end{array} $	mp. 224.3°C

C. Composition Examples

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The following formulations exemplify typical pharmaceutical compositions in dosage unit form suitable for systemic or topical administration to warm-blooded animals in accordance with the present invention.

"Active ingredient" (A.I.) as used throughout these examples relates to a compound of formula (I) or a compound of formula (VII) wherein Q represents (C₁₋₆alkyl or phenyl)oxycarbonyl, C₁₋₄alkylcarbonyl or C₁₋₆alkyl substituted with cyano or amino, a pharmaceutically acceptable acid addition salt or a stereochemically isomeric form thereof.

Example 60: Oral drops

500 g of the A.I. is dissolved in 0.5 l of 2-hydroxypropanoic acid and 1.5 l of the

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polyethylene glycol at 60–80°C. After cooling to 30–40°C there are added 35 l of polyethylene glycol and the mixture is stirred well. Then there is added a solution of 1750 g of sodium saccharin in 2.5 l of purified water and while stirring there are added 2.5 l of cocoa flavor and polyethylene glycol q.s. to a volume of 50 l, providing an oral drop solution comprising 10 mg/ml of the A.I. The resulting solution is filled into suitable containers.

Example 61: Oral solutions

9 g of methyl 4-hydroxybenzoate and 1 g of propyl 4-hydroxybenzoate are dissolved in 4 l of boiling purified water. In 3 l of this solution are dissolved first 10 g of 2,3-dihydroxybutanedioic acid and thereafter 20 g of the A.I. The latter solution is combined with the remaining part of the former solution and 12 l of 1,2,3-propanetriol and 3 l of sorbitol 70% solution are added thereto. 40 g of sodium saccharin are dissolved in 0.5 l of water and 2 ml of raspberry and 2 ml of gooseberry essence are added. The latter solution is combined with the former, water is added q.s. to a volume of 20 l providing an oral solution comprising 5 mg of the A.I. per teaspoonful (5 ml). The resulting solution is filled in suitable containers.

Example 62: Capsules

20 g of the A.I., 6 g sodium lauryl sulfate, 56 g starch, 56 g lactose, 0.8 g colloidal silicon dioxide, and 1.2 g magnesium stearate are vigorously stirred together. The resulting mixture is subsequently filled into 1000 suitable hardened gelatin capsules, each comprising 20 mg of the A.I..

25 Example 63: Film-coated tablets

Preparation of tablet core

A mixture of 100 g of the A.I., 570 g lactose and 200 g starch is mixed well and thereafter humidified with a solution of 5 g sodium dodecyl sulfate and 10 g polyvinyl-pyrrolidone (Kollidon-K 90®) in about 200 ml of water. The wet powder mixture is sieved, dried and sieved again. Then there are added 100 g microcrystalline cellulose (Avicel®) and 15 g hydrogenated vegetable oil (Sterotex ®). The whole is mixed well and compressed into tablets, giving 10.000 tablets, each comprising 10 mg of the active ingredient.

Coating

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To a solution of 10 g methyl cellulose (Methocel 60 HG®) in 75 ml of denaturated ethanol there is added a solution of 5 g of ethyl cellulose (Ethocel 22 cps ®) in 150 ml of dichloromethane. Then there are added 75 ml of dichloromethane and 2.5 ml 1,2,3-propanetriol. 10 g of polyethylene glycol is molten and dissolved in 75 ml of

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dichloromethane. The latter solution is added to the former and then there are added 2.5 g of magnesium octadecanoate, 5 g of polyvinylpyrrolidone and 30 ml of concentrated colour suspension (Opaspray K-1-2109®) and the whole is homogenated. The tablet cores are coated with the thus obtained mixture in a coating apparatus.

Example 64: Injectable solutions

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1.8 g methyl 4-hydroxybenzoate and 0.2 g propyl 4-hydroxybenzoate are dissolved in about 0.5 l of boiling water for injection. After cooling to about 50°C there are added while stirring 4 g lactic acid, 0.05 g propylene glycol and 4 g of the A.I..The solution is cooled to room temperature and supplemented with water for injection q.s. ad 1 l volume, giving a solution of 4 mg A.I. per ml. The solution is sterilized by filtration (U.S.P. XVII p. 811) and filled in sterile containers.

Example 65: Suppositories

3 g A.I. is dissolved in a solution of 3 g 2,3-dihydroxybutanedioic acid in 25 ml polyethylene glycol 400. 12 g surfactant (SPAN®) and triglycerides (Witepsol 555®) q.s. ad 300 g are molten together. The latter mixture is mixed well with the former solution. The thus obtained mixture is poured into moulds at a temperature of 37-38°C to form 100 suppositories each containing 30 mg of the A.I.

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Claims

1. A compound having the formula

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a pharmaceutically acceptable addition salt or a stereochemically isomeric form thereof, wherein each of the dotted lines independently represents an optional bond,

R¹ represents hydrogen, halo, C₁₋₄alkyl, or C₁₋₄alkyloxy;

10 R² represents hydrogen, halo, C₁₋₄alkyl or C₁₋₄alkyloxy;

R³ represents hydrogen, C₁-4alkyl, ethenyl substituted with hydroxycarbonyl or C₁-4alkyloxycarbonyl, C₁-4alkyl substituted with hydroxycarbonyl or C₁-4alkyloxycarbonyl, hydroxyC₁-4alkyl, formyl or hydroxycarbonyl;

R⁴ represents hydrogen, C₁-4alkyl, hydroxyC₁-4alkyl, phenyl or halo;

15 R^5 represents hydrogen, C_{1-4} alkyl or halo;

L represents hydrogen; C₁-6alkyl; C₁-6alkyl substituted with one substituent selected from the group consisting of hydroxy, halo, C₁-4alkyloxy, hydroxycarbonyl, C₁-4alkyloxycarbonyl, C₁-4alkyloxycarbonyl-C₁-4alkyloxy, C₁-4alkyloxycarbonylamino, C₁-4alkylaminocarbonyl,

20 C₁₋₄alkylaminocarbonylamino, C₁₋₄alkylaminothiocarbonylamino, aryl, aryloxy and arylcarbonyl; C₁₋₆alkyl substituted with both hydroxy and aryloxy;

C3-6alkenyl; C3-6alkenyl substituted with aryl;

wherein each aryl is phenyl or phenyl substituted with halo, cyano, hydroxy, C₁₋₄alkyl, C₁₋₄alkyloxy, aminocarbonyl or phenyl substituted with C₁₋₄alkyloxycarbonyl or

25 hydroxycarbonyl; or,

L represents a radical of formula

-Alk-Het³

-Alk-Y-Het¹

(a-1), (a-2) or

-Alk-NH-CO-Het²

(a-3); wherein

30 Alk represents C₁₋₄alkanediyl;

Y represents O, S or NH;

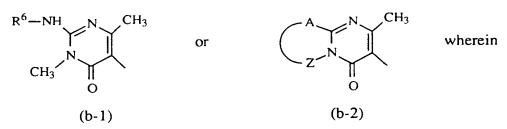
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Het 1 , Het 2 and Het 3 each represent furanyl, thienyl, oxazolyl, thiazolyl or imidazolyl each optionally substituted with one or two C_{1-4} alkyl substituents; pyrrolyl or pyrazolyl optionally substituted with formyl, hydroxy C_{1-4} alkyl, hydroxycarbonyl, C_{1-4} alkyloxycarbonyl or one or two C_{1-4} alkyl substituents; thiadiazolyl or oxadiazolyl optionally substituted with amino or C_{1-4} alkyl; pyridinyl, pyrimidinyl, pyrazinyl or pyridazinyl each optionally substituted with C_{1-4} alkyl, C_{1-4} alkyloxy, amino, hydroxy or halo; imidazo[4,5-c]pyridin-2-yl; and

Het ³ may also represent 4,5-dihydro-5-oxo- $1\underline{H}$ -tetrazolyl substituted with C₁₋₄alkyl, 2-oxo-3-oxazolidinyl, 2,3-dihydro-2-oxo- $1\underline{H}$ -benzimidazol-1-yl or a radical of formula



R6 represents hydrogen or C₁₋₄alkyl; and

A-Z represents -S-CH=CH-, -S-CH₂-CH₂-, -S-CH₂-CH₂-, -CH=CH-CH=CH-, -CH₂-CH₂-CH₂-, -N(CH₃)-C(CH₃)=CH- or -CH=C(CH₃)-O-; provided that 6,11-dihydro-11-(4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine is ecxluded.

- 2. A compound according to claim 1 wherein L is C₁₋₄alkyl or C₁₋₄alkyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl.
- 3. A compound according to claim 1 wherein

R³ represents hydrogen, C₁₋₄alkyl, formyl, hydroxyC₁₋₄alkyl or hydroxycarbonyl; R⁴ represents hydrogen, halo or hydroxyC₁₋₄alkyl; and

- 25 L represents hydrogen, C₁₋₄alkyl, haloC₁₋₄alkyl, hydroxycarbonylC₁₋₄alkyl, C₁₋₄alkyloxycarbonylC₁₋₄alkyl, C₁₋₄alkyloxycarbonylaminoC₁₋₄alkyl, aryl-C₁₋₄alkyl, propenyl, or
- L is a radical of formula (a-1), (a-2) or (a-3), wherein
 Het¹, Het², and Het³ each represent furanyl, oxazolyl or thiazolyl each optionally
 substituted with C₁₋₄alkyl; thiadiazolyl optionally substituted with amino, pyridinyl; or
 pyrimidinyl each optionally substituted with hydroxy; imidazo[4,5-c]pyridin-2-yl; and
 Het³ may also represent a radical of formula (b-2).
 - 4. A compound according to claim 3 wherein

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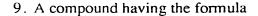
R¹ represents hydrogen or halo;

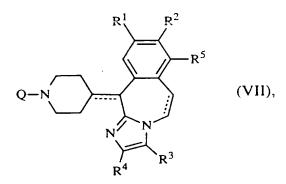
R² represents hydrogen, halo or C₁₋₄alkyloxy; and

- L represents hydrogen, C₁₋₄alkyl, haloC₁₋₄alkyl, hydroxycarbonylC₁₋₄alkyl, C₁₋₄alkyloxycarbonylC₁₋₄alkyl, or a radical of formula (a-1), wherein Y represents NH.
- 5. A compound according to claim 1 wherein said compound is selected from the group consisting of
- 5,6-dihydro-11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepine;
- 9-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]-benzazepine;
 - 11-(1-methyl-4-piperidinylidene)-11H-imidazo[2,1-b][3]benzazepine;
 - 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine-3-methanol;
- 8-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]-benzazepine;
 - 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine-3-carboxaldehyde;
- 6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]benzazepine-3-20 carboxylic acid;
 - 7-fluoro-6,11-dihydro-11-(1-methyl-4-piperidinylidene)-5<u>H</u>-imidazo[2,1-b][3]-benzazepine; and
 - 4-(8-fluoro-5,6-dihydro-11<u>H</u>-imidazo[2,1-b][3]benzazepin-11-ylidene)-1-piperidine-propanoic acid dihydrate.
 - 6. A pharmaceutical composition comprising as an active ingredient a therapeutically effective amount of a compound as defined in any one of claims 1 to 5 and a pharmaceutically acceptable carrier.
- 7. A method of preparing a pharmaceutical composition as claimed in claim 6, characterized in that a therapeutically effective amount of a compound as claimed in any one of claims 1 to 5 is intimately mixed with a pharmaceutical carrier.
 - 8. A compound as claimed in any one of claims 1 to 5 for use as a medicine.

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- an acid addition salt thereof or a stereochemically isomeric form thereof, wherein each of the dotted lines independently represents an optional bond,
 - R¹ represents hydrogen, halo, C₁-4alkyl, or C₁-4alkyloxy;
 - R² represents hydrogen, halo, C₁₋₄alkyl or C₁₋₄alkyloxy;
- R³ represents hydrogen, C₁₋₄alkyl, ethenyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl, C₁₋₄alkyl substituted with hydroxycarbonyl or C₁₋₄alkyloxycarbonyl, hydroxyC₁₋₄alkyl, formyl or hydroxycarbonyl;
 - R⁴ represents hydrogen, C₁₋₄alkyl, hydroxyC₁₋₄alkyl, phenyl or halo;
 - R⁵ represents hydrogen, C₁₋₄alkyl or halo;
- Q represents (C₁₋₆alkyl or phenyl)oxycarbonyl, C₁₋₄alkylcarbonyl or C₁₋₆alkyl substituted with halo, cyano, amino, isothiocyanato, (4-amino-3-pyridinyl)-aminothiocarbonylamino, (CH₃O)₂CH-CH₂-NH-C(=NCH₃)-NH- or methylsulfonyloxy; provided that 1-acetyl-4-(5,6-dihydro-11<u>H</u>-imidazol[1,2-b][3]-benzazepine-11-ylidene)piperidine is excluded.
- 20 10. A process for preparing a compound as defined in any one of claims 1 to 5, characterized by
 - a) cyclizing an alcohol of formula (II) or a ketone of formula (III) in the presence of an acid;

-76-ОН \mathbb{R}^2 R^1 (II) R^2 R^4 **(**T) **(III)** R^4

b) cyclizing an intermediate of formula (IV) wherein W represents a reactive leaving group, thus yielding a compound of formula (I) wherein the central ring of the tricyclic moiety does not contain an optional bond;

c) dehydrating an alcohol of formula (V) or (VI) in the presence of a dehydrating reagent, thus yielding a compound of formula (I) wherein a double bond exists 10 between the piperidinyl and the tricyclic moiety;

d) dehydrating an alcohol of formula (V) wherein the central ring of the tricyclic moiety does not contain an optional bond, in the presence of a dehydrating reagent, thus yielding a compound of formula (I) with a double bond in the tricyclic moiety and a single bond bridging the tricyclic moiety and the piperidine;

e) reacting an intermediate of formula (I-b) wherein —T represents an imidazo[2,1-b][3]benzazepine moiety of formula

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 \mathcal{Q}

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$$\begin{array}{c}
R^1 \\
R^2 \\
R^5 \\
R^4
\end{array}$$

with C₁₋₄alkylchloroformate in the presence of a base and in a reaction-inert solvent yielding a compound of formula (VII-a)

$$C_{1-6}alkyl-N \qquad \qquad C_{1-4}alkyl-O-C-Cl \qquad C_{1-4}alkyl-O-C-N \qquad \cdots T$$

$$(I-b) \qquad \qquad (VII-a)$$

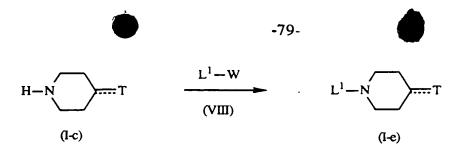
which can be hydrolyzed to a compound of formula (I-c)

in the presence of an acid or a base;

- f) reacting a compound of formula (I-b) with an α-halo-C₁₋₄alkyl chloroformate in a
 reaction-inert solvent yielding a compound of formula (I-c);
 - g) debenzylating a compound of formula (I-d) by catalytic hydrogenation in the presence of hydrogen and a catalyst in a reaction-inert solvent;

$$CH_2-N$$
 HN
 $(I-c)$

h) N-alkylating a compound of formula (I-c) with a reagent of formula (VIII) in a reaction-inert solvent, optionally in the presence of a base;



- i) reductively N-alkylating a compound of formula (I-c) with a reagent of formula L²=O
 (IX) wherein L² represents a geminal bivalent C₁₋₆alkylidene radical which
- 5 optionally may be substituted, in a reaction-inert solvent, in the presence of a base;

$$L^2H-N$$
(I-c)
$$L^2H-N$$
(I-f)

j) reacting a compound of formula (I-c) with a reagent of formula (X) in a reaction-inert
 solvent;

H-N
$$\frac{\operatorname{Het}^3-C_{2-4}\operatorname{alkenyl}-H}{(X)}$$
 $\operatorname{Het}^3-C_{2-4}\operatorname{alkyl}-N$ \cdots T (I-g)

k) reacting a compound of formula (I-c) with an epoxide of formula (XI) wherein R⁷
 represents hydrogen, C₁₋₄alkyl or aryloxyC₁₋₄alkyl in a reaction-inert solvent;

$$R^7$$
 R^7
 CH
 CH_2
 CH_2
 CH_2
 CH_2
 CH_3
 CH_4
 CH_2
 CH_4
 CH_5
 CH_5
 CH_6
 CH_7
 CH_7

reacting a compound of formula (I-k) with a reagent of formula (XII) in a reaction inert solvent in the presence of a base;

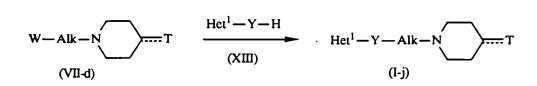
$$H-Y-Alk-N$$
 $(I-k)$
 $Het^1-Y-Alk-N$
 $(I-j)$

m) reacting a compound of formula (VII-d) with a reagent of formula (XIII) in a reaction-inert solvent in the presence of a base;

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n) N-acylating a compound of formula (VII-b) with a carboxylic acid of formula (XIV) in a reaction-inert solvent;

o) reacting a compound of formula (VII-b) with a C₁₋₄alkyliso(thio)cyanate in a reaction-inert solvent;

p) reacting a compound of formula (VII-b) with carbon disulfide in the presence of a dehydrating reagent yielding a compound of formula (VII-e)

$$\begin{array}{c|c} CS_2 & \\ \hline \\ (VII-b) & \\ \end{array} S = C = N - C_{2-4}alkyl - N \\ \hline \end{array}$$

which can be reacted with 3,4-diaminopyridine in a reaction-inert solvent, thus yielding a compound of formula (VII-f)

$$S=C=N-C_{2-4}alkyl-N \longrightarrow T \longrightarrow NH_2 \longrightarrow N$$

which can be cyclized with a metal oxide into a compound of formula (I-n);

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) ,

 q) reacting a compound of formula (VII-e) or the corresponding isocyanate with C₁₋₄alkylamine in a reaction-inert solvent;

$$D = C = N - C_{2.4} alkyl - N + C_{1-4} alkyl - NH_{2}$$

$$C_{1-4} alkyl - NH - C - NH - C_{2.4} alkyl - N$$

$$D \text{ is } S : (I-m-1)$$

$$D \text{ is } O : (I-m-2)$$

r) reacting a compound of formula (VII-b) with a reagent of formula (XV) in a reactioninert solvent yielding a compound of formula (VII-g)

which can be cyclized in an acidic aqueous solution into a compound of formula (I-o);

$$\begin{array}{c} \text{CH}_{3}\text{O} \\ \text{CH}-\text{CH}_{2}-\text{NH}-\text{C}-\text{NH}-\text{C}_{2\text{-4}}\text{alkyl}-\text{N} \\ \text{CH}_{3}\text{O} \\ \text{(VII-g)} \\ \end{array}$$

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s) reacting a compound of formula (I-p) with formaldehyde optionally in the presence of a carboxylic acid-carboxylate mixture

and optionally further oxidizing the compound (I-q) and (I-r) to the corresponding aldehyde or carboxylic acid;

t) halogenating a compound of formula (I-t) in the presence of a halogenating reagent;

 u) reacting a compound of formula (VII-b) with a reagent of formula (XVI) in the presence of an acid;

$$H_2N-C_{2-4}alkyl-N$$

$$(VII-b)$$

$$(XVI)$$

$$N-C_{2-4}alkyl-N$$

$$(I-u)$$

v) reacting a compound of formula (VII-b) with a reagent of formula (XVII) in the presence of an acid yielding a compound of formula (I-v)

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which optionally can be hydrolyzed in the corresponding 2-hydroxycarbonyl-1-pyridyl compound in the presence of an acid or a base;

w) formylating a compound of formula (I-u) in a reaction-inert solvent yielding a compound of formula (I-w)

$$\begin{array}{c|c}
 & H \\
 & C=0 \\
\hline
 & N-C_{2-4}alkyl-N \\
\hline
 & (I-w)
\end{array}$$

which optionally may be reduced in a reaction-inert solvent in the presence of a reductant yielding an alcohol of formula (I-x)

$$\begin{array}{c}
H \\
C = O \\
N - C_{2-4}alkyl - N
\end{array}$$

$$\begin{array}{c}
CH_{2}OH \\
N - C_{2-4}alkyl - N
\end{array}$$

$$\begin{array}{c}
N - C_{2-4}alkyl - N
\end{array}$$

$$\begin{array}{c}
(I-x)
\end{array}$$

15 x) reacting a compound of formula (I-z) with a reagent of formula (XVIII) in the presence of a base yielding a compound of formula (I-y)

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which optionally may be hydrolyzed in the presence of an acid or a base yielding the corresponding hydroxycarbonyl compound;

y) reacting a compound of formula (I-z) with a reagent of formula (XIX) in the presence of benzyl trimethyl ammonium hydroxide in a reaction-inert solvent yielding a compound of formula (I-aa)

which optionally can be hydrolyzed in the presence of an acid or a base into the corresponding hydroxycarbonyl compound;

15 and, if desired, convering the compounds of formula (I) into each other following art-known functional group transformation reactions, and further, if desired,

WO 92/22551 PCT/EP92/01330

converting the compounds of formula (I) into a therapeutically active non-toxic addition salt form by treatment with an acid or a base; or conversely, converting the salt into the free base or acid with alkali, respectively acid; and/or preparing stereochemically isomeric forms thereof.

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INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 92/01330

I. CLASSIFICATION OF		on symbols apply, indicate all) ⁶	
Int.Cl. 5 CO7D4	Patent Classification (a. C) or to both National 87/04; C07D519/00; 10,223:00)(C07D519/00,513	A61K31/55; //	(C07D487/00
II. FIELDS SEARCHED			
	Minimum Doc	umentation Searched	
Classification System		Classification Symbols	
Int.Cl. 5	CO7D ; A61K		
		ther than Minimum Documentation ats are Included in the Fields Searched ⁸	
	IDERED TO BE RELEVANT ⁹	the of the column accorded 12	Relevant to Claim No.13
Category ° Citatio	n of Document, 11 with indication, where appr	opriate, of the reievant passages	Raevant to Claim 1.44
A EP,/	A,O 000 716 (MERCK) 21 Fe claims 1,13	ebruary 1979	1,6
A EP, A	7,0 378 254 (JANSSEN) 18 claim 1; example 15	July 1990	1,6
° Special categories of		"T" later document published after the inter or priority date and not in conflict with	national filing date the application but
"E" earlier document if filing date "L" document which is cited to e citation or other s	the general state of the art which is not for particular relevance out published on or after the international may throw doubts on priority claim(s) or stablish the publication date of another pecial reason (as specified) g to an oral disclosure, use, exhibition or	cited to understand the principle or the invention "X" document of particular relevance; the cannot be considered novel or cannot b involve an inventive step "Y" document of particular relevance; the cannot be considered to involve an inventive and the considered to involve an inventive and the considered to involve an inventive and the considered to involve and the document is combined with one or more	ory underlying the inimed invention e considered to laimed invention intive step when the
other means	ed prior to the international filing date but	ments, such combination being obvious in the art. "&" document member of the same patent f	to a person skilled
IV. CERTIFICATION			
	tion of the International Search EPTEMBER 1992	Date of Mailing of this International S. 25. 09. 92	earch Report
International Searching As	sthority	Signature of Authorized Officer	h
_	ROPEAN PATENT OFFICE	ALFARO FAUS I.	Iles

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This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information. 08/09/92

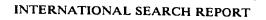
Patent document cited in search report	Publication date	Patent family member(s)		Publication date	
EP-A-0000716	21-02-79	US-A- JP-A-	4148903 54027597	10-04-79 01-03-79	
EP-A-0378254	18-07-90	AU-B- AU-A- JP-A- US-A-	622509 4778090 2233678 5008268	09-04-92 12-07-90 17-09-90 16-04-91	

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PATENT COOPERATION TREATY

PCT





05 JAN 1993

(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference	FOR FURTHER	see Notification of	Transmittal of International Search Report
JAB 812-PCT	ACTION	(Form PCT/ISA/2	20) as well as, where applicable, item 5 below.
International application No.	International filing date(iay/month/year)	(Earliest) Priority Date (day/month/year)
PCT/EP 92/01330	09/06/92		13/06/91
Applicant			
JANSSEN PHARMACEUTICA N.V	•		
This international search report has been according to Article 18. A copy is being to	prepared by this Internatio	nal Searching Autho onal Bureau.	ority and is transmitted to the applicant
This international search report consists It is also accompanied by a cop		sheets. nt cited in this repor	t.
1. Certain claims were found unsea	archable (see Box I).	·	·
2. Unity of invention is lacking (see	e Box II).		
The international application of international search was carried.	ontains disclosure of a nucle I out on the basis of the sec	otide and/or amino a uence listing	acid sequence listing and the
	d with the international app	_	
∑ fur	nished by the applicant sepa		
	but not accompanied matter going beyond	by a statement to th the disclosure in the	e effect that it did not include international application as filed.
Tra	unscribed by this Authority		
4. With regard to the title, X the	text is approved as submit	ted by the applicant.	
the	text has been established b	y this Authority to	read as follows:
		No.	
5. With regard to the abstract,		·	•
X the	text is approved as submit	ted by the applicant.	
Bo	text has been established, x III. The applicant may, w rch report, submit commer	ithin one month fro	3.2(b), by this Authority as it appears in m the date of mailing of this international
6. The figure of the drawings to be put	olished with the abstract is:		
Figure No as	suggested by the applicant.		None of the figures.
	cause the applicant failed to	-	
bed	cause this figure better char	acterizes the inventi	on.
1			

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 92/01330

	ICATION OF SUBJE		symbols apply, indicate all) ⁶	
	5 CO7D487/	Classification (IPC) or to both National (1)4; C07D519/00; 23:00)(C07D519/00,513:	Classification and IPC A61K31/55; // 00,487:00)(C07D519:00,	(C07D487/00
II. FIELDS	SEARCHED			
		Minimum Docum	nentation Searched ⁷	
Classificat	ion System		Classification Symbols	
Int.C1	. 5	CO7D ; A61K		-
		Documentation Searched othe to the Extent that such Document:	r than Minimum Documentation s are Included in the Fields Searched ⁸	
III. DOCU	MENTS CONSIDER	ED TO BE RELEVANT ⁹		
Category °		ocument, 11 with indication, where approp	riate, of the relevant passages 12	Relevant to Claim No.13
A	EP,A,O	000 716 (MERCK) 21 Feb		1,6
A	EP,A,O	378 254 (JANSSEN) 18 J im 1; example 15	July 1990	1,6
"A" do	onsidered to be of parti arlier document but pui iling date ocument which may the hich is cited to establis tation or other special ocument referring to a ther means	eneral state of the art which is not cular relevance blished on or after the international row doubts on priority claim(s) or the publication date of another reason (as specified) an oral disclosure, use, exhibition or to the international filing date but	"T" later document published after the inter- or priority date and not in conflict with cited to understand the principle or the invention "X" document of particular relevance; the cl cannot be considered novel or cannot be involve an inventive step "Y" document of particular relevance; the cl cannot be considered to involve an inve- document is combined with one or more ments, such combination being obvious in the art. "&" document member of the same patent for	the application but pry underlying the aimed invention a considered to taimed invention ntive step when the a other such docu- to a person skilled
	TIFICATION			neach Benort
Date of th		f the International Search EMBER 1992	Date of Mailing of this International Se	\
Internation	nal Searching Authorit EUROP	y EAN PATENT OFFICE	Signature of Authorized Officer ALFARO FAUS I.	Hey
Parm PCT/IS	A/210 (second sheet) (Jan	ury 1915)		See notes on accompanying she

Form PCT/ISA/210 (second sheet) (January 1985)

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information. 08/09/92

Patent document cited in search report	Publication date		Patent family member(s)		Publication date
EP-A-0000716	21-02-79	US-A- JP-A-	4148903 54027597		-04-79 -03-79
EP-A-0378254	18-07-90	AU-B- AU-A- JP-A- US-A-	622509 4778090 2233678 5008268	12 17	-04-92 -07-90 -09-90 -04-91

From the INTERNATIONAL SEARCHING AUTHORITY



	· PUI
To: JANSSEN PHARMACEUTICA N.V. Attn. Wante, Dirk Turnhoutseweg 30 2340 BEERSE BELGIUM INTL LAW DIVISION	NOTIFICATION OF TRANSMITTAL OF THE INTERNATIONAL SEARCH REPORT OR THE DECLARATION (PCT Rule 44.1) Leach Lept Response for 11/25/92.
OCT 6 1992	Leat Responde fue 11/2011
TO FILE REFER TO SEE	Date of maling (day/month/year) 25. 04. 92
Applicant's or agent's file reference	
JAB 812-PCT	FOR FURTHER ACTION See paragraphs 1 and 4 below
International application No. PCT/EP 92/01330	International filing date (day/month/year) 09/06/92
Applicant	
JANSSEN PHARMACEUTICA N.V.	3 0 SEP. 1992
1. X The applicant is hereby notified that the international search Filing of amendments and statement under Article 19: The applicant is entitled, if he so wishes, to amend the claim When? The time limit for filing such amendments is nor international search report; however, for more detailed Where? To the International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland Fascimile No.: (41-22) 740.14.35 For more detailed instructions, see the notes on the accompany of the applicant is hereby notified that no international search Article 17(2)(a) to that effect is transmitted herewith.	ns of the international application (see Rule 46): mally 2 months from the date of transmittal of the etails, see the notes on the accompanying sheet.
3. With regard to the protest against payment of (an) addition the protest together with the decision thereon has be	nal fee(s) under Rule 40.2; the applicant is notified that: en transmitted to the International Bureau together with the
applicants's request to forward the texts of both the no decision has been made yet on the protest; the ap 4.Further action(s): The applicant is reminded of the following Shortly after 18 months from the priority date, the international a If the applicant wishes to avoid or postpone publication, a not priority claim, must reach the International Bureau as provided completion of the technical preparations for international publi	;: application will be published by the International Bureau. ce of withdrawal of the international application, or of the in Rules 90bis.1 and 90bis.3, respectively, before the
Within 19 months from the priority date, a demand for internation wishes to postpone the entry into the national phase until 30 m	onal preliminary examination must be filed if the applicant nonths from the priority date (in some Offices even later).
Within 20 months from the priority date, the applicant must perfuse before all designated Offices which have not been elected within because they are not bound by Chapter II.	orm the prescribed acts for entry into the national phase n 19 months from the priority date or could not be elected

Name and mailing address of the International Searching Authority

European Patent Office, P.B. 5818 Patentlaan 2 NL-2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl, Fax: (+31-70) 340-3016

Natalie Weinberg

Form PCT/ISA/220 (July 1992)

P TENT COOPERATION TREAT

PCT



(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference	FOR FURTHER	see Notification of	f Transmittal of International Search Report
JAB 812-PCT	ACTION	(Form PCT/ISA/2	220) as well as, where applicable, item 5 below.
International application No.	International filing date(lay/month/year)	(Earliest) Priority Date (day/month/year)
PCT/EP 92/01330	09/06/92		13/06/91
Applicant			
JANSSEN PHARMACEUTICA N.V	•		
This international search report has been according to Article 18. A copy is being t	prepared by this Internatio	nal Searching Authonal Bureau.	ority and is transmitted to the applicant
This international search report consists of It is also accompanied by a cop		sheets.	r t .
1. Certain claims were found unsea	archable (see Box I).		
2. Unity of invention is lacking (see	e Box II).		
3. The international application cointernational search was carried	ontains disclosure of a nucle l out on the basis of the seq	otide and/or amino a uence listing	acid sequence listing and the
	d with the international app		
furi	nished by the applicant sepa		
1	but not accompanied matter going beyond	by a statement to th the disclosure in the	ne effect that it did not include international application as filed.
Tra	unscribed by this Authority		
I ===	text is approved as submitt		
the	text has been established b	y this Authority to	read as follows:
5. With regard to the abstract,			
X the	text is approved as submit	ted by the applicant	•
Во	text has been established, a x III. The applicant may, w rch report, submit commen	ithin one month fro	8.2(b), by this Authority as it appears in om the date of mailing of this international
6. The figure of the drawings to be pub	olished with the abstract is:		
Figure No as	suggested by the applicant.		None of the figures.
[cause the applicant failed to		
bed	cause this figure better char	acterizes the inventi	on.

International Application No

PCT/EP 92/01330

	1. 5 CO7D487/	Classification (IPC) or to both Nationa 04; C07D519/00; 23.00)(C07D519/00.513	A61K31/55; // :00,487:00)(C07D519:00,	/(C07D487/00
II. FIEL	DS SEARCHED	23.00)(00/0313/00,313	.00, 407.007(0070013.00)	
		Minimum Doc	umentation Searched?	
Classifi	cation System		Classification Symbols	
Int.C		CO7D ; A61K		
		Documentation Searched of to the Extent that such Documen	her than Minimum Documentation ats are Included in the Fields Searched ⁸	
		ED TO BE RELEVANT ⁹	12	Relevant to Claim No
Category	° Citation of D	ocument, 11 with indication, where appro	opriate, or the relevant passages	Reservant to Calab 140
A		000 716 (MERCK) 21 Fe	bruary 1979	1,6
A	EP,A,O see cla	378 254 (JANSSEN) 18 in 1; example 15	July 1990	1,6
		•		
· Sr	ecial categories of cited (locuments : ¹⁰	"T" later document published after the inte	rnational filing date
		general state of the art which is not	or priority date and not in conflict wit cited to understand the principle or th invention	eory underlying the
Æ,	earlier document but pu	blished on or after the international	"X" document of particular relevance: the	claimed invention
T.	filing date document which may the	row doubts on priority claim(s) or sh the publication date of another	cannot be considered novel or cannot involve an inventive step "Y" document of particular relevance; the	
	citation or other special	reason (as specified) un oral disclosure, use, exhibition or	cannot be considered to involve an inv	rentive step when the re other such docu-
	other means	or to the international filing date but	ments, such combination being obvious in the art. "A" document member of the same patent	s to a person skiller
īv. a	ERTIFICATION			
Date of	f the Actual Completion o	of the International Search	Date of Mailing of this International 5	Search Report
	08 SEPT	EMBER 1992	= 0. 03. 3 £	\
1				
1 Interns	itional Searching Authori	ty	Signature of Authorized Officer ALFARO FAUS I.	h.o.

Form PCT/ISA/210 (second sheet) (January 1985)

NOTES TO FORM PCT/ISA/220



These notes are intended to give the basic instructions concerning the filing of amendments under article 19. The Notes are based on the requirements of the Patent Cooperation Treaty and of the Regulations and the Administrative Instructions under that Treaty. In case of discrepancy between these Notes and those requirements, the latter are applicable. For more detailed information, see also the PCT Applicant's Guide, a publication of WIPO.

In these notes, "Article", "Rule", and "Section" refer to the provisions of the PCT, the PCT Regulations and the PCT administrative Instructions respectively.

INSTRUCTIONS CONCERNING AMENDMENTS UNDER ARTICLE 19

The applicant has, after having received the international search report, one opportunity to amend the claims of the international application. It should however be emphasized that, since all parts of the international application (claims, description and drawings) may be amended during the international preliminary examination procedure, there is usually no need to file amendments of the claims under Article 19 except where, e.g. the applicant wants the latter to be published for the purposes of provisional protection or has another reason for amending the claims before international pbulication. Furthermore, it should be emphasized that provisional protection is available in some States only.

What parts of the international application may be amended?

The claims only.

The description and the drawings may only be amended during international preliminary examination under

When?

Within 2 months from the date of transmittal of the international search report or 16 months from the priority date, whichever time limit expires later. It should be noted, however, that the amendments wil be considered as having been received on time if they are received by the International Bureau after the expiration of the applicable time limit but before the completion of the technical preparations for international publication (Rule 46.1).

Where not to file the amendments?

The amendments may only be filed with the International Bureau and not with the receiving Office or the International Searching Authority (Rule 46.2).

Where a demand for international preliminary examination has been/is filed, see below.

How?

Either by cancelling one or more entire claims, by adding one or more new claims or by amending the text of one or more of the claims as filed.

A replacement sheet must be submitted for each sheet of the claims which, on account of an amendment or amendments, differs from the sheet originally filed.

All the claims appearing on a replacement sheet must be numbered in Arabic numerals. Where a claim is cancelled, no renumbering of the other claims is required. In all cases where claims are renumbered, they must be renumbered consecutively (Administrative Instructions, Section 205(b)).

What documents must/may accompany the amendments?

Letter (Section 205(b)):

The amendments must be submitted with a letter.

The letter will not be published with the international application and the amended claims. It should not be confounded with the "Statement under Article 19(1)" (see below, under "Statement under Article 19(1)").

The letter must indicate the differences between the claims as filed and the claims as amended. It must, in particular, indicate, in connection with each claim appearing in the international application (it being understood that identical indications concerning several claims may be grouped), whether

- the claim is unchanged;
- (ii) the claim is cancelled;
- (iii) the claim is new;
- (iv) the claim replaces one or more claims as filed;
- the claim is the result of the division of a claim as filed.



The following examples illustrate the manner in which amendments must be explained in the accompanying letter:

- [Where originally there were 48 claims and after amendment of some claims there are 51]:
 "Claims 1 to 29, 31, 32, 34, 35, 37 to 48 replaced by amended claims bearing the same numbers;
 Claims 30, 33 and 36 unchanged; new claims 49 to 51 added."
- [Where originally there were 15 claims and after amendment of all claims there are 11]: "Claims 1 to 15 replaced by amended claims 1 to 11."
- 3. [Where originally there were 14 claims and the amendments consist in cancelling some claims and in adding new claims]:
 "Claims 1 to 6 and 14 unchanged; claims 7 to 13 cancelled; new claims 15, 16 and 17 added." or "Claims 7 to 13 cancelled; new claims 15, 16 and 17 added; all other claims unchanged."
- 4. [Where various kinds of amendments are made]: "Claims 1-10 unchanged; claims 11 TO 13, 18 and 19 cancelled; claims 14, 15 and 16 replaced by amended claim 14; claim 17 subdivided into amended claims 15, 16 and 17; new claims 20 and 21 added."

"Statement under article 19(1)" (Rule 46.4)

The amendments may be accompanied by a statement explaining the amendments and indicating any impact that such amendments might have on the description and the drawings which cannot be amended under Article 19(1).

The statement will be published with the international application and the amended claims.

The statement should be brief, it should not exceed 500 words if in English or if translated into English.

It should not be confouded with and does not replace the letter indicating the differences between the claims as filed and as amended. It must be filed on a separate sheet and must be identified as such by a heading, preferably by using the words "Statement under Article 19(1)."

It should not contain any disparaging comments on the international search report or the relevance of citations contained in that report. Reference to citations, relevant to a given claim, contained in the international search report may be made only in connection with an amendment of that claim.

In what language?

The amendments must be made in the language in which the international application is published. The letter and any statement accompanying the amendments must be in the same language as the international application if that language is English of French; otherwise, it must be in English or French, at the choice of the applicant.

Consequence if a demand for international preliminary examination has already been filed?

If, at the time of filing any amendments under Article 19, a demand for international preliminary examination has already been submitted, the applicant must preferably, at the same time of filing the amendments with the International Bureau, also file a copy of such amendments with the International Preliminary Examining Authority (see Rule 62.2(a), first sentence).

Consequence with regard to translation of the international application for entry into the national phase?

The applicant's attention is drawn to the fact that, where upon entry into the national phase, a translation of the claims as amended under Article 19 may have to be furnished to the designated/elected Offices, instead of, or in addition to, the translation of the claims as filed.

For further details on the requirements of each designated/elected Office, see Volume II of the PCT Applicant's Guide.

This annex lists the patent family members relating to the patent documents cited in the above-mentioned international search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information. 08/09/92

Patent document cited in search report	Publication date		Patent family member(s)	Publication date	
EP-A-0000716	21-02-79	US-A- JP-A-	4148903 54027597	10-04-79 01-03-79	
EP-A-0378254	18-07-90	AU-B- AU-A- JP-A- US-A-	622509 4778090 2233678 5008268	09-04-92 12-07-90 17-09-90 16-04-91	

PATENT COOPERATION TREATY

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FROM THE INTERNATIONAL PRELIMINARY Registered letter EXAMINING AUTHORITY Wante, Dirk JANSSEN PHARMACEUTICA N.V. WRITTEN OPINION issued pursuant to PCT rules 66.2.(1) or 66.4(a) (2) Turnhoutseweg 30 HE WELL TO WELL 2340 BEERSE 19 APR. 1993 BELGIQUE INTL. LAW DIVISION DATE OF MAILING by the International Preliminary Pamining Authority 1 5, 04, 93 Inscribe NAME and ADDRESS of the AGENT 13 FILE REFER TO PLICANT'S OF AGENT'S FILE REFERENCE or if there is no agent, of the APPLICANT___ JARSSEN_ANSWERED JAB 812-PCT BRING FILE DOCKET IDENTIFICATION OF THE INTERNATIONAL APPLICATION International Filing Date International Application No. 09/06/1992 PCT/EP 92/01330 Applicant (Name) Response Due JANSSEN PHARMACEUTICA N.V. et al. Priority date claimed Receiving Office 13/06/1991 RO/ EP WRITTEN OPINION With reference to the above-identified international application, this constitutes the first written opinion by this International Preliminary Examining Authority. I. BASIS OF OPINION The examination is being carried out on the following application documents: the application documents as filed , as originally filed description, pages description, pages ... , as originally filed claim(s) claim(s) , as originally filed This opinion has been established as if the amendments indicated on the extra sheet have not been made, since, for the reasons, they have been considered to go beyond the disclosure as filed. II. NON-ESTABLISHMENT OF OPINION ON NOVELTY, INVENTIVE STEP AND INDUSTRIAL APPLICABILITY The question whether the claimed invention appears to be novel, to involve an inventive step (to be non-obvious), and to be industrially applicable will not for the reasons indicated below be gone into in respect of: the above-identified international application. __ (specify particular claims). 2. relate to the following subject matter3) Said international application, or said claims Nos. which does not require an international preliminary examination. (specify) The description, claims, or drawings ((indicate particular elements) or said claims Nos. _ that no meaningful opinion could be formed. 3 are so inadequately supported by the description that no meaningful The claims, or said claims Nos. opinion could be formed. 3)

Form PCT/IPEA/408(first sheet)/01.91 P20479

424

(15/12/1992)

See notes on accompanying sheet

	WRITTEN OPINION (continued)
III.	NEGATIVE STATEMENT REGARD TO NOVELTY, INVENTIVE STEP AN INDUSTRIAL APPLICABILITY OF CLAIM.
The	e statement under Article 35 (2) should be negative in respect of the claims indicated below. e criteria not satisfied in respect of such claims are indicated by the letter abbreviation. for Novelty); IS (for Inventive Step); IA (for Industrial Applicablity).
	1 - 10 : ET
IV.	CITATIONS AND EXPLANATIONS IN REGARD TO NOVELTY, INVENTIVE STEP AND INDUSTRIAL APPLICABILITY OF CLAIMS
No	. of Claim / Relevant Supporting Documents Cited / Explanation
	please see separate sheet
İ	

V. CERTAIN DEFECTS IN THE VITERNATIONAL APPLICATION



The following defects in the form or contents of the above-identified international application under the Treaty or the Regulation have been noted.

All the relevant prior has not yet been considered in the application.

VI. CERTAIN OBSERVATIONS ON THE INTERNATIONAL APPLICATION

The following observations on the clarity of the claims, description, and drawings or on the question whether the claims are fully supported by the description, are notified:

please dee point 6 of the separate sheet

VII. INVITATION

APPLICANT IS INVITED TO SUBMIT A WRITTEN REPLY ACCOMPANIED, WHERE APPROPRIATE, BY AMENDMENTS (4) WITHIN 3 MONTHS/ ---- DAYS OF THE DATE OF MAILING INDICATED ON THE FIRST SHEET.

July 15, 93

THE INTERNATIONAL PRELIMINARY EXAMINING AUTHORITY

Name and Mailing Adress

___________ Eu ______ Pa

European Patent Office

Erhardtstraße 27 D-8000 München 2 2 089 / 2399-0 Tx 523 656 epmu d FAX 089 / 23 99-44 65 Authorized Officer

· Coller

L.A. Feller

1. Cited documents

WO-A-9206981 = D1

J. Med. Chem, 26(1983), 974-980 = D2

EP-A-0339978 = D3

EP-A-0000716 = D4

WO-A-8813138 = D5

The indicated designation is used throughout the examination procedure.

2. Novelty

The subject matter of the Claim 1 and Claim 9 (intermediates) differs from that of the cited prior art D2-D5 essentially due to the imidazo condensation in the tricyclic moiety. The disclaimer contained in Claims 1 and 9 refers obviously to compounds known from D1 which is an earlier document. In the absence of the priority documents it cannot be decided whether the claimed priority date is acceptable. On the condition that the claimed priority date is justifiable D1 may remain outside considersation during the present examination stage but is relevant during a possible regional phase; it is stressed that exclusion of specific compounds may not be sufficient to delimit from D1 (overlap).

The subject matter of the Claims 1 and 9 can be considered to be novel in view of D2-D5.

3. Inventive step - broadness of the claims

3.1 Subjective problem

According to the application (see page 1, line 26) the problem underlying the application appears to be the follow-

ing:

Provision of further benzazepine derivatives which have favourable antiallergic activity.

3.2 Closest prior art

D2-D5 are especially relevant: The compounds of these documents come structurally very close to those of your application and they also appear to show qualitatively the same antiallergic activity. D2 and D3 referring to specific pyrrolo condensed benzazepines are considered to be the closest prior art.

3.3 Problem which has been objectively solved

In view of the information given in the description page 30/31 it may be considered credible that the technical problem as defined under point 3.1 has been solved by the prepared compounds although no specific test results have been disclosed.

3.4 Evaluation of the solution of the problem

The solution to the problem defined above is considered to be obvious for the following reasons:

For the subject matter of Claim 1 D2 and D3 are closest prior art (see point 3.2).

With these products the problem defined under point 3.1 is also solved.

From D2 and D3 the person skilled in the art searching for a solution to this problem would have considered benzazepine derivatives condensed with further heterocyclic 5-membered rings.

Furthermore, D4 and D5 describe benzazepine derivatives condensed with 6-membered rings indicating that the kind of con-

densation does not alter the qualitative activity profile. The solution of the technical problem according to point 3.1 is therefore an obvious result from the prior art. Consequently the subject matter of Claim 1 can at the moment not considered to be inventive.

3.5 Proposals to meet the objection according to point 8.5

Inspite of the opinion indicated above the presence of an inventive step could be acknowledged if it is made credible by test results that apart from the problem defined under point 3.1 another more exacting problem, which can be deduced from the original application (e.g. surprising improvements), has actually been solved by using originally declosed technical characteristics which should be incorporated in claim 1. It should be taken into consideration that only the structurally closest compounds of the closest prior art D2 and D3 are useful for a meaningful comparison (see T1/80, O. J. EPO 1981, 206). A possible pair for comparison is e.g. a compound of D2 formula 4 and a compound of D3 (first compound of page 8) on the one hand and a compound of the application whereby the corresponding substituents are the same on the other.

3.6 Broadness of the claims

The breadth should be such that all objections indicated above are met and it also could be expected that the compounds comprised would actually solve the problem underlying the invention.

4. Intermediates

The intermediates of Claim 9 are only to be considered to be inventive if it turns out that the endproducts are inventive.

Sheet

5. Industrial applicability

As far as the subject matter claimed can actually be prepared and has some utility, no objection arises.

6. Clarity of the claims - conciseness

- 6.1 In Claim 9 definitions identical with those of Claim 1 should be incorporated by reference to Claim 1 thereby avoid ing unnecessary repetition.
- 6.2 Claim 10 is not acceptable in the present form since it contains a contradiction in itself. Whereas the preamble indicates that compounds of Claims 1-5 are to be prepared, certain of the processes listed in the characterising portion of Claim 10 are only useful to prepare certain types of compounds. Claim 10 should therefore be split into several claims whereby the preamble should define the group of compounds which can be prepared by the processes indicated in the characterising portion of the claim.

NOTES TO FORM PCT/IPEA/408

These Notes are intended to facilitate the use of the present form. For full information, see the text of the Patent Cooperation Treaty and the texts of the Regulations and the Administrative Instructions under that Treaty. In case of discrepancy between these Notes and the said texts, the latter are applicable. "Article" refers to Articles of the Treaty, "Rule" refers to Rules of the Regulations and "Section" refers to Sections of the Administrative Instructions.

- (1) "If the International Preliminary Examining Authority
 - (i) considers that the international application has any of the defects described in Article 34(4),
 - (ii) considers that the international preliminary examination report should be negative in respect of any of the claims because the invention claimed therein does not appear to be novel, does not appear to involve an inventive step (does not appear to be non-obvious, or does not appear to be industrially applicable),
 - (iii) notices that there is some defect in the form or contents of the international application under the Treaty or these Regulations,
 - (iv) considers that any amendment goes beyond the disclosure in the international application as filed, or
 - (v) wishes to accompany the international preliminary examination report by observations on the clarity of the claims, the description, and the drawings, or the question whether the claims are fully supported by the description,

the said Authority shall notify the applicant accordingly in writing. Where the national law of the national Office acting as International Preliminary Examining Authority does not allow multiple dependent claims to be drafted in a manner different from that provided for in the second and third sentences of Rule 6.4(a), the International Preliminary Examining Authority may, in case of failure to use that manner of claiming, apply Article 34(4) (b). In such case, it shall notify the applicant accordingly in writing.* (Rule 66.2(a)).

"The notification shall fully state the reasons for the opinion of the International Preliminary Examining Authority." (Rule 66.2 (b)).

"The notification shall invite the applicant to submit a written reply together, where appropriate, with amendments." (Rule 66.2 (c)).

The notification shall fix a time limit for the reply. The time limit shall be reasonable under the circumstances. It shall normally be 2 months after the date of notification. In no case shall it be shorter than 1 month after the said date. It shall be at least 2 months after the said date where the international search report is transmitted at the same time as the notification. In no case shall it be more than 3 months after the said date.* (Rule 66.2 (d)).

- (2) "If the International Preliminary Examining Authority wishes to issue one or more additional written opinions, it may do so, and Rules 66.2 and 66.3 shall apply." (Rule 66.4(a)).
- (3) "If the International Preliminary Examining Authority considers
 - (i) that the international application relates to a subject matter on which the International Preliminary Examining Authority is not required, under the Regulations, to carry out an international preliminary examination, and in the particular case decides not to carry out such examination, or
 - that the description, the claims, or the drawings, are so unclear, or the claims are so inadequately supported by the description, that no meaningful opinion can be formed on the novelty, inventive step (non-obviousness), or industrial applicability, of the claimed invention, the said Authority shall not go into the questions referred to in Article 33(1) and shall inform the applicant of this opinion and the reasons therefor.* (Article 34(4) (a)).

Rule 67 entitled "Subject Matter Under Article 34 (4) (a) (i)" reads as follows:

No International Preliminary Examining Authority shall be required to carry out an international preliminary examination on an international application of, and to the extent to which, its subject matter is any of the following:

- (i) scientific and mathematical theories.
- (ii) plant or animal varities or essentially biological processes for the production of plants and animals, other than microbiological processes and the products of such processes,
- (iii) schemes, rules or methods of doing business, performing purely mental acts or playing games,

NOTES TO FORM PCT/IPEA/408 (Continued)

- (iv) methods for treatment of the human or animal body by surgery of therapy, as well as diagnostic methods,
- (v) more presentations of information,
- (vi) computer programs to the extent that the International Preliminary Examining Authority is not equipped to carry out an international preliminary examination concerning such programs."
- (4) "The applicant may respond to the invitation referred to in Rule 66.2 (c) of the International Preliminary Examining Authority by making amendments or if he disagrees with the opinion of that Authority by submitting arguments, as the case may be, or do both." (Rule 66.3 (a)).

Any response shall be submitted directly to the International Preliminary Examining Authority. (Rule 66.3(b)).

On the request of the applicant, the International Preliminary Examining Authority may give him one or more additional opportunities to submit amendments or arguments. (Rule 66.4 (b)).

"The applicant shall re required to submit a replacement sheet for every sheet of the international application which, on account of an amendment, differs from the sheet originally filed. The letter accompanying the replacement sheets shall draw attention to the differences between the replaced sheets and the replacement sheets. To the extent that any amendment results in the cancellation of an entire sheet, that amendment shall be communicated in a letter." (Rule 66.8 (a)).

"If the international application has been filed in a language other than the language in which it is published, any amendment, as well as any letter referred to in Rule 66.8 (a), shall be submitted in the language of publication." (Rule 66.9).

Amendments to the claims under Article 19 or Article 34(2) (b) may be made either by cancelling one or more entire claims, by adding one or more new claims or by amending the text of one or more of the claims as filed. All the claims appearing on a replacement sheet shall be numbered in arabic numerals. Where a claim is cancelled, no renumbering of the other claims shall be required. In all cases where claims are renumbered, they shall be renumbered consecutively. (Section 205(a)).



CORRESPONDENCE WITH THE EPO ON PCT CHAPTER II DEMANDS

In order to ensure that your PCT chapter II demand is dealt with as promptly as possible you are requested to use the enclosed self-adhesive labels with any correspondence relating to the demand sent to the Munich Office.

One of these labels should be affixed to a prominent place in the upper part of the letter heading or form etc. which you are filing. 216

From the INTERNATIONAL PRELIMINARY EXAMINING AUTHORITY

PCT

Ex

To:

Wante, Dirk JANSSEN PHARMACEUTICA N.V. Turnhoutseweg 30 2340 BEERSE BELGIQUE

7 AUS. 1953

NOTIFICATION OF TRANSMITTAL OF INTERNATIONAL PRELIMINARY EXAMINATION REPORT

AL OFY A

(PCT Rule 71.1)

Date of mailing (day/month/year)

2 5. 08. 93

Applicant's or agent's file reference

PCT/EP 92/01330

JAB 812-PCT

IMPORTANT NOTIFICATION

International application No.

International filing date (day/month/year)

Priority date (day/month/year)

09/06/1992

13/06/1991

Applicant

JANSSEN PHARMACEUTICA N.V. et al.

- The applicant is hereby notified that this International Preliminary Examining Authority transmits herewith the international preliminary examination report and its annexes, if any, established on the international application.
- 2. A copy of the report and its annexes, if any, is being transmitted to the International Bureau for communication to all the elected Offices.
- Where required by any of the elected Offices, the International Bureau will prepare an English translation of the report (but not of any annexes) and will transmit such translation to those Offices.

4. REMINDER

The applicant must enter the national phase before each elected Office by performing certain acts (filing translations and paying national fees) within 30 months from the priority date (or later in some Offices)(Article 39(1))(see also the reminder sent by the International Bureau with Form PCT/IB/301).

Where a translation of the international application must be furnished to an elected Office, that translation must contain a translation of any annexes to the international preliminary examination report. It is the applicant's responsibility to prepare and furnish such translation directly to each elected Office concerned.

For further details on the applicable time limits and requirements of the elected Offices, see Volume II of the PCT Applicant's Guide.

Name and mailing address of the IPEA/

European Patent Office, Erhardtstrasse 27 W-8000 Munich 2

Tel. (+49-89) 2399-0, Tx: 523656 epmu d

Fax: (+49-89) 2399-4465

Authorized officer

D. Gran

Form PCT/IPEA/416 (July 1992) P20473

(15/12/1992)

TENT COOPERATION TREA'.

PCT

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference	<u> </u>			
JAB 812-PCT	FOR FURTHER ACTIO	N See Notificat Preliminary	tion of Transmittal of International Examination Report (Form PCT/IPEA/416)	
International application No.	International filing date (day/month/year)	Priority date (day/month/year)	
PCT/EP 92/01330	09/06/1992		13/06/1991	
International Patent Classification (IPC) or	national classification and I	PC		
	C07D487/04	_		
Applicant				
JANSSEN PHARMACEUTICA N.	V. et al.			
This international preliminary examely Authority and is transmitted to the			national Preliminary Examining	
2. This REPORT consists of a tota	al of <u>g</u> sheets.			
This report is also accompar during international prelimir	nied by ANNEXES, i.e., she nary examination and/or con	neets of the descript taining rectification	ion, claims and/or drawings amended s made before this Authority.	
These annexes consists of a total of	of sheets.			
3. This report contains indications a	nd corresponding pages relat	ting to the following	g items:	
I X Basis of the report				
II Priority				
III Non-establishment of opinion with regard to novelty, inventive step and industrial applicability				
IV Lack of unity of inven	IV Lack of unity of invention			
	vith regard to novelty, invent ions supporting such stateme		al applicability;	
VI 🔀 Certain documents cite	ed			
VII Certain defects in the	international application			
VIII 🔀 Certain observations o	on the international applicati	on		
1				
Date of submission of the demand		Date of completion		
08/12/1992			7 7. 02. 33	
Name and mailing address of the IPEA/		Authorized officer	-	
European Patent Office, Erha W-8000 Munich 2	ardtstrasse 27		7. (.,	
European Patent Office, Erhardtstrasse 27 W-8000 Munich 2 Tel. (+49-89) 2399-0, Tx: 523656 epmu d Fax: (+49-89) 2399-4465 LA. Feller				
		_		

I. Basis of the report	
1. This report has been drawn up on the basis of:	
[x] the international application as original	ly filed.
[] the description, pages	, as originally filed,
pages	, filed with the demand,
pages	, filed with the letter of,
pages	, filed with the letter of,
[] the claims, No	, as originally filed,
	, as amended under Article 19,
	, filed with the demand,
	, filed with the letter of,
	, filed with the letter of,
1 the drawings, sheets/fig	, as originally filed,
	, filed with the demand,
· -	, filed with the letter of
	, filed with the letter of
2. The amendments have resulted in the cancellation	n of: pages:
	figures No.:
3. [] This report has been established as if (som	me of) the amendments had not been made, since they have been
considered to go beyond the disclosure as	

4. Additional observations, if necessary:

٧.	Reasoned statement under Article 35(2) with regard to novelty	, inventive	step and	industrial	applicability;
	citations and explanations supporting such statement				

1. STATEMENT

Novelty (N)	Claims 1-10	
Inventive Step (IS)	Claims	YES
	Claims 1-10	NO
Industrial Applicability (IA)	Claims 1-10	YES
	Claims	NO

2. CITATIONS AND EXPLANATIONS

1. Cited documents

WO-A-9206981 = D1 J. Med. Chem, 26(1983), 974-980 = D2 EP-A-0339978 = D3 EP-A-0000716 = D4 WO-A-8813138 = D5

The indicated designation is used throughout the examination procedure.

2. Novelty

The subject matter of the Claim 1 and Claim 9 (intermediates) differs from that of the cited prior art D2-D5 essentially due to the imidazo condensation in the tricyclic moiety. The disclaimer contained in Claims 1 and 9 refers obviously to compounds known from D1 which is an earlier document. In the absence of the priority documents it cannot be decided

whether the claimed priority date is acceptable. On the condition that the claimed priority date is justifiable D1 may remain outside considersation during the present examination stage but is relevant during a possible regional phase; it is stressed that exclusion of specific compounds may not be sufficient to delimit from D1 (overlap).

The subject matter of the Claims 1 and 9 can be considered to be novel in view of D2-D5.

3. Inventive step - broadness of the claims

3.1 Subjective problem

According to the application (see page 1, line 26) the problem underlying the application appears to be the following: Provision of further benzazepine derivatives which have favourable antiallergic activity.

3.2 Closest prior art

D2-D5 are especially relevant: The compounds of these documents come structurally very close to those of your application and they also appear to show qualitatively the same antiallergic activity. D2 and D3 referring to specific pyrrolo condensed benzazepines are considered to be the closest prior art.

3.3 Problem which has been objectively solved

In view of the information given in the description page 30/31 it may be considered credible that the technical problem as defined under point 3.1 has been solved by the prepared compounds although no specific test results have been disclosed.

3.4 Evaluation of the solution of the problem

The solution to the problem defined above is considered to be obvious for the following reasons:

For the subject matter of Claim 1 D2 and D3 are closest prior art (see point 3.2).

With these products the problem defined under point 3.1 is also solved.

From D2 and D3 the person skilled in the art searching for a solution to this problem would have considered benzazepine derivatives condensed with further heterocyclic 5-membered rings.

Furthermore, D4 and D5 describe benzazepine derivatives condensed with 6-membered rings indicating that the kind of condensation does not alter the qualitative activity profile. The solution of the technical problem according to point 3.1 is therefore an obvious result from the prior art. Consequently the subject matter of Claim 1 can at the moment not considered to be inventive.

3.5 Proposals to meet the objection according to point 8.5

Inspite of the opinion indicated above the presence of an inventive step could be acknowledged if it is made credible by test results that apart from the problem defined under point 3.1 another more exacting problem, which can be deduced from the original application (e.g. surprising improvements), has actually been solved by using originally declosed technical characteristics which should be incorporated in claim 1. It should be taken into consideration that only the structurally closest compounds of the closest prior art D2 and D3 are useful for a meaningful comparison. A possible pair for comparison is e.g. a compound of D2 formula 4 and a compound of D3 (first compound of page 8) on the one hand and a compound of the application whereby the corresponding substituents are the same on the other.

3.6 Broadness of the claims

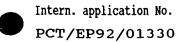
The breadth should be such that all objections indicated above are met and it also could be expected that the compounds comprised would actually solve the problem underlying the invention.

4. Intermediates

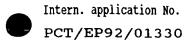
The intermediates of Claim 9 are only to be considered to be inventive if it turns out that the endproducts are inventive.

5. Industrial applicability

As far as the subject matter claimed can actually be prepared and has some utility, no objection arises.



. Certain published docum	ents		
Application No. Patent No.	Publication date (day/month/year)	Filing date (day/month/year)	Priority date (valid claim) (day/month/year)
WO92/06981	30/04/1992	04/10/1991	10/10/1990
. Non-written disclosures	S		
Kind of non-written o		non-written disclosure day/month/year)	Date of written disclosur referring to non-written disc (day/month/year)



VII. Certain defects in the international application

The following defects in the form or contents of the international application have been noted:

All the relevant prior art has not been considered in the application

VIII. Certain observations on the international application

The following observations on the clarity of the claims, description, and drawings or on the question whether the claims are fully supported by the description, are made:

- a) In Claim 9 definitions identical with those of Claim 1 should be incorporated by reference to Claim 1 thereby avoiding unnecessary repetition.
- b) Claim 10 is not acceptable in the present form since it contains a contradiction in itself. Whereas the preamble indicates that compounds of Claims 1-5 are to be prepared, certain of the processes listed in the characterising portion of Claim 10 are only useful to prepare certain types of compounds. Claim 10 should therefore be split into several claims whereby the preamble should define the group of compounds which can be prepared by the processes indicated in the characterising portion of the claim.

			28 Rec	d PCT/P	
TENT	COOPERATION	TREATY			

PATENT COOPERATION TREATY	
	INTERNATIONAL APPLICATION NO. PCT/EP92/01330
NOTIFICATION TO THE DESIGNATED OFFICE OF RECEIPT OF RECORD COPY issued under PCT Rule 24.2(a) APPLICANT'S OR AGENT'S FILE REFERENCE: JAB812-PCT	To: United States Patent and Trademark Office Washington, D.C. in its capacity as a designated Office
DATE OF MAILING OF THIS NOTIFICATION: 29 June 1992 (29.06.92)	From: The International Bureau of WIPO 1211 Geneva 20 Switzerland
NAME(S) OF APPLICANT(S): JANSSENS, Frans, Eduard et al.	
INTERNATIONAL FILING DATE:	June 1992 (09.06.92)
PRIORITY DATE(S) CLAIMED:	June 1991 (13.06.91) March 1992 (18.03.92)
DATE OF RECEIPT OF RECORD COPY	BY INTERNATIONAL BUREAU: June 1992 (29.06.92)
	J. Leitao (Authorized Officer)

PATENT COOPERATION TREAT

From the INTERNATIONAL BUREAU

10 David DOT/2000 4 M FFD 4009	TION CITE INVATIONAL BOILEAU
13 Rec'd PCT/PTPC 17 FEB 1993	То:
NOTIFICATION OF ELECTION	
(PCT Rule 61.2)	United States Patent and Trademark Office Washington, D.C.
Date of mailing: 09 February 1993 (09.02.93)	in its capacity as elected Office
International application No.: PCT/EP92/01330	Applicant's or agent's file reference: JAB812-PCT
International filing date: 09 June 1992 (09.06.92)	Priority date: 13 June 1991 (13.06.91)
Applicant: JANSSENS, Frans, Eduard et al	
1. The designated Office is hereby notified of its election made X in the demand filed with the International preliminary 08 December 1	Examining Authority on: 1992 (08.12.92) ational Bureau on:

The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland

Authorized officer:

J. Leitao

Telephone No.: (41-22) 730.91.11

Facsimile No.: (41-22) 740.14.35

PATENT COOPERATION TREATY

	•
	From the INTERNATIONAL BUREAU
PCT	То:
NOTIFICATION CONCERNING	
DOCUMENT TRANSMITTED	United States Patent and Trademark
	Office
	Washington, D.C.
Date of mailing:	
31 August 1993 (31.08.93)	in its capacity as elected Office
International application No.:	International filing date:
PCT/EP92/01330	09 June 1992 (09.06.92)
Applicant: JANSSEN PHARMACEUTICA N.V. et al	
	·
The International Bureau transmits herewith the following documents	nents and number thereof:
copy of the international preliminary examin	nation report (Article 36(3)(a))
	C (

The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland Authorised officer:

Facsimile No.: (41-22) 740.14.35

C. Carrié

Telephone No.: (41-22) 730.91.11

PATENT COOPERATION TREATY

PCT

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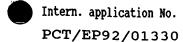
INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)

Applicant's or agent's file reference			
JAB 812-PCT	FOR FURTHER ACTION	See Notifica Preliminary	tion of Transmittal of International Examination Report (Form PCT/IPEA/416)
International application No.	International filing date (day)	month/year)	Priority date (day/month/year)
PCT/EP 92/01330	09/06/1992		13/06/1991
International Patent Classification (IPC) or	national classification and IPC		22, 33, 222
	C07D487/04		
Applicant			
JANSSEN PHARMACEUTICA N.	V. et al.		
This international preliminary exar Authority and is transmitted to the	nination report has been prepare applicant according to Article 3	ed by this Inter	national Preliminary Examining
2. This REPORT consists of a total	of sheets.	-	
This report is also accompanduring international prelimina	ied by ANNEXES, i.e., sheets ary examination and/or containin	of the descriptions	on, claims and/or drawings amended made before this Authority.
These annexes consists of a total o	f sheets.		
3. This report contains indications an	d corresponding pages relating to	the following	items:
I X Basis of the report			•
II Priority			
III Non-establishment of o	pinion with regard to novelty, in	ventive step an	d industrial applicability
IV Lack of unity of invention			
V 🔀 Reasoned statement wit citations and explanatio	h regard to novelty, inventive suns supporting such statement	ep or industrial	applicability;
VI 🔀 Certain documents cited	ı		
VII Certain defects in the in	ternational application		
	the international application	•	
	application of the second of t	•	
			_
			·
Date of submission of the demand	Date	of completion o	of this report
08/12/1992			2 5. 08. 93
Name and mailing address of the IPEA/	Autho	rized officer	
European Patent Office, Erhard	tstrasse 27	117	
W-8000 Munich 2 Tel. (+49-89) 2399-0, Tx: 52365	i6 epmu d	1. A · Z	
Fax: (+49-89) 2399-4465			A. Feller
form PCT/IPEA/409 (cover sheet) (July 1992	2) P20476 (07/04/1993	3 1	

Intern. application No. PCT/EP92/01330

This report has been drawn up on the basis of:	
[imes] the international application as original	ly filed.
[] the description, pages	, as originally filed,
pages	, filed with the demand,
pages	, filed with the letter of
pages	, filed with the letter of
[] the claims, No	, as originally filed,
No	, as amended under Article 19,
No	, filed with the demand,
	, filed with the letter of,
No.	, filed with the letter of,
[] the drawings, sheets/fig	, as originally filed,
sheets/fig	, filed with the demand,
sheets/fig	, filed with the letter of
sheets/fig	, filed with the letter of
The amendments have resulted in the cancellation	n of: pages:
sheets of drawings/f	figures No.:
[] This report has been established as if (som	e of) the amendments had not been made, since they have been
considered to go beyond the disclosure as f	



	tatement under Article 35(2) with rec and explanations supporting such stat	gard to novelty, inventive step and industrial a tement	applicability;
1. STATEMENT			
Novelty (•	ms 1-10	YES
	VIIII		no

Claims ______YES

Claims 1-10_____YES

Claims _____NO

Claims 1-10_____

2. CITATIONS AND EXPLANATIONS

Industrial Applicability (IA)

Inventive Step (IS)

1. Cited documents

WO-A-9206981 = D1 J. Med. Chem, 26(1983), 974-980 = D2EP-A-0339978 = D3 EP-A-0000716 = D4 \checkmark WO-A-8813138 = D5

The indicated designation is used throughout the examination procedure.

2. Novelty

The subject matter of the Claim 1 and Claim 9 (intermediates) differs from that of the cited prior art D2-D5 essentially due to the imidazo condensation in the tricyclic moiety. The disclaimer contained in Claims 1 and 9 refers obviously to compounds known from D1 which is an earlier document. In the absence of the priority documents it cannot be decided

whether the claimed priority date is acceptable. On the condition that the claimed priority date is justifiable D1 may remain outside considersation during the present examination stage but is relevant during a possible regional phase; it is stressed that exclusion of specific compounds may not be sufficient to delimit from D1 (overlap).

The subject matter of the Claims 1 and 9 can be considered to be novel in view of D2-D5.

3. Inventive step - broadness of the claims

3.1 Subjective problem

According to the application (see page 1, line 26) the problem underlying the application appears to be the following: Provision of further benzazepine derivatives which have favourable antiallergic activity.

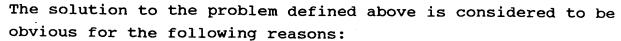
3.2 Closest prior art

D2-D5 are especially relevant: The compounds of these documents come structurally very close to those of your application and they also appear to show qualitatively the same antiallergic activity. D2 and D3 referring to specific pyrrolo condensed benzazepines are considered to be the closest prior art.

3.3 Problem which has been objectively solved

In view of the information given in the description page 30/31 it may be considered credible that the technical problem as defined under point 3.1 has been solved by the prepared compounds although no specific test results have been disclosed.

3.4 Evaluation of the solution of the problem



For the subject matter of Claim 1 D2 and D3 are closest prior art (see point 3.2).

With these products the problem defined under point 3.1 is also solved.

From D2 and D3 the person skilled in the art searching for a solution to this problem would have considered benzazepine derivatives condensed with further heterocyclic 5-membered rings.

Furthermore, D4 and D5 describe benzazepine derivatives condensed with 6-membered rings indicating that the kind of condensation does not alter the qualitative activity profile. The solution of the technical problem according to point 3.1 is therefore an obvious result from the prior art. Consequently the subject matter of Claim 1 can at the moment not considered to be inventive.

3.5 Proposals to meet the objection according to point 8.5

Inspite of the opinion indicated above the presence of an inventive step could be acknowledged if it is made credible by test results that apart from the problem defined under point 3.1 another more exacting problem, which can be deduced from the original application (e.g. surprising improvements), has actually been solved by using originally declosed technical characteristics which should be incorporated in claim 1. It should be taken into consideration that only the structurally closest compounds of the closest prior art D2 and D3 are useful for a meaningful comparison. A possible pair for comparison is e.g. a compound of D2 formula 4 and a compound of D3 (first compound of page 8) on the one hand and a compound of the application whereby the corresponding substituents are the same on the other.

3.6 Broadness of the claims

Intern. application No. PCT/EP92/01330

The breadth should be such that all objections indicated above are met and it also could be expected that the compounds comprised would actually solve the problem underlying the invention.

4. Intermediates

The intermediates of Claim 9 are only to be considered to be inventive if it turns out that the endproducts are inventive.

5. Industrial applicability

As far as the subject matter claimed can actually be prepared and has some utility, no objection arises.

Intern. application No. PCT/EP92/01330

VI. Certain documents cite	ed		
1. Certain published docum	ents		
Application No. Patent No.	Publication da (day/month/yea		Priority date (valid claim) (day/month/year)
WO92/06981	30/04/199	04/10/1991	10/10/1990
2. Non-written disclosures	3		•
Kind of non-written d	isclosure 1	oate of non-written disclosure (day/month/year)	Date of written disclosure referring to non-written disclosure (day/month/year)

Intern. application No. PCT/EP92/01330

VII. Certain defects in the international application

The following defects $\ddot{\text{in}}$ the form or contents of the international application have been noted:

All the relevant prior art has not been considered in the application

Intern. application No. PCT/EP92/01330

VIII. Certain observations on the international application

The following observations on the clarity of the claims, description, and drawings or on the question whether the claims are fully supported by the description, are made:

- a) In Claim 9 definitions identical with those of Claim 1 should be incorporated by reference to Claim 1 thereby avoiding unnecessary repetition.
- b) Claim 10 is not acceptable in the present form since it contains a contradiction in itself. Whereas the preamble indicates that compounds of Claims 1-5 are to be prepared, certain of the processes listed in the characterising portion of Claim 10 are only useful to prepare certain types of compounds. Claim 10 should therefore be split into several claims whereby the preamble should define the group of compounds which can be prepared by the processes indicated in the characterising portion of the claim.

RECORD COPY

INTERNATIONAL A UNDER TO PATENT COOPERATION TREATY

REQUEST

THE UNDERSIGNED REQUESTS THAT THE PRESENT INTERNATIONAL APPLICATION BE PROCESSED ACCORDING TO THE PATENT COOPERATION TREATY

11 Rec'd PCT/PTO 05 JAN 19 4 2 / 0 1 3 3 0 (The following is to be filled in INTERNATION APPLICATION AS TEP **0** 9. 06. 92 INTERNATIONAL FILING DATE: JUN 1992 **SUROPEAN PATENT OFFICE** PCT INTERNATIONAL APPLICATION

Name of receiving Office and "PCT International Application"

ACCORDING TO THE PA	TENT COOLE	RATION TREATT		nt's or agent's file ed by applicant if		JAB 812-l	PCT
Box No. I TITLE OF INVENTION Imidazo[2,1-b][3]benzazepine derivatives, compositions and method of use							
Box No. II APPLICANT IS APPLICANT. Use this (includes, where applicable,	box for indica	ting the applicant or,	if there	are several applica	ED STATES I	FOR WHICH Hem. If more than	IE/SHE/IT
The person identified in this box is (mark one check-box only): Name and address: ** applicant and inventor * applicant only							
JANSSEN P Turnhoutsew B-2340-Beer Belgium	veg 30	EUTICA N.V.					
Telephone number (including	-	Telegraphic addres	-		Teleprinter	address:	•
014/60.21		JANSS	_			32540 jani	far b
State of nationality: Be The person identified in this	lgium s box is <i>applica</i>	nt for the purposes of	State of mark of	of residence: * ne check-box only)): BE		
all designated States		ated States except d States of America		the United Stat of America onl		the States indic in the "Suppler	
Box No. III FURTHER APPLICANTS, IF ANY; (FURTHER) INVENTORS, IF ANY; DESIGNATED STATES FOR WHICH THEY ARE APPLICANTS (IF APPLICABLE). A separate sub-box has to be filled in in respect of each person (includes, where applicable, a legal entity). If the following two sub-boxes are insufficient, continue in the "Supplemental Box," (giving there for each additional person the same indications as those requested in the following two sub-boxes) or by using a "continuation sheet." The person identified in this sub-box is (mark one check-box only): Name and address:**							
Tinstra)-Bonheide					-	
If the person identified in the	his sub-box is a	pplicant (or applicant	and inven	tor), indicate also:	:		
State of nationality: Bell and whether that person is all designated States	all design	e purposes of (mark of ated States except d States of America				the States indicin the "Suppler	
The person identified in this Name and address:**	s sub-box is (m	ark one check-box on	ly):	X applicant inventor *	and	applicant only	inventor only *
Oosteir	nde 12)-Ravels	anislas Marcella	ı				
If the person identified in the	his sub-box is a	applicant (or applicant	and inven	tor), indicate also:	:		
State of nationality: Bell and whether that person is all designated States	all design	e purposes of (mark of ated States except d States of America				the States indicin the "Supple	
 If the person indicated as "applicant and inventor" or as "inventor only" is not an <i>inventor</i> for the purposes of all the designated States, give the necessary indications in the "Supplemental Box." Indicate the name of a natural person by giving his/her family name first followed by the given name(s). Indicate the name of a legal entity by its full official designation. In the address, include both the postal code (if any) and the State (name). 							

*** If residence is not indicated, it will be assumed that the State of residence is the same as the State indicated in the address.

(Stamp)

Sheet number....2

Box No. III CONTINUATION REQUIRED) FURTHER ANY; DESIGNATED STATES WHICH THEY ARE A to be filled in in respect of each person (includes, where appl	PPLICANTS (IF APP	(FURTHER) INVENTORS, IF ABLE). A separate sub-box has
The person identified in this sub-box is (mark one check-box only): Name and address: **	X applicant and inventor*	applicant inventor only *
LEENAERTS, Joseph Elisabeth Potbergstraat 35 B-2310-Rijkevorsel Belgium		
If the person identified in this sub-box is applicant (or applicant and in		
State of nationality: Belgium State and whether that person is applicant for the purposes of (mark one ch all designated States except the United States of America	eck-box only): X of America only	the States Indicated in the "Supplemental Box"
The person identified in this sub-box is (mark one check-box only): Name and address: **	applicant and inventor *	applicant inventor only *
If the person identified in this sub-box is applicant (or applicant and in	nventor), indicate also:	•
	ate of residence:***	
and whether that person is applicant for the purposes of (mark one chall designated all designated States except the United States of America	the United States of America only	the States indicated in the "Supplemental Box"
The person identified in this sub-box is (mark one check-box only): Name and address: **	applicant and inventor*	applicant inventor only *
If the person identified in this sub-box is applicant (or applicant and in	nventor), indicate also:	-
State of nationality: Stand whether that person is applicant for the purposes of (mark one ch	ate of residence:***	
all designated States all designated the United States of America	the United States of America only	the States indicated in the "Supplemental Box"
The person identified in this sub-box is (mark one check-box only): Name and address: **	applicant and inventor •	applicant inventor only *
. **		
If the person identified in this sub-box is applicant (or applicant and in		
State of nationality: and whether that person is applicant for the purposes of (mark one ch all designated all designated States all designated States except the United States of America	ate of residence:*** eck-box only): the United States of America only	the States indicated in the "Supplemental Box"
If the person indicated as "applicant and inventor" or as "invent	or only" is not an inventor for	
States, give the necessary indications in the "Supplemental box." Indicate the name of a natural person by giving his/her family name	ne first followed by the given i	name(s). Indicate the name of a lega
entity by its full official designation. In the address, include both *** If residence is not indicated, it will be assumed that the State of a	• • • • • • • • • • • • • • • • • • • •	•
f this continuation cheet is not used it need not be included in the D		

Box No. IV AGENT (IF ANY) R. COMMON REPRESENTATIVE (IF ANY); ADDR. S FOR NOTIFICATIONS (IN CERTAIN CASES). A common representative may be appointed only if there are several sicants and if no agent is or has been appointed; the common representative must be one of the applicants. The following person (includes, where applicable, a legal entity) is hereby/has been appointed as agent or common representative to act on behalf of the applicant(s) before the competent International Authorities:					
Name and address, including postal code and country: If the space below is used instead for an address for notifications, mark here:					
WAN	NTE, Dirk	-			
	nssen Pharmaceutica N.V.				
	ent Department Thoutseweg 30				
	2340-Beerse				
Bel;	gium oer (including area code): Telegraphic address:			Teleprinter address:	
	32 14 60 31 29 32540 janfar	h		00 32 14 60 55 22	
Box No. V D PROTECTION	DESIGNATION OF GROUPS OF STATES ON OR TREATMENT. The following designations a	OR S	TATE reby m	CS(1); CHOICE OF CERTAIN KINDS OF lade (please mark the applicable check-boxes):	
Regional Paten	nt				
DE	EP European Patent (2): AT Austria, BE Belgium, CH and LI Switzerland and Liechtenstein, DE Germany, DK Denmark, ES Spain, FR France, GB United Kingdom, GR Greece, IT Italy, LU Luxembourg, MC Monaco, NL Netherlands, SE Sweden, and any other State which is a Contracting State of the European Patent Convention and of the PCT				
OA OAPI Patent: Benin, Burkina Faso, Cameroon, Central African Republic, Chad, Congo, Côte d'Ivoire, Gabon, Guinea, Mali, Mauritania, Senegal, Togo, and any other State which is a Contracting State of OAPI and of the PCT; if other OAPI title desired, specify on dotted line(3):					
National Paten	at (if other kind of protection or treatment desired, sp				
				4.	
	stria ⁽³⁾	M		Republic of Korea (3)	
	stralia (3)	X		Sri Lanka	
X BB Bar		님		Luxembourg (3)	
I <u>—</u>	lgaria ⁽³⁾	X		Madagascar	
4	azil ⁽³⁾	H		Mongolia (3)	
	LI Switzerland and Liechtenstein			Netherlands	
	echoslovakia	X		Norway	
	ermany ⁽³⁾	X		Poland ⁽³⁾	
l 🖳	Denmark X RO Romania				
1 🚟	ain ⁽³⁾	肖	SD	Sudan	
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l 🛁	nited Kingdom			United States of America (3)	
MU Hu	• •	[X]	US	continuation-in-part	
! ≝ .	pan ⁽³⁾			994.0	
KP Democratic People's Republic of Korea (3)					
X RU Russian Federation					
X RU Russian Federation Space reserved for designating States (for the purposes of a national patent) which have become party to the PCT after the issuance of					
this sheet:					
		·			
-1	 The applicant's choice of the order of designations may be indicated by marking the check-boxes with sequential arabic numerals (see also the "Notes to Box No. V"). The selection of particular States for a European patent can be made upon entering the national (regional) phase before the European 				

Patent Office (see also the "Notes to Box No. V").

(3) If another kind of protection or a title of addition or, in the United States of America, treatment as a continuation or a continuation-in-part is desired, specify according to the instructions given in the "Notes to Box No. V."

Supplemental Box. Use this box in the following cases:

- (i) if more than three persons are molved as applicants and/or inventors; in such case, write ontinuation of Box No. III" and indicate for each additional person the same type of information as required in Box No. III;
- (ii) if, in Box No. II or any of the sub-boxes of Box No. III, the indication "the States indicated in the 'Supplemental Box," is checked; in such case, write "Continuation of Box No. II" or "Continuation of Box No. III" (as the case may be), indicate the name of the applicant(s) involved and, next to (each) such name, the State or States (or EP or OA, if applicable) for the purposes of which he/she/it is applicant;
- (iii) if, in Box No. II or any of the sub-boxes of Box No. III, a person indicated as "applicant and inventor" or "inventor only" is not inventor for the purposes of all designated States or for the purposes of the United States of America; in such case, write "Continuation of Box No. II" or "Continuation of Box No. III" or "Continuation of Boxes No. II and No. III" (as the case may be), indicate the name of the inventor and, next to such name, the State or States (or EP or OA, if applicable) for the purposes of which the named person is inventor;
- (iv) if there is more than one agent and their addresses are not the same; in such case, write "Continuation of Box No. IV" and indicate for each additional agent the same type of information as required in Box No. IV;
- (v) if, in Box No. V, the name of any country (or OAPI) is accompanied by the indication "patent of addition," or if in Box No. V, the name of the United States of America is accompanied by an indication "Continuation" or "Continuation-in-pan"; in such case, write "Continuation of Box No. V" and the name of each State involved (or OAPI), and after the name of each such State (or OAPI), the number of the parent title or parent application and the date of grant of parent title or filing of parent application;
- (vi) if there are more than three earlier applications whose priority is claimed; in such case, indicate "Continuation of Box No. VI" and indicate for each additional earlier application the same type of information as required in Box No. VI;
- (vii) if, in any of the Boxes, the space is insufficient to furnish all the information; in such case, write "Continuation of Box No..." [indicate the number of the Box] and furnish the information in the same manner as required according to the captions of the Box in which the space was insufficient;
- (viii) if the applicant intends to claim, in respect of any designated Office, the benefit of provisions of the national law concerning non-prejudicial disclosures or exceptions to lack of novelty; in such case, write "Statement Concerning Non-prejudicial Disclosures or Exceptions to Lack of Novelty" and furnish that statement below.

Continuation of Box V

United States of America, Continuation-in-part of
Application No. 853,631
filed 18 March, 1992 (18.03.92)

which in turn is a continuation of Application No. 714,486 filed 13 June 1991 (13.06.91)

If this Supplemental Box is not used, this sheet need not be included in the Request.

Box No. VI PRIORITY CL	AIM ANY). The priority of the	e following earlier application is	hereby claimed:		
Country (country in which it was filed if national application; one of the countries for which it was filed if regional or international application)	Filing Date (day, month, year)	Application No.	Office of filing (fill in only if the earlier application is an international application or a regional application)		
(I) U.S.	13 June 1991 (13.06.91)	714,486			
(2) U.S.	18 March 1992 (18.03.92)	853,631			
(3)	3)				
(Letter codes may be used to inc	dicate country and/or Office of filin	ng)			
Office, the applicant may, again the receiving Office is her	is filed with the Office which, for the st payment of the required fee, ask the teby requested to prepare and trans of the earlier applications identified	he following: mit to the International Rureau a c	pertified conv of the above-men-		
Searching Authority has already to the extent possible, on the res	ARCH (IF ANY). Fill in where a set been requested (or completed) and ults of the said earlier search. Idention by reference to the search reques	the said Authority is now requested ify such search or request either by i	to base the international search.		
International application number number and country (or regiona Office) of other application:	er or I	International/regional/national filing date:			
Date of request for search:		Number (if available) given to search request:			
•		•			
For Janssen Pharma Dirk WANTE, Proxy H If the present Request form is s	older	Frans E. JANSSENS Joseph E. LEENAER y an agent, a separate power of att	orney appointing the agent and		
Office), a copy thereof must be	ed. If in such case it is desired to mattached to this form. (To be filled in by the Applicant)	•	as filed is accompanied by the		
	on contains the following number	items marked below: 1. separate signed power of			
1. request	5 sheets	2. Copy of general power of	· ·		
2. description					
3. claims	14 sheets	3. X priority document(s) (se	, and the second		
4. abstract	1 sheets	4. receipt of the fees paid	or revenue stamps		
5. drawings	sheets	5. cheque for the payment	of fees		
	Total 91 sheets	6. Trequest to charge depos	it account		
Figure number					
- 	(The following is to be filled	in by the receiving Office)			
I. Date of actual receipt of the	purported international application		2 (0 9. 06. 92)		
	eipt due to later but timely received purported international application				
3. Date of timely receipt of the	e required corrections under Article	11 of the PCT:			
4. Drawings Received	X No Drawings				
Date of receipt of the record co	(The following is to be filled in py: 29 JUNE 1992)		(29.06.92)		